

2

3 **Temporal evolution of a post-caldera, mildly peralkaline**  
4 **magmatic system: Furnas volcano, São Miguel, Azores**

5

6 A. J. Jeffery<sup>a\*</sup>, R. Gertisser<sup>a</sup>, B. O'Driscoll<sup>b</sup>, J. M. Pacheco<sup>c</sup>, S. Whitley<sup>a</sup>, A. Pimentel<sup>d</sup>,  
7 <sup>c</sup>, S. Self<sup>e</sup>

8

9 **a)** *School of Physical and Geographical Sciences, Keele University, Keele,*  
10 *Staffordshire, ST5 5BG, United Kingdom*

11 **b)** *School of Earth, Atmospheric, and Environmental Sciences, The University of*  
12 *Manchester, Manchester, M13 9PL, United Kingdom*

13 **c)** *Centro de Vulcanologia e Avaliação de Riscos Geológicos, University of the*  
14 *Azores, 9501-801 Ponta Delgada, Azores, Portugal*

15 **d)** *Centro de Informação e Vigilância Sismovulcânica dos Açores, 9501-801*  
16 *Ponta Delgada, Azores, Portugal*

17 **e)** *Department of Earth and Planetary Science, University of California,*  
18 *Berkeley, CA 94720, U.S.A.*

19

20

21

22 \* Corresponding author:

23 *Tel.* +44 (0) 1782 733620

24 *E-mail* a.j.jeffery@keele.ac.uk

25

26

## 27 **Abstract**

28 Furnas is one of three active central volcanoes on São Miguel island, Azores, and is  
29 considered to be one of the most hazardous in the archipelago. In this study, the pre-  
30 eruptive magma plumbing system of the 10 young (< 5 ka), intra-caldera, sub-Plinian,  
31 trachytic eruptions of the Upper Furnas Group (UFG) is investigated via whole rock  
32 major and trace element geochemistry, mineral chemistry, thermobarometry, and  
33 petrogenetic modelling. The main aim of this work is to elucidate the petrogenesis of  
34 the Furnas trachytes, constrain the P-T-fO<sub>2</sub> conditions under which they evolve, and  
35 investigate the temporal evolution of the magma plumbing system.

36 Results indicate that the trachytes are derived predominantly from extended  
37 fractional crystallisation of alkali basalt parental magmas, at depths between ~ 3 and  
38 4 km. This is considered to take place in a density stratified reservoir, with alkali  
39 basalt magmas at the base and hydrous trachytes forming an upper cap or cupola.  
40 The presence of this reservoir at shallow crustal depths beneath the caldera likely  
41 inhibits the ascent and subsequent eruption of mafic magmas, generating a  
42 compositional Daly Gap. Rare syenitic ejecta represent in-situ crystallisation of  
43 trachytic magmas in the thermal boundary zone at the top of the reservoir. Trachytic  
44 enclaves within these syenites, in addition to banded pumices and ubiquitous  
45 clinopyroxene antecrysts in the UFG pumice falls, provide evidence for  
46 mingling/mixing processes within the magmatic system.

47 Despite relatively uniform major element compositions, systematic trace  
48 element variations within individual eruptions highlight the importance of fractional  
49 crystallisation during late stage evolution of the trachytes. This is facilitated by the  
50 accumulation of water and the development of mild peralkalinity, which contribute to  
51 low pre-eruptive melt viscosities and efficient crystal settling. Higher-order  
52 compositional zoning patterns between individual eruptions cannot be accounted for  
53 by periodic tapping of a single magma batch undergoing fractional crystallisation.

54 Instead, up to four individual cycles are recognised, in which a zoned cap of eruptible  
55 trachytic magma, formed at the top of the reservoir, was erupted in one or more  
56 eruptions, and was re-established via intermittent replenishment and subsequent  
57 fractional crystallisation.

58

59 **Keywords:** Furnas volcano, Peralkaline trachyte, Fractional crystallisation, Zoned  
60 magma reservoir, Post caldera volcanism

61

62

## 63 **Introduction**

64 Whilst peralkaline volcanic rocks are volumetrically less significant than their  
65 subalkaline counterparts, the petrogenetic processes that lead to their formation may  
66 be more complex (e.g. Mahood 1984; Mungall and Martin 1995; Scaillet and  
67 Macdonald 2001; White et al. 2009; Markl et al. 2010). Recent studies of peralkaline  
68 silicic magmatic systems have led to the general consensus that mafic magmas act  
69 either as a parental magma, linked to the peralkaline silicic magmas primarily by  
70 fractional crystallisation (e.g. Barberi et al. 1975; Self and Gunn 1976; Civetta et al.  
71 1998; Peccerillo et al. 2007; Macdonald 2012), or alternatively act as a heat source  
72 generating crustal-derived partial melts which may then differentiate via fractional  
73 crystallisation (e.g. Bohrson and Reid 1997; Trua et al. 1999; Avanzinelli et al. 2004).  
74 Both of these models can be affected by varying degrees of crustal assimilation and  
75 magma mixing (Macdonald et al. 2015).

76 Here, we investigate the petrogenetic processes that operate in the basalt-  
77 trachyte suite of Furnas volcano, São Miguel, Azores (Fig. 1). Furnas volcano is an  
78 example of an oceanic island volcanic centre that exhibits many of the petrogenetic  
79 problems associated with peralkaline magmatic systems, including a volumetric  
80 abundance of felsic products with a peralkaline affinity, an apparent Daly Gap, and a

81 spatial concentration of felsic products within the volcano's caldera complex, with  
82 mafic products on the flanks (Booth et al. 1978; Moore 1991). Guest et al. (1999)  
83 suggested that, similar to neighbouring volcanoes Sete Cidades and Fogo (Agua de  
84 Pau), the felsic magmas of Furnas were largely generated via fractional  
85 crystallisation (cf. Storey et al. 1989; Moore 1991; Widom et al. 1992), and that the  
86 observed geographical distribution of mafic and felsic eruption products could result  
87 from the presence of established trachytic magma reservoirs that intercept ascending  
88 mafic magmas (cf. Wolff 1987). However, the evolution of peralkaline trachyte has  
89 not been quantitatively demonstrated at Furnas, and constraints upon the plumbing  
90 system are currently limited to geophysical studies (e.g. Machado 1972; Camacho et  
91 al. 1997; Montesinos et al. 1999), which broadly point towards a magma reservoir at  
92 shallow depths in the crust (~ 5 km).

93 In this contribution, detailed petrography, mineral chemistry,  
94 thermobarometry, whole rock major and trace element geochemistry, as well as  
95 petrogenetic modelling, is applied to 9 of the 10 recent (< 5 ka) post-caldera  
96 eruptions that together constitute the Upper Furnas Group (UFG). With this dataset,  
97 this study aims to: (1) investigate the petrogenesis and differentiation of mildly  
98 peralkaline silicic magmas at Furnas volcano and the generation of a Daly Gap, (2)  
99 place constraints upon the P-T-fO<sub>2</sub> conditions of the magma plumbing system  
100 beneath Furnas, and (3) evaluate the compositional evolution of a periodically-  
101 tapped, metaluminous to mildly peralkaline magma plumbing system over a period of  
102 ~ 5,000 years (Guest et al. 1999). A greater understanding of the pre-eruptive  
103 plumbing system of Furnas has implications for the processes that generate mildly  
104 peralkaline silicic magmas in such settings (e.g. Macdonald 2012), and the re-  
105 establishment and temporal evolution of post-caldera magmatic systems (e.g.  
106 Bachmann et al. 2012), as well as for the ongoing risk management of this highly  
107 active and hazardous volcano (e.g. Jones et al. 1999; Chester et al. 2002).

108

## 109 **Geological background**

110 The Azores archipelago is located in the central North Atlantic Ocean (Lat. 37°N to  
111 40°N, Long. 25°W to 32°W), 1,300 km west of the Portuguese mainland. The islands  
112 are situated in a unique geodynamic setting as a result of the triple junction where  
113 the North-American, Eurasian and Nubian plates meet (Fig. 1a). The nine islands  
114 themselves are the emergent portions of large volcanic centres that rise from the  
115 Azores Plateau, a bathymetric and gravitational anomaly associated with a  
116 morphologically complex area ( $\sim 5.8 \times 10^6 \text{ km}^2$ ) of thickened oceanic crust that  
117 formed between 20 and 7 Ma (Kaula 1970; Ridley et al. 1974; Gente et al. 2003).  
118 The islands are divided into three groups based upon their geographical position.  
119 The central (Pico, Faial, São Jorge, Graciosa, Terceira) and eastern (Santa Maria  
120 and São Miguel) groups lie within a broadly triangular portion of the Azores plateau,  
121 defined by three major tectonic features; the Mid-Atlantic Ridge in the west, the East  
122 Azores Fracture Zone (EAFZ) to the south and the Terceira Rift to the north-east  
123 (Krause and Watkins 1970; Ridley et al. 1974; Vogt & Jung 2004; Luis and Miranda  
124 2008). In contrast, the western group (Flores and Corvo) lie outside of this feature, on  
125 the western side of the Mid-Atlantic Ridge.

126         The island of São Miguel belongs to the eastern group and is the largest of  
127 the Azorean islands, comprising three active Quaternary trachytic central volcanoes  
128 (from west to east: Sete Cidades, Fogo (or Agua de Pau) and Furnas) (Fig. 1b). Sete  
129 Cidades and Fogo are separated by the Picos basaltic fissure system (the Waist  
130 zone of Booth et al. 1978), whilst Furnas is separated from Fogo by the Congro  
131 basaltic fissure system (the Achada das Furnas complex; Booth et al. 1978). The  
132 easternmost part of the island is composed by the Pliocene-Pleistocene  
133 Povoação/Nordeste volcanic system, which is considered extinct (Fig. 1b). Unlike  
134 Sete Cidades and Fogo, Furnas lacks a distinctive constructive morphology, and is  
135 instead made up of an approximately 7 x 6 km nested caldera complex (Fig. 1c) that

136 is considered to have been formed by at least two caldera-forming eruptions dated at  
137 ~30,000 <sup>14</sup>C y B.P. and ~12,000 <sup>14</sup>C y B.P. (Duncan et al. 1999; Guest et al. 1999; cf.  
138 Sibrant et al. 2015).

139

## 140 **Furnas eruptive history**

141 Furnas is considered to be one of the most active and hazardous volcanoes in the  
142 entire Azores archipelago, having erupted at least 10 times in the last 5,000 years.  
143 During its subaerial existence, Furnas has exhibited effusive (dome-forming) to highly  
144 explosive (caldera-forming) eruptions of felsic magmas (Cole et al. 1999; Guest et al.  
145 1999, 2015). Approximately 1,500 people inhabit Furnas caldera, with several  
146 hundreds more in the immediate vicinity. A stratigraphical study by Guest et al.  
147 (1999) revealed that more than 90 % of eruption products relating to Furnas are of  
148 pyroclastic nature, suggesting a volcanic history in which explosive activity is  
149 prevalent, certainly more so than the neighbouring Povoação/Nordeste volcanic  
150 system. Volcanism within the caldera complex has been exclusively trachytic in  
151 composition, with mafic products being limited to vents on the volcano's flanks.  
152 Although Guest et al. (1999) recognised a compositionally continuous sequence from  
153 basalt to trachyte, intermediate compositions were found to be subordinate in  
154 volume, indicating the presence of a compositional Daly Gap. The same study  
155 divided the stratigraphy of Furnas into three groups: the Upper (UFG), Middle (MFG)  
156 and Lower (LFG) Furnas Groups. The LFG and MFG comprise numerous trachytic  
157 pyroclastic units, with minor trachytic domes and basaltic cones (Moore 1990). The  
158 boundary between the two groups is marked by the Povoação Ignimbrite Formation  
159 (~ 30,000 <sup>14</sup>C y B.P.), a voluminous, partially welded, metaluminous to mildly  
160 peralkaline ignimbrite that fills much of the neighbouring Povoação caldera and is  
161 believed to represent the first caldera-forming eruption of Furnas volcano (Schmincke  
162 and Weibel 1972; Booth et al. 1978; Duncan et al. 1999).

163           The UFG is made up of at least 10 intra-caldera, sub-Plinian eruptions of  
164 trachytic pumice, named Furnas A to J (the latter is also known as Furnas 1630; Cole  
165 et al. 1995; Guest et al. 2015) in order of decreasing age (Booth et al. 1978) (Online  
166 Resource 1). Of these eruptions, the latter two are known to have been historic,  
167 occurring after settlement on the island. Three of these eruptions are known to have  
168 also produced lava domes in their final stages (Furnas E, I and J) (Cole et al.  
169 1999). The dominant lithologies of the UFG are inter-bedded pumice lapilli and ash  
170 beds that are inferred to be the result of complex transitions between magmatic and  
171 phreatomagmatic activity during eruption, though Furnas H appears to have  
172 comprised exclusively magmatic activity (see stratigraphic logs in Online Resource 2)  
173 (Cole et al. 1995; Guest et al. 1999). The base of the UFG is marked by a deposit  
174 from nearby Fogo volcano, named Fogo A (Walker and Croasdale 1971).  
175 Radiocarbon ages for Fogo A range from 4480 to 5380 <sup>14</sup>C y B.P. (e.g. Moore and  
176 Rubin 1991; Wallenstein 1999; Pensa et al. 2015), but its age is generally reported  
177 as ~ 5 ka. Although the deposit is not genetically related to Furnas, it is well-known  
178 and easily identified, and, in the absence of well-exposed deposits relating to the  
179 second caldera-forming event at Furnas (~ 12,000 <sup>14</sup>C y B.P.), was chosen as the  
180 reference stratigraphic marker of the island that delimits the base of the UFG (Guest  
181 et al. 1999). Other deposits from Fogo are also commonly included within the  
182 stratigraphical sequence of the UFG to act as chronological reference points (e.g.  
183 Cole et al. 1999).

184

## 185 **Petrography**

### 186 ***Pumice falls***

187 The fall deposits of the UFG are dominated by highly vesicular (~ 85 %), aphyric to  
188 crystal poor (< 5 % phenocryst content) pumice clasts. The dominant mineralogy  
189 shows little variation throughout the deposits, comprising alkali feldspar (~ 67 vol. %),

190 Fe-Ti oxides (~ 20 vol. %), biotite (~ 10 vol. %), clinopyroxene (~ 3 vol. %), and  
191 apatite (accessory). Rarely, alkali feldspar microlites with high aspect ratios are  
192 observed. Additionally, a small number of pumice clasts from layer Lf of Furnas J  
193 (see Cole et al. 1995) exhibit pronounced banding, notably higher crystal contents,  
194 more abundant groundmass microlites, and rare inclusions of dense, crystalline  
195 material of similar mineralogy to the pumice, which range in size from ~ 0.5 to 6 mm.

196 Alkali feldspar is the most abundant phase, and exists as phenocrysts (up to  
197 ~ 2 mm), microphenocrystic fragments, and, more rarely, groundmass microlites.  
198 Phenocrysts are commonly tabular and euhedral (Fig. 2a), but rare examples are  
199 rounded, with pronounced embayment structures and resorbed cores (Fig. 2b). Small  
200 crystal fragments with irregular, fractured forms are also present, which may be  
201 heavily resorbed. Such examples are particularly common in the banded pumices of  
202 Furnas J, described above. Where groundmass microlites are present, they are  
203 almost exclusively acicular, and do not exceed 100  $\mu\text{m}$  in length.

204 Clinopyroxene exists as relatively well formed, green microphenocrysts,  
205 typically less than 0.5 mm in length (Fig. 2c). Inclusions of acicular apatite are  
206 frequently observed, and Fe-Ti oxide inclusions are ubiquitous. More rarely, small (<  
207 50  $\mu\text{m}$ ) clinopyroxene crystals are seen with irregular crystal forms, either heavily  
208 fractured, or rounded and resorbed. Such examples are more abundant in the  
209 banded pumices of Furnas J.

210 Biotite occurs as small (< 0.5 mm) microphenocrysts (Fig. 2d). Crystal forms  
211 are dominantly euhedral to subhedral and may show evidence for resorption. Fe-Ti  
212 oxides are generally very small (< 0.5 mm), subhedral, and frequently associated  
213 with clinopyroxenes. More rarely, they may be found as inclusions within feldspars.  
214 They are ubiquitously associated with small, acicular apatite crystals, which may be  
215 partially or fully included within the Fe-Ti oxide host.

216

217 ***Lava domes***



218 Samples collected from the two lava domes associated with Furnas I and J reveal  
219 mineral assemblages that are broadly similar to those of the UFG pumice falls,  
220 including phenocrysts and microphenocrysts of alkali feldspar, clinopyroxene, Fe-Ti  
221 oxides, biotite, and apatite, in approximate order of abundance. However, due to  
222 several key mineralogical and textural differences, the two domes are treated  
223 separately.

224 The lava dome associated with Furnas J is dominated by alkali feldspar,  
225 which can be found as phenocrysts up to ~ 5 mm in size, and also as relatively  
226 coarse, acicular microlites in the groundmass (up to ~ 0.4 mm). Phenocrystic  
227 feldspars range from euhedral to subhedral, with rounded edges and uncommon  
228 embayments providing evidence for partial resorption. Feldspar microlites are  
229 frequently aligned, giving the groundmass a well defined trachytic texture, which is  
230 also taken up by proximal phenocrysts. Rarely, feldspars are observed as  
231 glomerocrysts that reach sizes of up to ~ 2 mm, and comprise a number of partially  
232 resorbed feldspar phenocrysts that meet at irregular and often lobate boundaries  
233 (Fig. 2e).

234 Clinopyroxene is found throughout the dome as phenocrysts (up to ~ 1 mm)  
235 and groundmass microlites (up to ~ 0.1 mm). As observed in the UFG pumice falls, it  
236 is characteristically green and ubiquitously associated with smaller Fe-Ti oxide  
237 crystals which are partially or totally included within the larger clinopyroxene crystal.  
238 Phenocrystic and microphenocrystic clinopyroxenes display brown alteration rims  
239 that also extend along any fractures that cut through the crystal (Fig. 2f). In addition  
240 to Fe-Ti- oxides, small apatite needles are also frequently observed as inclusions.

241 Fe-Ti oxides are found as uncommon microphenocrysts (generally between  
242 0.25 and 0.5 mm in size) and abundant groundmass crystals (< 50  $\mu\text{m}$ ).  
243 Microphenocrysts frequently exhibit exsolution lamellae of ilmenite. Such features are  
244 rare in the groundmass oxides. Like the UFG pumice falls, inclusions and partial  
245 inclusions of apatite are abundant.

246 Biotite is uncommon and is generally found as microphenocrysts, less than  
247 0.2 mm in length. Examples are frequently well formed and often aligned with the  
248 trachytic texture in the groundmass. Rare examples of larger, phenocrystic biotite, up  
249 to 0.75 mm in length, are present in some samples. Unlike microphenocrystic biotite,  
250 these examples are ragged and exhibit prominent reaction rims (Fig. 2g), which  
251 contain abundant small Fe-Ti oxides, and also run along some fractures and  
252 cleavage planes.

253 In contrast to the UFG pumice falls, the Furnas J dome also contains sodalite,  
254 which can be found as well-formed, equant microphenocrysts that are typically 0.2  
255 mm in size. Inclusions of feldspar and clinopyroxene are extremely common, and  
256 feldspar microlites frequently impinge upon the sodalite crystal faces. In some cases,  
257 sodalite appears rounded and partially embayed, indicating resorption.

258 The Furnas I lava dome rock is similar to that from the Furnas J dome,  
259 containing alkali feldspar phenocrysts surrounded by a coarse, alkali feldspar-based  
260 groundmass, as well as uncommon alkali feldspar glomerocrysts that show irregular  
261 and lobate interior grain boundaries. Green clinopyroxene is found as anhedral  
262 phenocrysts (up to ~ 1 mm) in length, and as small microlites in the groundmass. Fe-  
263 Ti oxides are present as frequently exsolved microphenocrysts that are associated  
264 with the clinopyroxene phenocrysts, and as groundmass crystals. Biotite was  
265 observed in the Furnas I dome, as small subhedral microphenocrysts (< 0.2 mm), or  
266 rare, ragged phenocrysts with an Fe-Ti oxide-rich reaction rim. Sodalite  
267 microphenocrysts are generally small (< 0.2 mm) and euhedral, with abundant  
268 inclusions of feldspar and clinopyroxene (Fig. 2h). The Furnas I dome also contains  
269 green amphibole, found only as anhedral, interstitial crystals in the groundmass (Fig.  
270 2h).

271

272 ***Syenite nodules***

273 A small number of coarse to medium grained syenitic clasts ejected as cognate  
274 xenoliths were sampled from the pyroclastic surge units associated with the Furnas J  
275 eruption. Individual specimens rarely exceed ~ 10 cm in size, and, in some instances,  
276 contain enclaves of darker, finer-grained rock. Boundaries between enclave material  
277 and host syenite range from sharp and well defined to irregular and diffuse, and, in  
278 some samples, both boundary types can be observed. Syenite mineral assemblages  
279 are dominated by cumulus alkali feldspar, which reaches sizes of up to 5 mm, and  
280 frequently displays perthitic textures. Intercumulus phases include green  
281 clinopyroxene, amphibole, Ti-magnetite, ilmenite, biotite, pyrrhotite, and rare zircon  
282 (Online Resource 3). Mirolitic cavities are common in both the syenites and  
283 enclaves. Unlike reported syenitic ejecta from Terceira (Jeffery et al. in press),  
284 Ascension (e.g. Harris 1983), and nearby Fogo volcano (Ridolfi et al. 2006), the  
285 Furnas syenites do not contain any of the Na-K-Ca-Ti-Zr silicate minerals that  
286 characterise agpaitic phase assemblages (e.g. aenigmatite, astrophyllite, eudialyte,  
287 cf. Marks et al. 2011). Instead, Ti-magnetite, ilmenite, and trace amounts of zircon  
288 indicate that the syenites are miaskitic.

## 289 **Methods**

### 290 ***Analytical techniques***

#### 291 **Whole rock analyses**

292 Whole rock major and trace element analyses for the UFG were undertaken at The  
293 Open University, U.K., using an Applied Research Laboratories 8420+ dual  
294 goniometer wavelength dispersive X-ray fluorescence (XRF) spectrometer. Additional  
295 analyses were produced using a Bruker AXS S4 Pioneer XRF spectrometer at the  
296 University of East Anglia, U. K. The trace element concentrations of a small subset of  
297 samples were determined via inductively coupled plasma mass spectrometry (ICP-  
298 MS), using an Agilent 7500s quadrupole mass spectrometer at The Open University,  
299 U.K. Additional major and trace element analyses were undertaken at Bureau Veritas

300 Mineral Laboratories, Canada, using inductively coupled plasma atomic emission  
301 spectroscopy (ICP-AES) for major element analysis and ICP-MS for trace element  
302 analysis.

303 The whole rock dataset was supplemented by 117 quantitative analyses from  
304 various stratigraphic levels within the UFG, focussing on selected trace elements (Zr,  
305 Nb, Rb, Sr) due to their sensitivity to fractionation of olivine, clinopyroxene, and  
306 feldspar. This was achieved by X-ray fluorescence (XRF) at Keele University, U.K.,  
307 using a Niton XI3t energy dispersive XRF spectrometer. Further details of analytical  
308 procedures and data quality are provided in Online Resources 4 and 5.

309

### 310 **Mineral and glass analyses**

311 Major element compositions of mineral phases and glass (both melt inclusions and  
312 groundmass glass) were analysed using a CAMECA SX 100 electron microprobe at  
313 The Open University, U.K., and a JEOL JXA 8900 RL electron microprobe at the  
314 University of Göttingen, Germany. Full details of the analytical procedure and data  
315 quality are given in Online Resources 4 and 5, respectively.

316

### 317 ***Thermobarometry***

318 In the presence of both Ti-magnetite and ilmenite, magmatic temperatures and  
319 oxygen fugacity ( $fO_2$ ) estimates were calculated using the ILMAT program of Lepage  
320 (2003), applying the model of Andersen et al. (1993) with the calculation scheme of  
321 Stormer (1983) to determine values for  $X_{hem}$ ,  $X_{ilm}$ ,  $X_{mag}$ ,  $X_{ulv}$ . The application of  
322 alternative calculation schemes (e.g. Carmichael 1967; Anderson 1968; Lindsley and  
323 Spencer 1982) led to variation of no more than 5 % in the calculated results.  
324 Equilibrium between mineral pairs was evaluated using the Mn-Mg partitioning test of  
325 Bacon and Hirschmann (1988). Pre-eruptive magmatic temperatures were also  
326 estimated using the alkali feldspar-melt thermometer of Putirka (2008). In addition to

327 pre-eruptive temperatures, magmatic H<sub>2</sub>O contents were also estimated, using the  
328 hygrometer of Mollo et al. (2015). Due to the absence of notable chemical zonation  
329 within Fe-Ti oxide and alkali feldspar crystals, both core and rim analyses were used  
330 in the calculations, which were based upon individual mineral (and glass) analyses  
331 rather than averages. Full details of the applied procedure are provided in Online  
332 Resource 4.

333

### 334 ***Geochemical modelling***

335 To investigate the petrogenesis of the UFG trachytes, a variety of approaches were  
336 applied, including Rayleigh fractionation and batch melting trace element modelling,  
337 least squares mass balance modelling, and thermodynamic modelling. The full  
338 details of the applied models are given in Online Resource 4.

## 339 **Results**

### 340 ***Major and trace element geochemistry***

341 Whole rock major element data for the UFG units is reported in Online Resource 6.  
342 Notably, analyses for Furnas A and B have lower SiO<sub>2</sub> and Na<sub>2</sub>O values, in addition  
343 to elevated LOI values, which are indicative of post-depositional alteration, and lead  
344 to misleadingly low calculated peralkalinity. For this reason, neither analysis is  
345 considered in the following discussion.

346 The UFG whole rock samples exhibit little major element variation, with SiO<sub>2</sub>  
347 values that cluster around 63 wt. %, low MgO contents (< 0.5 wt. %), and high total  
348 alkali abundances (~ 13 wt. %). Using the Total-Alkali vs. Silica diagram of Le Bas et  
349 al. (1986), the UFG samples plot uniformly within the trachyte field (Fig. 3). The UFG  
350 pumice falls are mildly peralkaline, with peralkalinity indices (P.I.) that range from  
351 1.02 to 1.13. The UFG lava domes range from metaluminous to mildly peralkaline,  
352 with a P.I. range of 0.95 to 1.05. All of the samples are silica undersaturated, with a

353 range in normative nepheline of 0.15 to 6.10 wt. %, correlating positively with  
354 peralkalinity index. Due to the crystal poor nature of the UFG pumice falls, melt  
355 inclusion and groundmass glass analyses are very similar in composition to whole  
356 rock and can be classified as trachyte (Fig. 3).

357 Trace element analyses for the UFG are presented in Online Resource 6.  
358 Chondrite-normalised REE patterns shown in Fig. 4a, and compared with a standard  
359 OIB, reveal a relatively uniform enrichment in LREEs relative to HREEs, with  $La_N/Yb_N$   
360 ratios between 19 and 31. They also display a pronounced, negative Eu anomaly,  
361 with  $Eu/Eu^*$  values between 0.12 and 0.22. This anomaly is somewhat smaller in the  
362 lava domes (0.30 to 0.32), and in the first two eruptions of the UFG (Furnas A and B,  
363 0.37 and 0.45, respectively), though the effects of post depositional alteration cannot  
364 be ruled out in the latter.

365 Figure 4b presents a multi-element diagram, showing UFG data and a  
366 standard OIB, normalised against the primitive mantle composition of Sun and  
367 McDonough (1989). Notable troughs occur for Ba, Sr, Eu, Ti, and, in some cases,  
368 Pb. In particular, Sr and Ba reach concentrations which are lower than primitive  
369 mantle concentrations.

370

### 371 ***Mineral chemistry***

372 All of the mineral chemical data discussed in the following section is provided in  
373 Online Resource 7.

374

### 375 **Feldspar**

376 The majority of UFG feldspar analyses lie at the anorthoclase-sanidine boundary,  
377 with compositions within  $Or_{23-48}, Ab_{48-69}, An_{0-7}$  (Fig. 5a). Feldspar phenocryst analyses  
378 from the lava domes of Furnas J and Furnas I exhibit a similar range to their  
379 respective pumice falls, though they extend to slightly more sodic compositions.

380 Substituting elements such as Ba, Sr and Fe are frequently below the detection limit.  
381 Feldspars from the Furnas J syenites are similar in composition to those of the other  
382 UFG units, with a range of  $Or_{26-57}$ ,  $Ab_{43-72}$ ,  $An_{0-5}$ , and no variation between syenite  
383 and enclave. Concentrations of Ba, Sr, and Fe are, like the UFG, frequently below  
384 detection limits.

385

### 386 **Clinopyroxene**

387 For classification, the 10 component scheme of Marks et al. (2008) was applied.  
388 When plotted in the quadrilateral, three component (Wo-En-Fs) scheme of Morimoto  
389 et al. (1988) (Fig. 5b), UFG clinopyroxenes lie at the diopside-augite boundary.  
390 Analyses derived from clinopyroxene phenocrysts in the Furnas J and Furnas I lava  
391 domes do not show any significant compositional variation compared of their  
392 corresponding pumice falls. The analyses are best defined using Di, Hd, and Aeg,  
393 which exhibit ranges of  $Di_{45-70}$ ,  $Hd_{12-37}$ , and  $Aeg_{2-18}$  (Fig. 5b). In Di-Hd-Aeg space,  
394 some Furnas samples exhibit a trend of enrichment in Hd, with little to no increase in  
395 the aegirine component. Other units show more homogenous compositions (e.g.  
396 Furnas H), though some comparatively Di-rich outliers are present. The Furnas J  
397 lava dome has a larger compositional range than the Furnas J pumice fall, extending  
398 to more Di-rich compositions. Similarly, the Furnas I lava dome clinopyroxenes show  
399 a somewhat greater range than the Furnas I pumice fall, though this range extends to  
400 more Aeg-rich compositions. Two analysed groundmass crystals are outliers within  
401 this unit, with compositions of  $Di_{33}$ ,  $Hd_{37}$ ,  $Aeg_{26}$ , and  $Di_{12}$ ,  $Hd_7$ ,  $Aeg_{58}$ , and so may  
402 represent a continuation of the described trend, where Aeg enrichment begins after  
403 initial Hd enrichment.

404 The clinopyroxene analyses from the Furnas J syenite ejecta lie in stark  
405 contrast with the rest of the UFG units, with a compositional range from diopside,  
406 through aegirine-augite, to aegirine ( $Qd_{0-93}$ ,  $Jd_{0-8}$ ,  $Aeg_{6-96}$ ). The Ti-aegirine component  
407 is generally present in small concentrations (~ 6 mol. %), but extends to values as

408 high as 34 mol. % in a small number of analyses. In Di-Hed-Aeg space, the data  
409 reveal a poorly defined trend of initial enrichment in hedenbergite, followed by  
410 aegirine enrichment.

411

#### 412 **Fe-Ti oxides**

413 Quantitative analyses of both Ti-magnetite and ilmenite for the UFG are presented in  
414 the  $\text{TiO}_2\text{-FeO-Fe}_2\text{O}_3$  ternary system in Fig. 5c. Analyses were stoichiometrically  
415 corrected for ferric iron. Ti-magnetite shows a compositional range of Magnetite<sub>35-68</sub>,  
416 Ulvöspinel<sub>22-63</sub>, Spinel<sub>1-10</sub>, with an average  $\text{Al}_2\text{O}_3$  content of 0.9 wt. %. MgO and MnO  
417 contents reach 1.3 and 2.7 wt. %, respectively. The available ilmenite analysis from  
418 the Furnas J dome has a relatively MgO and MnO rich composition (Ilmenite<sub>78</sub>,  
419 Pyrophanite<sub>9</sub>, Geikielite<sub>13</sub>). Ti-magnetite from the Furnas J syentite ejecta is similar to  
420 that of other UFG lithologies, with a range of Magnetite<sub>49-67</sub>, Ulvöspinel<sub>32-50</sub>, Spinel<sub>0-1</sub>,  
421 whereas ilmenite from the syenites is notably enriched in MnO and shows essentially  
422 no MgO (Ilmenite<sub>53-75</sub>, Pyrophanite<sub>24-46</sub>, Geikielite<sub>0-1</sub>)

423

#### 424 **Biotite**

425 Biotite analyses from the UFG are Mg-rich, characterised by relatively constant Fe /  
426 (Fe + Mg) ratios (~ 0.41), and variable Si contents (5.44 to 5.71 atoms per formula  
427 unit (apfu)) (Fig. 5d). UFG analyses also exhibit high Ti contents (0.56 to 0.93 apfu).  
428 Occupation of the hydroxyl sites is characterised by variably high F contents (0.32 to  
429 1.63 apfu) and low Cl contents (< 0.02 apfu). Biotite from the Furnas J syenite ejecta  
430 are very similar to those found in the pumice falls and lava domes, with variable Si  
431 contents (5.54 to 5.90 apfu), and high Ti (0.42 to 0.84 apfu). Similarly, the hydroxyl  
432 sites are dominated by F (0.75 to 1.12 apfu), with little Cl (< 0.02 apfu). However, Fe /  
433 (Fe + Mg) ratios extend to larger values (0.44 to 0.72).

434

#### 435 **Amphibole**



436 Following the nomenclature of Leake et al. (1997), all of the Furnas I dome analyses  
437 are Na-Ca amphiboles of magnesio-katophoritic to richteritic composition (Fig. 5e, f).  
438 Chlorine concentrations are generally close to the detection limit (< 0.004 apfu). The  
439 role of F in the hydroxyl site was not quantified. Amphiboles in the Furnas J syenites  
440 exhibit a larger compositional range, which includes Na-Ca amphiboles of  
441 kataphoritic composition, and Na amphiboles of ferric-ferronyböitic and arfvedsonitic  
442 composition. Fluorine concentrations in the hydroxyl site range from 0.45 to 1.21  
443 apfu, and show no significant variation between Na-Ca and Na groups. Chlorine  
444 contents are extremely low (< 0.006 apfu).

445

446

## 447 **Discussion**

448 In this section, the combined dataset is used to provide insights into the pre-eruptive  
449 magmatic system of the young (< 5 ka) volcanic activity of Furnas, exploring first the  
450 pre-eruptive P-T-fO<sub>2</sub> conditions of the UFG trachytes, and the petrogenetic processes  
451 that generate them. Ultimately this allows the construction of a conceptual model for  
452 the magma plumbing system. Second, the compositional zonation patterns observed  
453 throughout the UFG, both within and between individual eruptions, are used to place  
454 constraints upon the temporal evolution of the magma reservoir from which the UFG  
455 eruptions were fed.

456

### 457 ***P-T-fO<sub>2</sub> constraints on the plumbing system***

#### 458 **Thermobarometrical models**

459 Due to the scarcity of ilmenite in the Furnas samples, temperature and fO<sub>2</sub> estimates  
460 for the trachytes, based upon two-oxide equilibria, could only be achieved for the  
461 Furnas J lava dome and the syenite ejecta. Ti-magnetite analyses from the Furnas J

462 pumice fall were also tested for equilibrium with Furnas J dome ilmenite, and a single  
463 additional pairing was found to be suitable for calculation. Estimates for the Furnas J  
464 lava dome indicate a temperature range of ~ 80 °C between 814 and 897 °C and  $fO_2$   
465 values that are slightly above the FMQ buffer (Fig. 6a). As the temperature  
466 decreases, the  $fO_2$  values increase slightly, reaching approximately 0.5 log units  
467 above the FMQ curve at the lowest recorded temperatures. These values are  
468 representative of the final pre-eruptive magmatic temperatures and/or syn-eruptive  
469 conditions along the plumbing system, due to the rapid re-equilibration timescales of  
470 coexisting Fe-Ti oxides (e.g. Gardner et al. 1995; Venezky and Rutherford 1999;  
471 Pimentel et al. 2015). The pre-eruptive redox conditions of the UFG trachytes appear  
472 to be more reducing than those of nearby Sete Cidades, where a total range of ~  
473 FMQ + 1 to ~ MH – 1 was reported by Beier et al. (2006). Interestingly, in the case of  
474 Sete Cidades volcano, both the caldera-forming and the post-caldera deposits (the  
475 latter being comparable to the UFG) record significantly more oxidising conditions  
476 than the UFG. This substantial variation may be linked to degassing of  $H_2$  following  
477 the thermal disassociation of  $H_2O$  (cf. Mungall & Martin 1995), and would indicate a  
478 greater degree of magma degassing in the post-caldera magma system of Sete  
479 Cidades compared to that of Furnas. Equilibrium testing for the syenites indicate only  
480 four equilibrium pairs, yielding temperature between 621 and 780 °C, and  $fO_2$  values  
481 that become more reduced at lower temperatures, ranging from 0.5 to 2 units below  
482 FMQ (Online Resource 8).

483         Pre-eruptive temperature estimates for the trachytes were also calculated via  
484 alkali feldspar-melt thermometry (Putirka 2008), using an initial estimated pressure  
485 input of 100 MPa. The effect of pressure is, in this case, negligible, with a change in  
486 calculated temperature of < 1 °C following a pressure increase of one order of  
487 magnitude. In each case, a range of up to 19 whole rock and groundmass glass  
488 compositions were used as potential equilibrium liquid compositions and the most  
489 suitable was identified using the  $KD_{Or-Ab}$  equilibrium test presented by Mollo et al.

490 (2015). In the majority of cases, all of the alkali feldspar analyses from a single  
491 eruption were found to be well equilibrated with a single groundmass glass or whole  
492 rock analysis from the same eruption. However, in the case of Furnas J, the most  
493 applicable equilibrium liquid for alkali feldspars from both the pumice fall and the lava  
494 dome was found to be a whole rock analysis of the lava dome (SM7-1) (Online  
495 Resource 8).

496         Calculated temperature estimates for the UFG are summarised in Fig. 6b.  
497 The UFG exhibits a temperature range of 855 to 965 °C, with some fluctuation  
498 between individual eruptions. For example, the pumice fall and lava dome of Furnas  
499 J record temperatures in excess of 915 °C, whilst the majority of earlier UFG  
500 eruptions yield somewhat lower temperatures that are generally below 900 °C.  
501 Furnas F represents a notable exception to this, with a narrow range of hotter  
502 temperature estimates between 922 and 935 °C. Estimates derived from syenite  
503 ejecta are similar to those of the UFG falls and domes, but with greater scatter,  
504 ranging from 892 to 969 °C. The limited available estimates for the enclaves yield  
505 temperatures that are not distinguishable from the host syenites. The comparatively  
506 wide range of temperature estimates observed in the syenites may reflect the  
507 temporal variation in feldspar nucleation that inevitably results from in-situ  
508 crystallisation.

509         Estimates of pre-eruptive H<sub>2</sub>O contents (H<sub>2</sub>O<sub>melt</sub>) of the trachytes were  
510 determined for each unit of the UFG via alkali feldspar-melt equilibria (Mollo et al.  
511 2015). In each case, equilibrium liquid compositions were carried over from alkali  
512 feldspar-melt thermometry. Due to the dependency of calculated H<sub>2</sub>O<sub>melt</sub> values upon  
513 the input temperature, the average temperature estimate derived from alkali feldspar-  
514 melt thermometry was used as an input for each of the UFG units. Calculated H<sub>2</sub>O<sub>melt</sub>  
515 estimates for the UFG are summarised in Fig. 6b. Results indicate that the UFG was  
516 hydrous, with average H<sub>2</sub>O<sub>melt</sub> values around 5 wt. %, and a total range from 2.4 to  
517 6.1 wt. %. This range is consistent with the measured water contents of trachytic

518 glass inclusions within alkali feldspars sampled from Furnas J (Jeffery 2016). The  
519 negative correlation between temperature and  $H_2O_{melt}$  indicate that the UFG  
520 trachytes were  $H_2O$  undersaturated prior to eruption. There is some evidence for  
521 variation in  $H_2O_{melt}$  within individual eruptions. In the case of Furnas J and Furnas I,  
522 the late stage lava domes both yield  $H_2O_{melt}$  estimates that extend to somewhat lower  
523 values than the preceding pumice falls. This is most obvious for Furnas J, where the  
524 dome analyses return a range that extends as low as 2.4 wt. %, which is > 1 wt. %  
525 lower than the lowest estimate for the Furnas J pumice fall. Syenite ejecta and their  
526 enclaves yield a range of estimates that are most comparable to those of Furnas J  
527 and Furnas F, but show substantial scatter, ranging from 1.3 to 5.3 wt. %.

528 Equilibrium testing of clinopyroxene indicated that none of the reported UFG  
529 clinopyroxenes were in equilibrium with their host trachytes, precluding quantitative  
530 estimation of pre-eruptive conditions. Beier et al. (2006) noted a similar lack of  
531 crystal-melt equilibrium in trachytes from Sete Cidades. Application of hypothetical  
532 liquid compositions derived from Rhyolite-MELTS modelling indicates that the most  
533 suitable equilibrium liquid compositions are broadly trachyandesitic. Despite the lack  
534 of equilibrium, clinopyroxene chemistry may still allow qualitative insights into the  
535 magma system. Beier et al. (2006) observed two trends in clinopyroxene  $Al_2O_3$  and  
536  $TiO_2$  content at Sete Cidades: 1) high  $Al_2O_3$  and  $TiO_2$  in the mafic rocks, and 2) low  
537  $Al_2O_3$  and  $TiO_2$  in the trachytes. These trends were attributed to high and low  
538 pressure crystallisation conditions, respectively. With the exception of a small  
539 number of clinopyroxenes from a number of the UFG units, the UFG clinopyroxene  
540 population adheres to the low  $Al_2O_3$  and  $TiO_2$  trend (Fig. 7). On this basis, the bulk of  
541 clinopyroxene crystallisation may be constrained to shallow crystal conditions. The  
542 small number of high pressure clinopyroxenes may reflect an earlier high pressure  
543 fractionation step, similar to that of Sete Cidades, which is poorly represented in the  
544 UFG trachytes.

545

546 **Constraints from H<sub>2</sub>O solubility**

547 Further constraints can be placed upon the minimum depth of the magma reservoir  
548 through the calculation of saturation pressures of the melt. To achieve this, the  
549 solubility models of Di Matteo et al. (2004) and Papale et al. (2006) were applied.  
550 The former applies a polynomial fit of experimental data to parameterize H<sub>2</sub>O  
551 solubility, and yields an estimated minimum pressure of 122 MPa ± 15 MPa,  
552 assuming water saturation at 5.7 wt. %. The second model was applied using a  
553 representative anhydrous MET whole rock composition, and an input temperature of  
554 908 °C (average alkali feldspar-melt equilibria-derived T of UFG), and produces a  
555 minimum pressure of 156 MPa.

556

557 ***Petrogenesis of the UFG trachytes***

558 To investigate the roles of fractional crystallisation and partial melting of crustal rocks  
559 in the generation of the UFG trachytes, we employ Rayleigh fractionation and batch  
560 melting models, using sample 98SM33 of Elliott et al. (2007) as the parental magma,  
561 or a partially melted crustal gabbro, respectively (see Online Resource 4 for full  
562 details). Applied mineral-melt partition coefficients are given in Table 1. Modelling of  
563 compatible trace elements such as Sr, Ba, and Ni indicate that batch melting models  
564 of gabbroic crust with variable mineral assemblages are unable to reproduce the low  
565 concentrations observed in the dataset (Fig. 8a, b, c). Instead, Rayleigh fractionation  
566 curves provide a better fit, and suggest that the compositional variability of the UFG  
567 trachytes (LET to MET) is largely controlled by ~ 75 to 85 % fractionation of an alkali  
568 basaltic parent. Modelled fractionation curves for incompatible elements such as Nb,  
569 and Zr indicate a similar range of fractionation values (~ 70 to 85 %) (Fig. 8d). This is  
570 consistent with nearby Sete Cidades (Beier et al. 2006) and Fogo (e.g. Storey 1982;  
571 Widom et al. 1992), as well as other notable peralkaline centres such as Terceira

572 (e.g. Mungall & Martin 1995) and Pantelleria (e.g. Neave et al. 2012), where  
573 fractional crystallisation plays a dominant role in magma evolution.

574 To further investigate the role of fractional crystallisation, major element mass  
575 balance models were applied using the mineral compositions reported in Table 2.  
576 The results indicate that evolution from parental basalt to the least evolved trachyte  
577 (LET) (step 1) can be adequately explained ( $\sum r^2 = 0.023$ ) by 87.5 % fractional  
578 crystallisation of a phase assemblage including clinopyroxene (45 %), plagioclase  
579 (24 %), Ti-magnetite (12 %), olivine (7 %), biotite (10 %), and apatite (2 %) (Table 3).  
580 The second evolutionary step (LET to the most evolved trachyte (MET)) can be  
581 achieved by a further 38 % fractional crystallisation of an assemblage dominated by  
582 alkali feldspar (94 %), with subsidiary Ti-magnetite (1 %), ilmenite (1 %), biotite (3  
583 %), and apatite (1 %) ( $\sum r^2 = 0.018$ ). Together, steps 1 and 2 indicate that the  
584 differentiation of parental basalt to the most evolved UFG trachyte can be accounted  
585 for by a total of 92 % fractional crystallisation of an assemblage that is dominated by  
586 clinopyroxene and plagioclase during basalt to trachyte evolution, and alkali feldspar  
587 during continued evolution within the trachytes. The predicted late stage prevalence  
588 of alkali feldspar fractionation is in agreement with the mineral assemblages of the  
589 trachytes, where alkali feldspar constitutes the dominant phenocryst phase.  
590 Predicted mafic to intermediate assemblages are broadly similar to available  
591 descriptions of mafic to intermediate rocks from across São Miguel (e.g. Moore 1991;  
592 Beier et al. 2006; Jeffery unpublished data).

593 The results of 150 Rhyolite-MELTS models indicate that a liquid of similar  
594 composition to the UFG trachytes (with respect to MgO, SiO<sub>2</sub>, Al<sub>2</sub>O<sub>3</sub>, total alkalis,  
595 peralkalinity index, and H<sub>2</sub>O<sub>melt</sub>), can be produced via fractional crystallisation of a  
596 hydrous (1.5 wt. %) alkali basalt parent under low pressure (150 MPa), at redox  
597 conditions close to the FMQ buffer reaction curve (Fig. 9). These conditions are  
598 consistent with those predicted via thermobarometrical models. Higher pressure (300  
599 and 500 MPa) models fail to reach the silica and total alkali values exhibited by the

600 UFG, peaking at approximately trachyandesitic compositions. Lower pressure (50  
601 MPa) models underestimate liquid  $\text{Al}_2\text{O}_3$  contents due to the earlier crystallisation of  
602 feldspar, leading to higher peralkalinity indices than observed in the UFG. Polybaric  
603 models also perform well, generating liquid compositions similar to the UFG when  
604 initial water content and redox conditions are set to 1.5 wt. % and the FMQ buffer,  
605 respectively. However, these models yield silica and total alkali contents that are  
606 slightly lower than the UFG. This model predicts olivine as the liquidus phase (1,265  
607 °C), followed by clinopyroxene and Ti-magnetite (1,115 °C), with clinopyroxene  
608 crystallisation ceasing at 905 °C. This is followed by apatite (1,065 °C) plagioclase  
609 feldspar (985 °C), biotite(900 °C), and ilmenite (795 °C). The model reaches  
610 compositions similar to the METs of the UFG at ~ 815 °C, at which point the melt has  
611 undergone ~ 86 % fractionation of olivine (~ 10 %), clinopyroxene (~ 42 %), feldspar  
612 (~ 22 %), Ti-magnetite (~ 10 %), apatite (~ 1 %), and biotite (~ 1 %).

613 In addition to the described petrogenetic models, geochemical characteristics  
614 of the UFG highlight the role of fractional crystallisation. The observed depletion of Ti  
615 and P indicate the fractionation of Fe-Ti oxides and apatite, respectively, whilst the  
616 extreme depletion of Sr and Ba, and the large negative Eu anomaly that is typical of  
617 Furnas trachytes, reflects extensive fractionation of both plagioclase and alkali  
618 feldspars. Based upon the presented evidence, it is suggested that the trachytes of  
619 the UFG are primarily derived from between ~ 70 and ~ 90 % low pressure fractional  
620 crystallisation of an alkali basaltic parental magma under redox conditions close to  
621 the FMQ buffer, although a minor contribution from crustal assimilation or partial  
622 melting of crustal lithologies cannot be ruled out entirely (e.g. Widom and Farquhar  
623 2003; Snyder et al. 2004; Genske et al. 2013, 2014; Larrea et al. 2014). These  
624 elevated fractionation values are compatible with estimates from other peralkaline  
625 systems such as Pantelleria (White et al. 2009; Neave et al. 2012), Sete Cidades,  
626 Azores (Renzulli and Santi 2000; Beier et al. 2006), Fogo, Azores (Widom et al.  
627 1992) and the East African Rift (Scaillet and Macdonald 2003; Rooney et al. 2012).

628 The LETs are likely to have evolved via at least 70 % fractionation from alkali basalt,  
629 through trachybasalt, basaltic trachyandesite, and trachyandesite, fractionating an  
630 assemblage that is initially dominated by clinopyroxene with subsidiary olivine, and  
631 with feldspar becoming more significant at intermediate compositions. The METs are  
632 then produced by further fractional crystallisation of an assemblage dominated by  
633 alkali feldspar, with small amounts of biotite, Ti-magnetite, and apatite. The  
634 crystallisation of ilmenite is not predicted until 795 °C, which may suggest that the  
635 UFG magmas were predominantly stored at temperatures in excess of this value.

636

### 637 ***Origin of syenitic cognate xenoliths***

638 Syenitic ejecta from the Furnas J eruption are considered to represent the final  
639 stages of crystallisation (< 800 °C), characterised by an assemblage that, whilst still  
640 dominated by alkali feldspar, also includes a range of clinopyroxene compositions  
641 (diopside, augite, hedenbergite, aegirine-augite, aegirine), sodic amphiboles, Ti-  
642 magnetite and ilmenite. Amphiboles and clinopyroxenes of broadly similar  
643 composition were observed as interstitial phases in the Furnas I lava dome,  
644 suggesting that, like ilmenite, they only form at temperatures that are lower than  
645 those at which the UFG magmas were stored.

646 Based upon their geochemical and petrographical characteristics, the  
647 syenites are considered to represent in-situ crystallisation at the thermal boundary  
648 zone of a magma reservoir, rather than a cumulate from which significant melt  
649 volumes were extracted (cf. Widom et al. 1993). For example, evolution from LET to  
650 MET has been shown to be controlled by the fractionation of alkali feldspar, biotite,  
651 Ti-magnetite, and apatite. Neither Na-amphibole, nor Na-clinopyroxene are predicted  
652 in thermodynamic models, and both significantly decrease the fit of mass balance  
653 models when included, suggesting that the syenites cannot represent a cumulate  
654 from which the trachytes are derived. Similarly, the syenites and their enclaves show



655 a wide range of trace element compositions that encompass those of the UFG,  
656 rendering many of the UFG eruptions less evolved than the syenites. If the syenite  
657 nodules did represent residual cumulates, they should be uniformly less evolved than  
658 the UFG trachytes. Furthermore, the syenites exhibit Eu anomalies that are  
659 comparable to, or greater than, those of the UFG trachytes. In the UFG rocks, a  
660 negative correlation is observed between  $\text{Eu}/\text{Eu}^*$  and differentiation indices such as  
661 Zr and Nb, suggesting that the accumulation of feldspar could only result in a less  
662 negative, or even positive, Eu anomaly. As such, it is suggested that the syenites  
663 represent near-complete, in-situ crystallisation of unerupted trachytic magmas similar  
664 to those of the UFG. Their trace element variability is ultimately derived from the  
665 varying degrees of fractionation that also yield trace element diversity in the UFG  
666 trachytes, though hydrothermal alteration and the presence of accessory phases  
667 such as zircon also play a role.

668

### 669 ***The role of magma mixing/mingling processes***

670 In addition to FC processes, a number of features of the UFG eruption products  
671 provide evidence for the role of replenishment and magma mingling between two  
672 variably evolved trachytes in the magma reservoir. For example, enclaves hosted in  
673 syenite ejecta display chilled margins, rounded, lobate forms, as well as large alkali  
674 feldspar phenocrysts which exhibit similar compositions to host syenite feldspar and  
675 pronounced disequilibrium textures at their rims, suggesting that they are xenocrysts  
676 derived from the host syenite. These features are all indicative of liquid  
677 emplacement, suggesting that the enclaves represent trachyte-syenite mingling in  
678 the highly crystalline mush zone at the margins of a magma reservoir.

679 Additionally, the presence of banded pumices in the uppermost pumice fall  
680 deposit of Furnas J (Lf) may provide direct evidence for mingling between two  
681 variably evolved trachytes during the later stages of the eruption, as has been

682 described for the C11 explosive eruption (~ 1,000 <sup>14</sup>C y B.P.) of Caldeira volcano,  
683 Faial, Azores (Pimentel et al. 2015). Grey bands within the predominantly white  
684 pumices of Lf exhibit markedly lower vesicularity and thus may have been volatile  
685 poor (Cole et al. 1995). The same study presented evidence for variable viscosity of  
686 the two types based upon deformation characteristics, as well as geochemical  
687 analyses for each type, which, on the basis of Zr content, suggest that the dense  
688 grey pumice may be marginally less evolved. However, it should be noted that both  
689 analyses are ~ 250 to 300 ppm lower in Zr than the Lf analyses of this study, and are  
690 instead comparable to the Furnas J lava dome analyses. This observation implies  
691 that there may be a relatively sharp geochemical transition from high Zr (~ 1200 ppm)  
692 to low Zr (~ 900 ppm) within the Lf unit. Additionally, the analyses of Cole et al.  
693 (1995) exhibit Ba contents an order of magnitude greater than typical UFG  
694 concentrations, which may indicate local resorption of alkali feldspar. A single  
695 analysis within the dataset of this study reveals an anomalous enrichment of Sr (95  
696 compared to < 10 ppm), which may reflect the same process.

697

698

### 699 ***A conceptual model for the UFG magma plumbing system***

700 Based upon the range of pressure estimates derived from H<sub>2</sub>O solubility models and  
701 thermodynamic models (~ 122 to 156 MPa), the magma storage region of the UFG  
702 eruptions can be constrained to shallow crustal depths between ~ 3.3 and ~ 4.1 km  
703 (based upon an average crustal density of 2,800 kg/m<sup>3</sup>). This range compares  
704 favourably with previous estimates of magma storage beneath Furnas volcano  
705 (Machado 1972; Sigmundsson et al. 1995; Blanco et al. 1997; Camacho et al. 1997;  
706 Montesinos et al. 1999), and also the shallow magma reservoir of Sete Cidades  
707 (Beier et al. 2006). The results of thermodynamic modelling, as well as the

708 predominantly low  $\text{Al}_2\text{O}_3$  and  $\text{TiO}_2$  contents of clinopyroxene, suggest that much of  
709 the compositional range from basalt to MET is present at this depth (Fig. 10).

710         Based upon the decreasing melt densities predicted by Rhyolite-MELTS from  
711 basalt to trachyte ( $\sim 2580$  to  $\sim 2230 \text{ kg/m}^3$ ), it is suggested that the reservoir beneath  
712 Furnas is compositionally zoned, with dense mafic magmas at the base, and low  
713 density, hydrous silicic magmas at the roof. Trachytic magmas at the roof may be  
714 further enriched in volatiles via upward migration of a vapour phase, further  
715 stabilising the density stratification. This configuration may account for the apparent  
716 Daly Gap observed at Furnas (Guest et al. 1999), as ascending mafic magmas may  
717 stall in an established shallow crustal magma reservoir and differentiate, ultimately  
718 contributing relatively small volumes of residual trachytic liquid to the silicic cap of the  
719 reservoir rather than erupting themselves (e.g. Mahood 1984; White et al. 2009;  
720 Neave et al. 2012). The likely presence of older, frozen reservoirs at this level is  
721 envisaged to further inhibit rapid magma ascent (e.g. Zanon and Pimentel 2015). In  
722 contrast, basalts that ascend outside the confines of the caldera and avoid the  
723 shallow system, reached the surface and were erupted as the monogenetic cones  
724 and fissure eruptions that are typical of the volcano's flanks and the neighbouring  
725 Congro fissure volcanic system (Zanon 2015).

726         The generation of crystal poor silicic magmas, such as those of the UFG, is  
727 commonly accounted for through the application of models which involve the  
728 extraction of interstitial melt from large crystal mush bodies, citing processes such as  
729 compaction, micro-settling, hindered settling, and gas-driven filter pressing (e.g.  
730 Sisson and Bacon 1999; Bachmann and Bergantz 2004; Hildreth 2004; Pistone et al.  
731 2015). Such models are attractive primarily because of the difficulties associated with  
732 two-phase flow in relatively cool, high viscosity ( $\sim 10^{4.5} \text{ Pa s}$ ) melts (Scaillet et al.  
733 1998). However, their application to peralkaline magmatic systems is complicated by  
734 the apparent absence of erupted crystal-rich melts (equivalent to the 'monotonous  
735 intermediates' associated with non-peralkaline systems, Macdonald 2012), and also

736 by the tendency of peralkaline melts to have lower viscosities than their  
737 metaluminous and peraluminous equivalents (e.g. Giordano et al. 2006; Neave et al.  
738 2012). The efficiency of two phase flow in the UFG trachytes was investigated  
739 through the estimation of crystal free melt viscosities. Application of the model of  
740 Giordano et al. (2008) indicates that the trachytes, at a temperature of 908 °C and  
741  $H_2O_{\text{melt}}$  values from 2.4 to 5.7 wt. %, have viscosities ranging from  $10^{2.7}$  to  $10^{4.1}$  Pa s.  
742 The UFG trachytes therefore had pre-eruptive viscosities that were up to two orders  
743 of magnitude lower than typical silicic melts (Scaillet et al. 1998), which has  
744 implications for the dominant differentiation mechanism.

745 The efficiency of crystal settling was evaluated through the calculation of  
746 Stokes' settling velocities for alkali feldspar crystals (determined to be the most  
747 important crystallising phase during the later stages of magma evolution). This was  
748 supplemented with the hindered settling equation, which allows the determination of  
749 settling rates in polydispersed suspensions (Bachmann and Bergantz 2004):

750

$$U_{hs} = U_{Stokes} \times f(c)$$

751

752 where  $U_{hs}$  is the hindered settling velocity,  $U_{Stokes}$  is Stokes settling velocity, and  $f(c)$   
753 is a correction factor calculated as:

754

$$f(c) = \frac{(1 - c)^2}{(1 + c^{1/3})^{\left[\frac{5c}{3(1-c)}\right]}}$$

755

756 where  $c$  is equal to the crystal fraction. Crystal sizes were set to 2 mm, in accordance  
757 with petrographic observations. Results indicate that even for the highest viscosity  
758 estimate for the UFG, settling rates range from  $0.51 \text{ m/a}^{-1}$  (hindered settling, 50 %  
759 crystallinity) to  $1.76 \text{ m/a}^{-1}$  (unhindered). Settling rates for the lowest viscosity  
760 trachytes reach values as high as  $13.7 \text{ m/a}^{-1}$  (hindered settling, 50 % crystallinity),

761 and  $47 \text{ m/a}^{-1}$  (unhindered). Assuming a volume of melt that is typical for the UFG  
762 eruptions ( $0.1 \text{ km}^3$  DRE; Booth et al. 1978), and hypothetical magma reservoir aspect  
763 ratios of 1:1:1 and 8:1:1, alkali feldspar crystals would need to travel maximum  
764 distances of  $\sim 470$  to  $235 \text{ m}$  to be removed from the volume of erupted melt.  
765 Available radiocarbon ages (and associated errors) for the UFG indicate that repose  
766 times do not exceed 918 years, but are likely to be  $\sim 190$  years (Furnas I to J) or less  
767 (e.g. Furnas H to I, Guest et al. 1999). Over such short timescales (190 years), alkali  
768 feldspar crystals are capable of settling  $\sim 100 \text{ m}$  at the very least (highest viscosity),  
769 but are capable of settling far greater distances (up to  $9600 \text{ m}$  using the lowest  
770 calculated viscosity). It is therefore suggested that, unlike the rhyolitic systems to  
771 which crystal mush models may apply, crystal settling was a viable and efficient  
772 mechanism of differentiation in the UFG trachytes, and may account not only for the  
773 nearly aphyric nature of the erupted trachytes, but also for the rapid rates of  
774 differentiation implied by the wide range in their degree of evolution.

775 Syenitic ejecta are considered to represent a solidification front at the volatile  
776 rich cap of the reservoir, in which trachytic magmas near the margins of the reservoir  
777 solidify via in-situ crystallisation. Phase assemblages including aegirine, Na-  
778 amphibole and ilmenite are consistent with near-complete in-situ crystallisation,  
779 whilst the presence of miarolitic cavities and pyrrhotite indicate the accumulation of  
780 volatile phases. Crystallisation temperatures estimated from Ti-magnetite-ilmenite  
781 pairs are at least  $70 \text{ }^\circ\text{C}$  lower than estimates for the UFG pumice falls. High  
782 temperature estimates derived from alkali feldspar-melt equilibria ( $892$  to  $969 \text{ }^\circ\text{C}$ ) are  
783 similar to those of Furnas J, and likely reflect the relatively early formation of feldspar  
784 compared to later ilmenite, aegirine, and amphibole. Enclaves within the syenites are  
785 considered to represent interaction between the upper thermal boundary layer and  
786 the crystal poor liquids in the underlying crystal poor cupola. Small volumes of crystal  
787 poor trachytes are inferred to have infiltrated the overlying crystal mush and been  
788 quenched. In Zr-Nb space, the majority of syenites and enclaves appear to lie along

789 the same liquid line of descent as the UFG, implying a cogenetic association.  
790 Although Zr-Nb relationships could be uncoupled by the crystallisation of Zr-bearing  
791 phases, the observed rarity of zircon, as well as its restriction to pore spaces, imply  
792 that late-stage zircon was retained rather than fractionated. The substantial range in  
793 the degree of evolution observed in their whole rock chemistry (~ 700 to 1700 ppm  
794 Zr) is likely to be a temporal effect, where the composition of a given syenite clast or  
795 enclave is related to the dominant melt composition in the roof zone of the reservoir,  
796 a feature which is shown to vary considerably. Additionally, it seems probable that  
797 this cupola would be zoned to some extent, exhibiting the same degree of  
798 composition variation as the UFG trachytes themselves. We therefore invoke an  
799 'onion-skin' arrangement, similar to that of Widom et al. (1993), where random  
800 sampling of various layers of the roof zone yields a range of compositions. It should  
801 also be noted that a single syenite analysis appears to deviate from the UFG liquid  
802 line of descent in Zr-Nb space, suggesting that it may represent a fragment of an  
803 alternative liquid line of descent. It thus cannot be ruled out that some syenitic ejecta  
804 may instead represent the 'fossilised' remnants of a pre-UFG magmatic system,  
805 entrained pre- or syn-eruptively within the erupted trachyte.

806         Attempts to employ clinopyroxene thermobarometry revealed that the  
807 clinopyroxenes are not in equilibrium with the host trachytes (see above), instead  
808 being equilibrated with a broadly trachyandesitic composition. This is consistent with  
809 thermodynamic models, which suggest that clinopyroxene becomes saturated in the  
810 melt at 1115 °C and crystallises until 905 °C. Settling rates of clinopyroxene through  
811 the UFG trachytes are approximately three times faster than those of alkali feldspar,  
812 suggesting that the ubiquitous presence within the erupted trachytes requires a  
813 mechanism that is either continuous within the magma plumbing system, or occurs  
814 immediately prior to each eruption. Two possibilities for an origin are considered: (1)  
815 they are xenocrysts derived from the incorporation and assimilation of mafic wall rock  
816 material or (2) they are antecrystic relics derived from liquid-liquid interaction

817 between the erupted trachytes and less evolved magmas. There is no petrographic  
818 evidence, such as crustal or cumulate xenoliths, to support the former hypothesis,  
819 and trace element models do not indicate partial melting as a significant petrogenetic  
820 process in the UFG. As such, the second hypothesis is favoured and, given the small  
821 but consistent presence of clinopyroxene throughout the UFG, implies the episodic  
822 interaction between trachytic and less-evolved, clinopyroxene-bearing magmas, most  
823 likely at an interface between strata with contrasting densities.

824

825

### 826 ***Development of zoned magma bodies***

827 Despite the apparent major element homogeneity and low eruptive volumes  
828 (generally ~ 0.1 km<sup>3</sup> DRE, Booth et al. 1978) of the intra-caldera UFG trachytes,  
829 trace element compositions indicate substantial variation. For example, both the  
830 Furnas J and Furnas I eruptions began with an initial pumice fall, followed by a late  
831 stage lava dome; in both cases, the lava dome is, according to a variety of  
832 differentiation indices (Zr, Nb, REE, Eu/Eu\*), less evolved than its respective pumice  
833 fall. Such phenomena are generally attributed to the development of a zoned magma  
834 reservoir prior to eruption (e.g. Hildreth 1981), and suggest that the uppermost,  
835 eruptible region of the Furnas magma reservoir is able to develop rapidly and  
836 maintain compositional zonation over the timescales of the UFG. A variety of  
837 processes have been suggested to play a role in the development of zonation (e.g.  
838 Hildreth and Wilson 2007); however, due to the described low magma viscosities, it  
839 seems likely that zoning patterns in the UFG are the result of the efficient  
840 fractionation of alkali feldspar crystals, driving the melt towards MET compositions.  
841 The upwards increase in volatile content and associated decrease in viscosity may  
842 lead to a geochemical gradient via variably efficient fractionation (e.g. Furnas J, L1 to  
843 Lf).

844           The preservation of zonation in the UFG magmas until eruption indicates the  
845 absence of efficient convective cells able to rehomogenise the reservoir, which may  
846 be accounted for by a strong density gradient linked to the ~ 4 wt. % range in  
847 estimated melt water contents (cf. Hildreth and Wilson 2007). Alternatively,  
848 compositional zonation might be stabilised through the generation of multiple,  
849 individually convecting layers (e.g. Huppert and Sparks 1984), though the  
850 applicability of such a model in a low volume of magma (< 0.3 km<sup>3</sup>) may be  
851 restricted. If the zoning patterns of the UFG magmas are considered to have been  
852 largely unaltered by conduit processes during ascent, then the potentially sharp  
853 compositional contact between the pumice falls and lava domes of Furnas I and J  
854 may provide evidence for individual strata, reflecting a compositional discontinuity  
855 that was present in the magma reservoir between hydrous MET, and underlying,  
856 volatile-poor LET. This would also indicate that either a deeper part of the reservoir  
857 was tapped than in the majority of previous eruptions, or that the cap had not yet  
858 regenerated the comparatively large volumes of MET associated with older eruptions  
859 (e.g. Furnas C, ~ 0.3 km<sup>3</sup>).

860           The zoning patterns observed in the Furnas C (and possibly F) deposits imply  
861 a more complex regime in which less evolved melts are capable of erupting prior to  
862 more evolved melts. The observed zoning patterns require more complex reservoir  
863 geometries, such as sequential eruption of multiple magma bodies, each acting as an  
864 independent evolving system (e.g. Cooper et al. 2012; Zanon et al. 2013; Pimentel et  
865 al. 2015). Alternatively, this complex zonation might originate from an irregular  
866 chamber geometry which allows lower regions to be tapped first.

867

### 868 ***Temporal evolution of the UFG magma reservoir***

869 In addition to trace element zonation within individual eruptions, the UFG as a whole  
870 exhibits higher order variations which provide insights into the temporal evolution of



871 the previously described eruptible cap of the shallow crustal reservoir. The complex  
872 trace element profiles of the UFG (Fig. 11) are clearly inconsistent with the tapping of  
873 a single large magma batch, emplaced prior to the Furnas A eruption, and allowed to  
874 fractionate for ~ 5,000 years (maximum age of the UFG). Instead, the observed  
875 variability suggests a cyclical regime, in which a zoned, silicic cap forms at the top of  
876 the reservoir, and is partially erupted in one or more individual eruption events,  
877 before being re-established via continued fractionation and potentially mafic  
878 replenishment (Fig. 12). This is consistent with neighbouring Fogo volcano, where a  
879 similar regime of zoned cap formation and tapping over the same time period has  
880 been suggested (Widom et al. 1992). At Furnas, four periods are recognised, during  
881 which a single zoned magma cap was erupted sequentially in up to three individual  
882 eruptive events, reforming after each period. Due to the relatively large errors and  
883 consequent degree of overlap in the radiocarbon ages of the UFG, the repose  
884 periods between individual eruptions cannot be determined accurately. However, it is  
885 suggested that the length of the repose time is likely to be linked to the temporal  
886 model of this study, with the longest repose times being associated with periods  
887 during which the zoned, eruptible cap was being reconstructed.

888         The first zoned magma body of the UFG is the least constrained, but is  
889 inferred to have been tapped twice during the Furnas A and B eruptions, during a  
890 period of up to ~ 3,000 years, followed by a lengthy re-establishment phase, during  
891 which the reservoir cap evolved via 30 to 40 % fractionation prior to the Furnas C  
892 eruption. Available radiocarbon ages suggest that the Furnas B to C repose period  
893 may have been up to 918 years.

894         The Furnas C, E, and F eruptions together indicate a trend of progressively  
895 less evolved trachytes, suggesting that they represent sequential, and probably  
896 relatively rapid, tapping of a single zoned magma body that formed following Furnas  
897 B. This suggests that, despite the relatively substantial volume of the Furnas C  
898 eruption compared to the other UFG eruptions (0.3 km<sup>3</sup> DRE), the system was

899 tapped a further three times prior to re-establishment, which may be linked to the  
900 repose time between eruptions, or the overall rate of trachyte replenishment in the  
901 cap. The repose periods between these eruptions are likely to have been  
902 comparatively short, limiting the effects of continued differentiation between  
903 eruptions. This period was followed by a period of fractionation, which culminated in  
904 the Furnas H eruption, and equated to ~ 10 to 15 % fractionation. During this period,  
905 the reservoir was tapped once by the Furnas G eruption, but this did not inhibit  
906 continued fractionation prior to Furnas H.

907         The third zoned magma body is considered to have formed prior to Furnas H,  
908 and been sequentially tapped by the Furnas H and I eruptions. This is consistent with  
909 the comparatively short repose time between Furnas H and I (~ 20 years, Booth et al.  
910 1978; Guest et al. 1999). Together, these eruptions define a trend of decreasing  
911 evolution that reverses the effect of fractional crystallisation by up to ~ 35 %. This  
912 was followed by a repose period of ~ 190 years, during which the system evolved by  
913 up to 20 % to produce the fourth zoned magma body, which was subsequently  
914 erupted in the Furnas J eruption. Based upon the similarity between the trace  
915 element profiles of the Furnas I and J eruptions, it is possible that the shallow  
916 magmatic system of Furnas has entered a new repose period whilst the eruptible cap  
917 reforms, though it should be noted that the current repose time of 385 years is  
918 approximately twice that of the I to J repose time (~ 190 years).

919

### 920 ***Implications for silicic volcanism on São Miguel***

921 The magma system of Furnas appears to share a number of features with  
922 neighbouring Fogo and Sete Cidades volcanoes. For example, a prominent, sub-  
923 caldera, shallow crustal (~ 3 to 4 km) magma reservoir has been identified at each  
924 volcano, in which mafic to intermediate magmas differentiate primarily via fractional  
925 crystallisation to form a cap of variably evolved trachyte (e.g. Storey 1982; Widom et

926 al. 1992; Renzulli & Santi 2000; Beier et al. 2006). Evidence for magma mingling and  
927 mixing, such as phenocryst textures (Beier et al. 2006), hybrid magmas (Storey et al.  
928 1989), and banded pumices (Widom et al. 1992; this study), is present at all three  
929 centres. However, felsic eruptive products are somewhat less dominant at Sete  
930 Cidades (~75 vol. %; Moore 1991) than at Furnas and Fogo ( $\geq$  90 vol. %; Moore  
931 1991), and the Daly gap observed at both Furnas and Fogo is absent (Beier et al.  
932 2006). One solution is that the upper, trachytic caps of the magma plumbing systems  
933 of both Furnas and Fogo are more developed, possibly due to periods of reduced  
934 basaltic input to the lower part of the systems (cf. Macdonald 2012). They would  
935 therefore form a more significant barrier against the ascent and eruption of basaltic  
936 magmas. In contrast, the magma plumbing system of Sete Cidades may be  
937 somewhat less developed, and so unable to entirely prevent the ascent of basaltic  
938 magmas.

939

940

## 941 **Conclusions**

942 The compositional characteristics of the young (< 5 ka), intra-caldera trachytes of  
943 Furnas volcano can be adequately accounted for by extended fractional  
944 crystallisation from an alkali basalt parental magma at minimum depths of ~ 3 to 4  
945 km. Fractionation is initially dominated by clinopyroxene, which is eventually replaced  
946 by alkali feldspar in the latter stages of differentiation. Despite their major element  
947 homogeneity, trace element compositions highlight the prominent role of continued  
948 fractional crystallisation within the trachytes, revealing up to 50 % fractionation from  
949 the least evolved trachytes (LETs) to the most evolved trachytes (METs). This  
950 process is enhanced by the accumulation of water and the development of  
951 peralkalinity.

952           The magma plumbing system is shown to comprise a single prominent zone,  
953    which may comprise multiple individual magma bodies, in which alkali basalts evolve  
954    through to trachytic residual liquids. Due to density variations, this reservoir is  
955    envisaged to be stratified, with high density mafic magmas at the base, and relatively  
956    cool, hydrous trachytes forming an upper cap. Rare syenitic ejecta are considered to  
957    represent in-situ, near-complete crystallisation of the trachytes in the roof zone of the  
958    reservoir, which is sampled randomly during eruption. Mingling and mixing within the  
959    trachytes is recorded by trachytic enclaves within the syenite ejecta, and by the  
960    presence of banded pumices in Furnas J. Further evidence for mixing can be seen in  
961    the ubiquitous presence of small quantities of clinopyroxene antecrysts within all of  
962    the UFG eruptions, which are likely to be introduced to the trachytes during  
963    interaction with underlying, hotter, intermediate magmas. The presence of such a  
964    shallow reservoir may account for the Daly Gap at Furnas, as ascending mafic  
965    magmas are intercepted before reaching the surface, instead ponding in the lower  
966    reservoir and fractionating towards felsic compositions.

967           Despite the low volumes of the UFG eruptions ( $\leq 0.3 \text{ km}^3$ ), variations in trace  
968    element concentrations within individual eruptions indicate zonation of the upper cap  
969    of the reservoir prior to eruption, with the most evolved melts typically being the first  
970    to erupt. Evidence exists for the presence of both chemical gradients and relatively  
971    sharp compositional stratification, which, in the case of the latter, are considered to  
972    represent sharp compositional discontinuities in the pre-eruptive magma reservoir,  
973    providing direct evidence for the presence of density stratification. Less distinct  
974    gradients likely result from the accumulation of volatiles, as well as the associated  
975    increase in crystal settling rates. Overall, trace element profiles of the UFG are not  
976    compatible with continued fractionation and tapping of a single magma batch, and  
977    are instead indicative of a cyclic regime in which a compositionally stratified trachytic  
978    cap has formed and been sequentially erupted in up to three individual eruptions,  
979    before being re-established. This study suggests that up to there have been up to

980 four such cycles of eruption and re-establishment within the UFG, with the 385 year  
981 repose time since the most recent eruption most likely representing a re-  
982 establishment phase.

983

## 984 **Acknowledgements**

985 We gratefully acknowledge A. Tindle and A. Kronz for analytical support and access  
986 to electron microprobe facilities at the Open University, U.K., and the University of  
987 Göttingen, Germany, respectively. We also acknowledge J. Watson and L. Thomas  
988 for assistance with XRF and ICP-MS analyses at the Open University, U.K. We are  
989 grateful to B. Leze for his support in the production of additional XRF analyses at the  
990 University of East Anglia, U.K. For diligent assistance in the field, K. Ross is also  
991 acknowledged. We are also grateful to P. Greatbatch and D. Wilde for the production  
992 of thin sections and invaluable assistance in the preparation of melt inclusions for  
993 analysis. This manuscript was greatly improved by the insightful and constructive  
994 reviews of A Klügel and C. Beier. We are grateful to J. Hoefs for editorial handling.  
995 This work was supported by the Faculty of Natural Sciences Research Office, Keele  
996 University, who provided use of facilities and financial support. A. Pimentel was  
997 financially supported by CIVISA/CVARG

998

## 999 **References**

1000

1001 Andersen DJ, Lindsley DH, Davidson PM (1993) QUILF: A Pascal program to assess  
1002 equilibria among Fe-Mg-Mn-Ti oxides, pyroxenes, olivine and quartz. *Comput*  
1003 *Geosci*, 19: 1333-1350

1004

1005 Anderson AT (1968) Oxidation of the La Blanche Lake titaniferous magnetite deposit,  
1006 Québec. *J Geol*, 76: 528-547  
1007

1008 Asimow PD, Ghiorso MS (1998) Algorithmic modifications extending MELTS to  
1009 calculate subsolidus phase relations. *Am Mineral*, 83: 1127-1132  
1010

1011 Avanzinelli R, Bindi L, Menchetti S, Conticelli S (2004) Crystallisation and genesis of  
1012 peralkaline magmas from Pantelleria volcano, Italy: an integrated petrological and  
1013 crystal-chemical study. *Lithos*, 73: 41-69  
1014

1015 Bachmann O, Bergantz GW (2004) On the origin of crystal-poor rhyolites: extracted  
1016 from batholithic crystal mushes. *J Petrol*, 45: 1565-1582  
1017

1018 Bachmann O, Deering CD, Ruprecht JS, Huber C, Skopelitis A, Schnyder C (2012)  
1019 Evolution of silicic magmas in the Kos-Nisyros volcanic center, Greece: a petrological  
1020 cycle associated with caldera collapse. *Contrib Mineral Petrol*, 163: 151-166  
1021

1022 Bacon CR, Hirschmann MM (1988) Mg/Mn partitioning as a test for equilibrium  
1023 between coexisting Fe-Ti oxides. *Am Mineral*, 73: 57-61  
1024

1025 Barberi F, Ferrara G, Santacroce R, Treuil M, Varet J (1975) A transitional basalt-  
1026 pantellerite sequence of fractional crystallisation, the Boina Centre (Afar Rift,  
1027 Ethiopia). *J Petrol*, 16:22-56  
1028

1029 Beier C, Haase KM, Hansteen TH (2006) Magma evolution of the Sete Cidades  
1030 volcano, São Miguel, Azores. *J Petrol*, 47: 1375-1411  
1031

1032 Blanco I, García A, Torta JM (1997) Magnetic study of the Furnas caldera (Azores).  
1033 Ann Geofis, 40: 341-359  
1034

1035 Bohrson WA, Reid MR (1997) Genesis of silicic peralkaline volcanic rocks in an  
1036 oceanic island setting by crustal melting and open-system processes: Socorro Island,  
1037 Mexico. J Petrol, 38: 1137-1166  
1038

1039 Booth B, Croasdale R, Walker GPL (1978) A quantitative study of five thousand  
1040 years of volcanism on São Miguel, Azores. Philos T R Soc A, 288: 271-319  
1041

1042 Bryan WB, Finger LW, Chayes F (1969) Estimating proportions in petrographic  
1043 mixing equations by least-squares approximation. Science, 163: 926-927  
1044

1045 Camacho AG, Montesinos FG, Vieira R (1997) A three-dimensional gravity inversion  
1046 applied to São Miguel Island (Azores). J Geophys Res, 102: 7717-7730  
1047

1048 Carmichael ISE (1967) The iron-titanium oxides of silic volcanic rocks and their  
1049 associated ferromagnesian silicates. Contrib Mineral Petrol, 14: 36-64  
1050

1051 Chester DK, Dibben JL, Duncan AM (2002) Volcanic hazard assessment in western  
1052 Europe. J Volcanol Geoth Res, 115: 411-435  
1053

1054 Civetta L, D'antonio M, Orsi G, Tilton GR (1998) The geochemistry of volcanic rocks  
1055 from Pantelleria Island, Sicily Channel: petrogenesis and characteristics of the  
1056 mantle source region. J Petrol, 39: 1453-1491  
1057

1058 Cole PD, Guest JE, Queiroz G, Wallenstein N, Pacheco JM, Gaspar JL, Ferreira T,  
1059 Duncan AM (1999) Styles of volcanism and volcanic hazards on Furnas volcano, São  
1060 Miguel, Azores. *J Volcanol Geoth Res*, 92: 39-53  
1061  
1062 Cole PD, Queiroz G, Wallenstein N, Gaspar JL, Duncan AM, Guest JE (1995) An  
1063 historic subplinian/phreatomagmatic eruption: the 1630 AD eruption of Furnas  
1064 volcano, São Miguel, Azores. *J Volcanol Geoth Res*, 69: 117-135  
1065  
1066 Cooper GF, Wilson CJ, Millet M, Baker JA, Smith EG (2012) Systematic tapping of  
1067 independent magma chambers during the 1 Ma Kidnappers super eruption. *Earth  
1068 Planet Sc Lett*, 313: 23-33  
1069  
1070 Di Matteo V, Carroll MR, Behrens H, Vetere F, Brooker RA (2004) Water solubility in  
1071 trachytic melts. *Chem Geol*, 213: 187-196  
1072  
1073 Duncan AM, Queiroz G, Guest JE, Cole PD, Wallenstein N, Pacheco JM (1999) The  
1074 Povoação Ignimbrite, Furnas volcano, São Miguel, Azores. *J Volcanol Geoth Res*,  
1075 92: 55-65  
1076  
1077 Eggins SM, Woodhead JD, Kinsley LPJ, Mortimer GE, Sylvester P, McCulloch MT,  
1078 Hergt JM, Handler MR (1997) A simple method for the precise determination of  $\geq 40$   
1079 trace elements in geological samples by ICPMS using enriched isotope internal  
1080 standardisation. *Chem Geol*, 134: 311-326  
1081  
1082 Elliott T, Blichert-Toft J, Heumann A, Koetsier G, Forjaz V (2007) The origin of  
1083 enriched mantle beneath São Miguel, Azores. *Geochim Cosmochim Ac*, 71: 219-240  
1084



1085 Freire Luis J, Miranda JM, Galdeano A, Patriat P, Rossignol JC, Mendes Victor LA  
1086 (1994) The Azores triple junction evolution since 10 Ma from an aeromagnetic survey  
1087 of the Mid-Atlantic Ridge. *Earth Planet Sc Lett*, 125: 439-459  
1088

1089 Gardner JE, Rutherford M, Carey S, Sigurdsson H (1995) Experimental constraints  
1090 on pre-eruptive water contents and changing magma storage prior to explosive  
1091 eruptions of Mount St Helens volcano. *Bull Volc*, 57: 1-17  
1092

1093 Genske FS, Beier C, Haase KM, Turner SP, Krumm S, Brandl PA (2013) Oxygen  
1094 isotopes in the Azores islands: crustal assimilation recorded in olivine. *Geology*, 41:  
1095 491-494  
1096

1097 Genske FS, Turner SP, Beier C, Chu MF, Tonarini S, Pearson NJ, Haase KM (2014)  
1098 Lithium and boron isotope systematics in lavas from the Azores islands reveal crustal  
1099 assimilation. *Chem Geol*, 373: 27-36  
1100

1101 Gente P, Dymant J, Maia M, Goslin J (2003) Interaction between the Mid-Atlantic  
1102 Ridge and the Azores hot spot during the last 85 Myr: emplacement and rifting of the  
1103 hotspot derived plateaus. *Geochem Geophys Geosy*, 4: 8514, doi:  
1104 10.1029/2003GC000527  
1105

1106 Ghiorso MS, Sack RO (1995) Chemical mass transfer in magmatic processes IV. A  
1107 revised and internally consistent thermodynamic model for the interpolation and  
1108 extrapolation of liquid-solid equilibria in magmatic systems at elevated temperatures  
1109 and pressures. *Contrib Mineral Petrol*, 119: 197-212  
1110

1111 Giordano D, Mangiacapra A, Potuzak M, Russell JK, Romano C, Dingwell DB, Di  
1112 Muro A (2006) An expanded non-Arrhenian model for silicate melt viscosity: A

1113 treatment for metaluminous, peraluminous and peralkaline liquids. Chem Geol, 229:  
1114 42-56  
1115  
1116 Giordano D, Russell JK, Dingwell DB (2008) Viscosity of magmatic liquids: A model.  
1117 Earth Planet Sc Lett, 271: 123-134  
1118  
1119 Gualda GA, Ghiorso MS, Lemons RV, Carley TL (2012) Rhyolite-MELTS: a modified  
1120 calibration of MELTS optimized for silica-rich, fluid-bearing magmatic systems. J  
1121 Petrol, 53: 875-890  
1122  
1123 Guest JE, Gaspar JL, Cole PD, Queiroz G, Duncan AM, Wallenstein N, Ferreira T,  
1124 Pacheco JM (1999) Volcanic geology of Furnas volcano, São Miguel, Azores. J  
1125 Volcanol Geoth Res, 92: 1-29  
1126  
1127 Guest JE, Pacheco JM, Cole PD, Duncan AM, Wallenstein N, Queiroz G, Gaspar JL,  
1128 Ferreira T (2015) The volcanic history of Furnas Volcano, São Miguel, Azores. In:  
1129 Gaspar JL, Guest JE, Duncan AM, Barriga FJAS, Chester DK (eds) 2015, Volcanic  
1130 geology of São Miguel island (Azores archipelago). Geol Soc Lond Mem, 44: 125-  
1131 134  
1132  
1133 Harris C (1983) The petrology of lavas and associated plutonic inclusions of  
1134 Ascension Island. J Petrol, 24: 424-470  
1135  
1136 Hildreth W (1981) Gradients in silicic magma chambers: Implications for lithospheric  
1137 magmatism. J Geophys Res, 86: 10153-10192  
1138

1139 Hildreth W (2004) Volcanological perspectives on Long Valley, Mammoth Mountain,  
1140 and Mono Craters: several contiguous but discrete systems. J Volcanol Geoth Res,  
1141 136: 169-198  
1142

1143 Hildreth W, Wilson CJN (2007) Compositional zoning of the Bishop Tuff. J Petrol, 48:  
1144 951-999  
1145

1146 Huppert HE, Sparks RSJ (1984) Double diffusive convection due to crystallisation in  
1147 magmas. Ann Rev Earth Planet Sci, 12: 11-37  
1148

1149 Jeffery AJ (2016) Petrogenesis and contrasting eruption styles of peralkaline silicic  
1150 magmas from Terceira and São Miguel, Azores. Unpublished PhD thesis, Keele  
1151 University, U.K.  
1152

1153 Jeffery AJ, Gertisser R, Jackson RA, O'Driscoll B, Kronz A (In Press) On the  
1154 compositional variability of dalyite,  $K_2ZrSi_6O_{15}$ : a new occurrence from Terceira,  
1155 Azores. Mineral Mag  
1156

1157 Jones G, Chester DK, Shooshtarian F (1999) Statistical analysis of the frequency of  
1158 eruptions at Furnas Volcano, São Miguel, Azores. J Volcanol Geoth Res, 92: 31-38  
1159

1160 Kaula WM (1970) Earth's gravity field: relation to global tectonics. Science, 169: 982-  
1161 985  
1162

1163 Krause DC, Watkins, ND (1970) North Atlantic crustal genesis in the vicinity of the  
1164 Azores. Geophys J Roy Astr S, 19: 261-283  
1165

1166 Larrea P, Galé C, Ubide T, Widom E, Lago M, França Z (2014) Magmatic evolution  
1167 of Graciosa (Azores, Portugal). *J Petrol*, 55: 2125-2154  
1168

1169 Leake BE, Woolley AR, Arps CES, Birch WD, Gilbert MC, Grice JD, Hawthorne FC,  
1170 Kato A, Kisch HJ, Krivovichev VG, Linthout K, Laird J, Mandarino JA, Maresch WV,  
1171 Nickel EH, Rock NMS, Schumacher JC, Smith DC, Stephenson NCN, Ungaretti L,  
1172 Whittaker EJW, Youzhi G (1997) Nomenclature of amphiboles: report of the  
1173 subcommittee on amphiboles of the international mineralogical association,  
1174 commission on new minerals and mineral names. *Can Mineral*, 35: 219-246  
1175

1176 Le Bas MJ, Le Maitre RW, Streckeisen A, Zanettin B (1986) A chemical classification  
1177 of volcanic rocks based on the Total Alkali-Silica diagram. *J Petrol*, 27: 745-750  
1178

1179 Lepage LD (2003) ILMAT: an excel worksheet for ilmenite-magnetite  
1180 geothermometry and geobarometry. *Comput Geosci*, 29: 673-678  
1181

1182 Lindsley DH, Spencer KJ (1982) Fe-Ti oxide geothermometry: reducing analyses of  
1183 coexisting Ti-magnetite (Mt) and ilmenite (Ilm). *Eos Trans Amer Geophys Union*, 63:  
1184 471  
1185

1186 Luis JF, Miranda JM (2008) Reevaluation of magnetic chrons in the North Atlantic  
1187 between 35°N and 47°N: implications for the formation of the Azores Triple Junction  
1188 and associated plateau. *J Geophys Res*, 113: B10105  
1189

1190 Macdonald R (2012) Evolution of peralkaline silicic complexes: lessons from the  
1191 extrusive rocks. *Lithos*, 152: 11-22  
1192

1193 Macdonald R, Sumita M, Schmincke HU, Bagiński B, White JC, Ilnicki SS (2015)  
1194 Peralkaline felsic magmatism at the Nemrut volcano, Turkey: impact of volcanism on  
1195 the evolution of Lake Van (Anatolia) IV. *Contrib Mineral Petrol*, 169: 34  
1196  
1197 Machado F (1972) Acid volcanoes of San Miguel, Azores. *Bull Volc*, 36: 319-327  
1198  
1199 Mahood GA (1984) Pyroclastic rocks and calderas associated with strongly  
1200 peralkaline magmatism. *J Geophys Res*, 89: 8540-8552  
1201  
1202 Markl G, Marks MAW, Frost BR (2010) On the controls of oxygen fugacity in the  
1203 generation and crystallisation of peralkaline melts. *J Petrol*, 51, 1831-1847  
1204  
1205 Marks MAW, Hettmann K, Schilling J, Frost BR, Markl G (2011) The mineralogical  
1206 diversity of alkaline igneous rocks: critical factors for the transition from miaskitic to  
1207 agpaitic phase assemblages. *J Petrol*, 52: 439-455  
1208  
1209 Marks MAW, Schilling J, Coulson IM, Wenzel T, Markl G (2008) The alkaline-  
1210 peralkaline Tamazeght complex, High Atlas Mountains, Morocco: mineral chemistry  
1211 and petrological constraints for derivation from a compositionally heterogeneous  
1212 mantle source. *J Petrol*, 49: 1097-1131  
1213  
1214 Mollo S, Masotta M, Forni F, Bachmann O, De Astis G, Moore G, Scarlato P (2015) A  
1215 K-feldspar-liquid hygrometer specific to alkaline differentiated magmas. *Chem Geol*,  
1216 392: 1-8  
1217  
1218 Montesinos FG, Camacho AG, Vieira R (1999) Analysis of gravimetric anomalies in  
1219 Furnas caldera (São Miguel, Azores). *J Volcanol Geoth Res*, 92: 67-81  
1220

1221 Moore RB (1990) Volcanic geology and eruption frequency, São Miguel, Azores. Bull  
1222 Volc, 52: 602-614  
1223  
1224 Moore RB (1991) Geology of three Late Quaternary stratovolcanoes on São Miguel,  
1225 Azores. U.S. Geol Sur Bull 1900, 1-46  
1226  
1227 Moore RB, Rubin M (1991) Radiocarbon dates for lava flows and pyroclastic deposits  
1228 on São Miguel, Azores. Radiocarbon, 33: 151-164  
1229  
1230 Morimoto N, Fabries J, Ferguson AK, Ginzburg IV, Ross M, Seifert FA, Zussman J,  
1231 Aoki K, Gottardi G (1988) Nomenclature of pyroxenes. Mineral Mag, 52:535-550  
1232  
1233 Mungall JE, Martin RF (1995) Petrogenesis of basalt-comendite and basalt  
1234 pantellerite suites, Terceira, Azores, and some implications for the origin of oceanic-  
1235 island rhyolites. Contrib Mineral Petrol, 119: 43-55  
1236  
1237 Neave DA, Fabbro G, Herd RA, Petrone CM, Edmonds M (2012) Melting,  
1238 differentiation and degassing at the Pantelleria volcano, Italy. J Petrol, 53: 637-663  
1239  
1240 Nielsen R (2006) Geochemical Earth Reference Model (GERM) partition coefficient  
1241 (Kd) database. Available at: [www.earthref.org/KDD/](http://www.earthref.org/KDD/)  
1242  
1243 Papale P, Moretti R, Barbato D (2006) The compositional dependence of the  
1244 saturation surface of H<sub>2</sub>O + CO<sub>2</sub> fluids in silicate melts. Chem Geol, 229: 78-95  
1245  
1246 Peccerillo A, Donati C, Santo AP, Orlando A, Yirgu G, Ayalew D (2007) Petrogenesis  
1247 of silicic peralkaline rocks in the Ethiopian rift: geochemical evidence and  
1248 volcanological implications. J Afr Earth Sci 48: 161-173

1249

1250 Pensa A, Giordano G, Cas RAF, Porreca M (2015) Thermal state and implications for  
1251 eruptive styles of the intra-Plinian and climactic ignimbrites of the 4.6 ka Fogo A  
1252 eruption sequence, São Miguel, Azores. *Bull Volc*, 77: 99

1253

1254 Pimentel A, Pacheco J, Self S (2015) The ~1000-years BP explosive eruption of  
1255 Caldeira Volcano (Faial, Azores): the first stage of incremental caldera formation. *Bull*  
1256 *Volc*, 77:42

1257

1258 Pistone M, Arzilli F, Dobson KJ, Cordonnier B, Reusser E, Ulmer P, Marone F,  
1259 Whittington AG, Mancini L, Fife JL, Blundy JD (2015) Gas-driven filter pressing in  
1260 magmas: Insights into in-situ melt segregation from crystal mushes. *Geology*, 43:  
1261 699-702

1262

1263 Putirka KD (2008) Thermometers and barometers for volcanic systems. *Rev Mineral*  
1264 *Geochem*, 69:61-120

1265

1266 Reimer PJ, Bard E, Bayliss A, Beck JW, Blackwell PG, Bronk Ramsey C, Buck CE,  
1267 Cheng H, Edwards RL, Friedrich M (2013) IntCal13 and Marine13 radiocarbon age  
1268 calibration curves 0-50,000 years cal BP. *Radiocarbon*, 55: 1869-1887

1269

1270 Renzulli A, Santi P (2000) Two-stage fractionation history of the alkali basalt-trachyte  
1271 series of Sete Cidades volcano (São Miguel island, Azores) *Eur J Mineral*, 12: 469-  
1272 494

1273

1274 Ridley WI, Watkins ND, MacFarlane DJ (1974) Chapter 12: The Oceanic Islands:  
1275 Azores. In: Nairn AEM, Stehli FG (ed.) *The Ocean Basins and Margins, Volume 2:*  
1276 *The North Atlantic*. London: *New-York/London: Plenum Press*. 445-484

1277

1278 Rooney TO, Hart WK, Hall CM, Ayalew D, Ghiorso MS, Hidalgo P, Yirgu G (2012)

1279 Peralkaline magma evolution and the tephra record in the Ethiopian Rift. Contrib

1280 Mineral Petrol, 164: 407-426

1281

1282 Scaillet B, Holtz F, Pichavant M (1998) Phase equilibrium constraints on the viscosity

1283 of silicic magmas 1. Volcanic-plutonic comparison. J Geophys Res, 103: 27257-

1284 27266

1285

1286 Scaillet B, Macdonald R (2001) Phase relations of peralkaline silicic magmas and

1287 petrogenetic implications. J Petrol, 42: 825-845

1288

1289 Scaillet B, Macdonald R (2003) Experimental constraints on the relationships

1290 between peralkaline rhyolites of the Kenya Rift Valley. J Petrol, 44: 1867-1894

1291 Schmincke HU, Weibel M (1972) Chemical study of rocks from Madeira, Porto Santo

1292 and São Miguel, Terceira (Azores). Neues Jb Mineral Abh, 117: 253-281

1293

1294 Searle R (1980) Tectonic pattern of the Azores spreading centre and triple junction.

1295 Earth Planet Sc Lett, 51: 415-434

1296

1297 Self S, Gunn BM (1976) Petrology, volume and age relations of alkaline and

1298 saturated peralkaline volcanics from Terceira, Azores. Contrib Mineral Petrol, 54:

1299 293-313

1300

1301 Sibrant ALR, Hildenbrand A, Margues FO, Wiess B, Boulesteix T, Hübscher C,

1302 Lüdmann T, Costa ACG, Catalão JC (2015) Morpho-structural evolution of a volcanic

1303 island developed inside an active oceanic rift: S. Miguel Island (Terceira Rift,

1304 Azores). J Volcanol Geoth Res, 301: 90-106



1305

1306 Sigmundsson F, Tryggvason E, Alves MM, Alves JL, Pálsson K, Ólafsson H (1995)

1307 Slow inflation of the Furnas volcano, São Miguel, Azores, suggested from initial

1308 levelling and Global Positioning System measurements. *Geophys Res Lett*, 22: 1681-

1309 1684

1310

1311 Sisson TW, Bacon CR (1999) Gas-driven filter pressing in magmas. *Geology*, 27:

1312 613-616

1313

1314 Snyder DC, Widom E, Pietruszka AJ, Carlson RW (2004) The role of open-system

1315 processes in the development of silicic magma chambers: a chemical and isotopic

1316 investigation of the Fogo A trachyte deposit, São Miguel, Azores. *J Petrol*, 45: 723-

1317 738

1318

1319 Storey M, Wolff JA, Norry MJ, Marriner GF (1989) Origin of hybrid lavas from Agua

1320 de Pau volcano, São Miguel, Azores. In: Saunders, A. D., Norry, M. J. (Eds.)

1321 *Magmatism in the Ocean Basins*. Blackwell, London, pp. 161-180

1322

1323 Stormer Jr JC (1983) The effects of recalculation on estimates of temperature and

1324 oxygen fugacity from analyses of multi-component iron-titanium oxides. *Am Mineral*,

1325 68: 586-594

1326

1327 Stuiver M, Reimer PJ, Reimer R (2010-last update) CALIB 5.0. 2 [WWW program

1328 and documentation]. Available: <http://radiocarbon.pa.qub.ac.uk/calib/> [4/25, 2015]

1329

1330 Sun S, McDonough WF (1989) Chemical and isotopic systematics of ocean basins:

1331 implications for mantle composition and processes. In: Saunders AD, Norry MJ (eds)

1332 Magmatism in the Ocean Basins, Geological Society of London Special Publication,  
1333 42:313-346  
1334

1335 Trua T, Deniel C, Mazzuoli R (1999) Crustal control in the genesis of Plio-Quaternary  
1336 bimodal magmatism of the Main Ethiopian Rift (MER): geochemical and isotopic (Sr,  
1337 Nd, Pb) evidence. Chem Geol 155: 201-231  
1338

1339 Venezky DY, Rutherford MJ (1999) Petrology and Fe-Ti oxide reequilibration of the  
1340 1991 Mount Unzen mixed magma. J Volcanol Geotherm Res, 89: 213-230  
1341

1342 Vogt PR, Jung WY (2004) The Terceira Rift as hyper-slow, hotspot-dominated  
1343 oblique spreading axis: a comparison with other slow-spreading plate boundaries.  
1344 Earth Planet Sc Lett, 218: 77-90  
1345

1346 Walker GPL, Croasdale R (1971) Two plinian-type eruptions in the Azores. J Geol  
1347 Soc London, 127: 17-55  
1348

1349 Wallenstein N (1999) Estudo da história recente e do comportamento eruptivo do  
1350 vulcão do Fogo (S.Miguel, Açores). A valiação preliminar do hazard. Unpublished  
1351 PhD thesis, Departamento de Geociências Universidades dos Açores, São Miguel  
1352 Island (Portugal)  
1353

1354 White JC, Parker DF, Ren M (2009) The origin of trachyte and pantellerite from  
1355 Pantelleria, Italy: insights from major element, trace element, and thermodynamic  
1356 modelling. J Volcanol Geoth Res, 179: 33-55  
1357

1358 Widom E, Farquhar J (2003) Oxygen isotope signatures in olivines from São Miguel  
1359 (Azores) basalts: implications for crustal and mantle processes. *Chem Geol*, 193:  
1360 237-255  
1361  
1362 Widom E, Gill JB, Schmincke HU (1993) Syenite nodules as a long-term record of  
1363 magmatic activity in Agua de Pau Volcano, São Miguel, Azores. *J Petrol*, 34: 929-953  
1364  
1365 Widom E, Schmincke HU, Gill JB (1992) Processes and timescales in the evolution  
1366 of a chemically zoned trachyte: Fogo A, São Miguel, Azores. *Contrib Mineral Petrol*,  
1367 111: 311-328  
1368  
1369 Wolff JA (1987) Crystallisation of nepheline syenite in a subvolcanic magma system:  
1370 Tenerife, Canary Islands. *Lithos*, 20: 207-223  
1371  
1372 Zanon V (2015) Conditions for mafic magma storage beneath fissure zones at  
1373 oceanic islands. The case of São Miguel island (Azores archipelago). In: Caricchi, L.,  
1374 Blundy, J. D. (eds) *Chemical, physical and temporal evolution of magmatic systems*,  
1375 Geological Society, London, Special Publications, 422, doi: 10.1144/SP422.4  
1376  
1377 Zanon V, Kueppers U, Pacheco JM, Cruz I (2013) Volcanism from fissure zones and  
1378 the Caldeira central volcano of Faial Island, Azores archipelago: geochemical  
1379 processes in multiple feeding systems. *Geol Mag*, 150: 536-555  
1380  
1381 Zanon V, Pimentel A (2015) Glasses, melts, and Fluids, as tools for understanding  
1382 volcanic processes and hazards. Spatio-temporal constraints on magma storage and  
1383 ascent conditions in a transtensional tectonic setting: the case of the Terceira island  
1384 (Azores). *Am Mineral*, 100: 795-805

1385 **Figure captions**

1386

1387 **Figure 1** Location maps of Furnas volcano **a)** Summary map of the major tectonic  
1388 features associated with the Azores triple junction, constructed after Searle (1980),  
1389 Freire Luis et al. (1994), and Vogt and Jung (2004) **b)** Schematic map of São Miguel,  
1390 highlighting the three active central volcanoes and fissure volcanic systems (adapted  
1391 from Moore 1990) **c)** Summary map of the caldera complex of Furnas volcano and  
1392 the neighbouring Povoação caldera. The lava domes associated with the Furnas E, I  
1393 and J eruptions are labelled. After Guest et al. (1999) and Cole et al. (1999)

1394

1395 **Figure 2** Representative photomicrographs for the UFG **a)** Euhedral alkali feldspar  
1396 phenocryst in vesicular glass **b)** Highly resorbed and embayed alkali feldspar  
1397 phenocryst **c)** Clinopyroxene phenocryst with numerous Fe-Ti oxide and apatite  
1398 inclusions **d)** Subhedral biotite phenocryst in highly vesicular glass **e)** Alkali feldspar  
1399 glomerocryst with irregular internal boundaries, surrounded by a medium grained,  
1400 trachytic groundmass (Furnas I lava dome) **f)** Clinopyroxene phenocryst with  
1401 distinctive oxidation rim and numerous Fe-Ti oxide inclusions (Furnas J lava dome)  
1402 **g)** Large, reacted biotite phenocryst remnant (Furnas I lava dome) **h)** Na-Ca  
1403 amphibole, aegirine-augite, and sodalite groundmass crystals, surrounded by alkali  
1404 feldspar laths in the Furnas I lava dome. Abbreviations used: Fsp = alkali feldspar,  
1405 Cpx = clinopyroxene, Bt = biotite, Amp = amphibole, Sdl = sodalite

1406

1407 **Figure 3** UFG compositions plotted into the Total-Alkali-Silica classification scheme  
1408 of Le Bas et al. (1986). Whole rock analyses, groundmass glass, and melt inclusions  
1409 are plotted together, and show uniform trachytic compositions

1410

1411 **Figure 4** Trace element variation diagrams for the UFG **a)** REE compositions of the  
1412 UFG. All data are normalised relative to chondrite values and plotted against typical  
1413 oceanic island basalt (OIB) compositions (Sun and McDonough 1989) **b)** Multi-  
1414 element trace-element variation diagram of the UFG. All data are normalized to  
1415 primitive mantle values (Sun and McDonough 1989)

1416

1417 **Figure 5** Mineral compositions of the UFG **a)** Alkali feldspar compositions plotted into  
1418 the ternary An-Ab-Or system **b)** Clinopyroxene compositions plotted into the  
1419 pyroxene quadrilateral and, where relevant, the ternary Qd-Jd-Aeg system (Morimoto  
1420 et al. 1988) and the ternary Di-Hd-Aeg system **c)** Fe-Ti oxide compositions plotted  
1421 into the  $\text{TiO}_2\text{-FeO-Fe}_2\text{O}_3$  ternary system **d)** Biotite compositions for the Furnas H, I,  
1422 and J pumice falls, and the Furnas J syenites **e)** Na-Ca amphibole compositions for  
1423 the Furnas I lava dome and the Furnas J syenites, plotted in the scheme of Leake et  
1424 al. (1997) **f)** Na-amphibole compositions for the Furnas J syenites, plotted in the  
1425 scheme of Leake et al. (1997)

1426

1427 **Figure 6** Summarised results of thermobarometry and hygrometry **a)** T-f $\text{O}_2$  estimates  
1428 for Furnas J derived from two-oxide models. FMQ buffer reaction curve calculated at  
1429 100 MPa **b)** T-H $_2\text{O}_{\text{melt}}$  estimates produced via alkali feldspar-melt models

1430

1431 **Figure 7** Variations in  $\text{TiO}_2$  and  $\text{Al}_2\text{O}_3$  content of clinopyroxene, plotted with reference  
1432 to the high and low pressure trends identified at nearby Sete Cidades (Beier et al.  
1433 2006)

1434

1435 **Figure 8** Results of trace element modelling. Calculated Rayleigh fractionation  
1436 curves are labelled with RFC and batch melting curves are labelled with BM. Ticked  
1437 intervals represent 10 % fractionation or partial melting. Analyses derived from ED-

1438 XRF are given with filled symbols, whilst those produced via WD-XRF or ICP-MS are  
1439 shown with open symbols. Parental compositions are shown with a black circle

1440

1441 **Figure 9** Results of Rhyolite-MELTS fractional crystallisation modelling from an alkali  
1442 basalt (black circle). Modelled curves produced at 150 MPa, with an initial water  
1443 content of 1.5 wt. %

1444

1445 **Figure 10** A schematic model for the structure of the shallow plumbing system of  
1446 Furnas volcano. A density stratified magma reservoir with a trachytic cap is  
1447 envisaged, at depths between 3 and 5 km **a)** Syenite ejecta derived from an upper  
1448 solidification front may be sampled during eruption. The solidified remnants of pre-  
1449 UFG magma systems may provide an additional source of syenitic xenoliths **b)**  
1450 Interaction between individually convecting layers introduces an antecrystic  
1451 clinopyroxene population to the cap, whilst rapidly settling feldspars form a Sr-, and  
1452 Ba-rich feldspar layer. Settling biotite phenocrysts originating from uppermost METs  
1453 are heavily resorbed by hotter underlying LETs and intermediates

1454

1455 **Figure 11** Variations in trace element concentrations, temperature estimates, and  
1456  $H_2O_{melt}$  estimates with stratigraphic height. Relevant errors are shown with red lines  
1457 at the top of each column. Modal mineralogy: alkali feldspar = blue, biotite = orange,  
1458 clinopyroxene = green, Fe-Ti oxides = grey

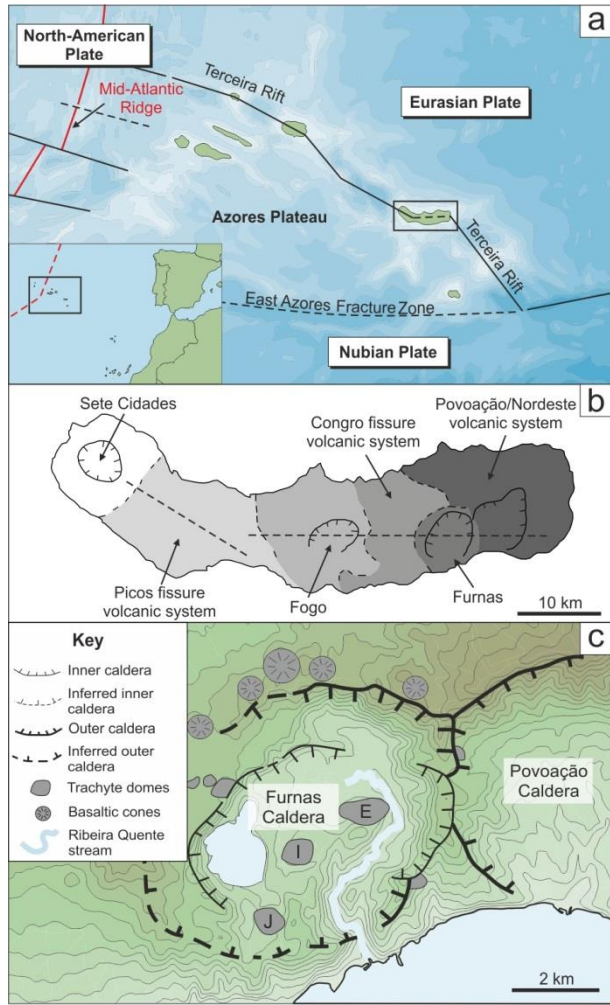
1459

1460 **Figure 12** A schematic model for cyclic replenishment and eruption phases  
1461 throughout the UFG, based upon trace element compositions. See text for  
1462 discussion. FC = fractional crystallisation

1463

1464

1465



**Figure 1**

1466

1467

1468

1469

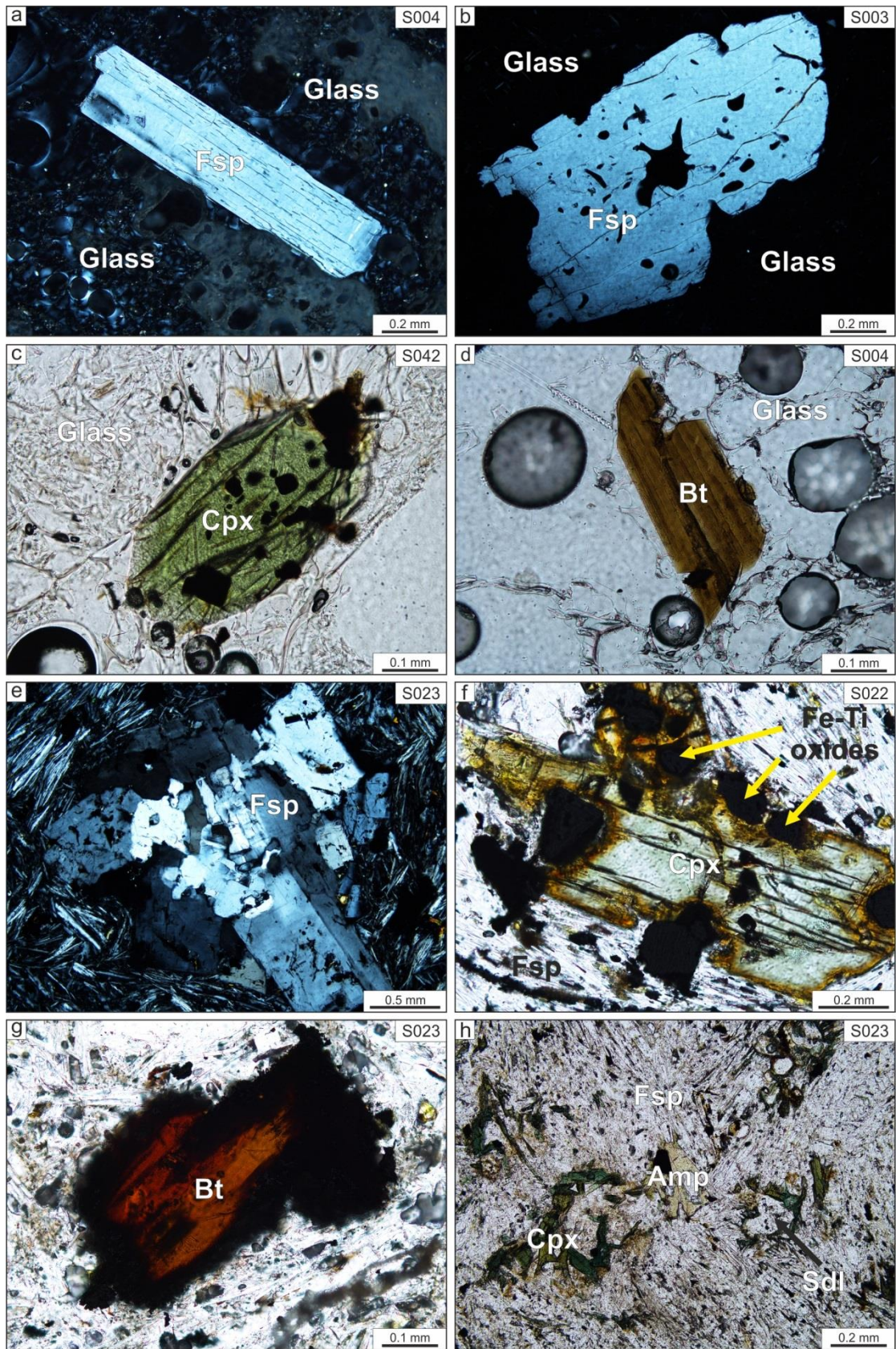
1470

1471

1472

1473

1474



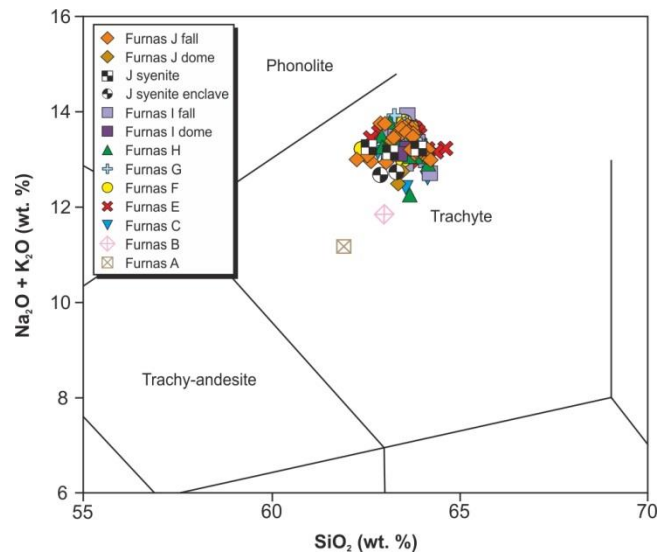
1475

1476

1477

Figure 2





**Figure 3**

1478

1479

1480

1481

1482

1483

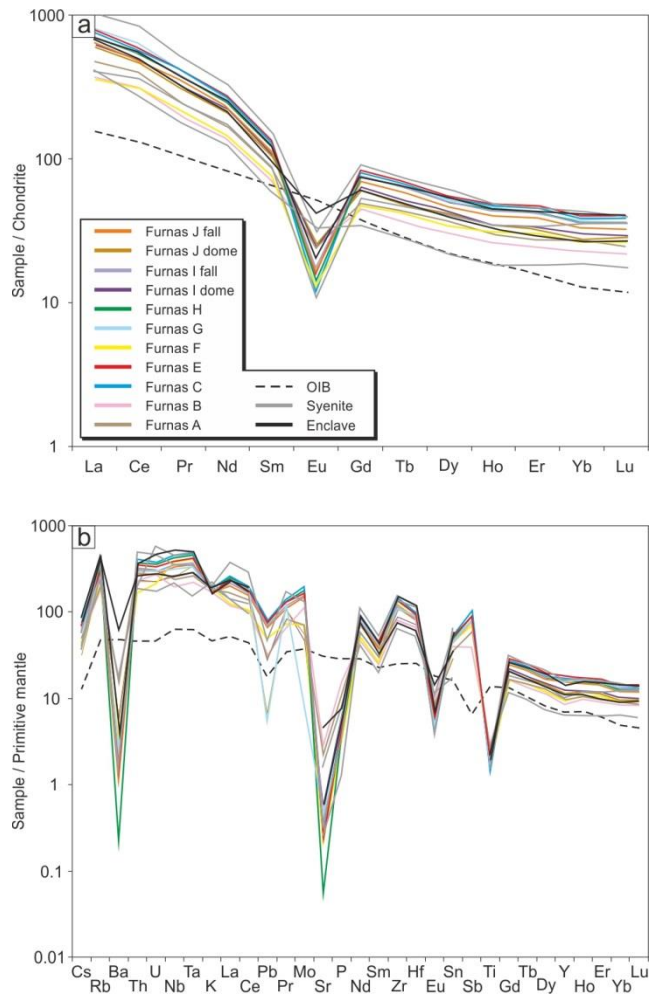
1484

1485

1486

1487

1488



1489

1490

**Figure 4**

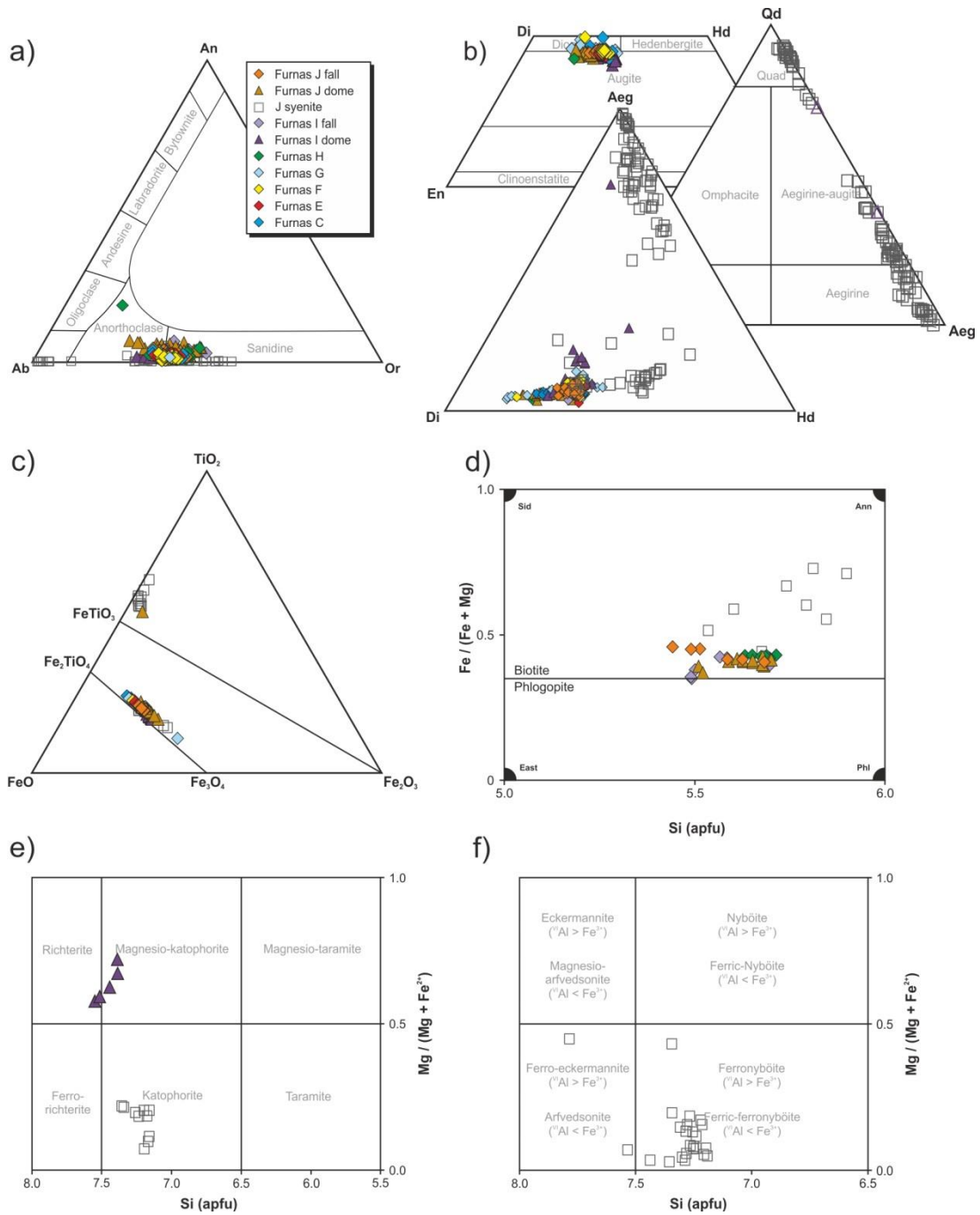


Figure 5

1491

1492

1493

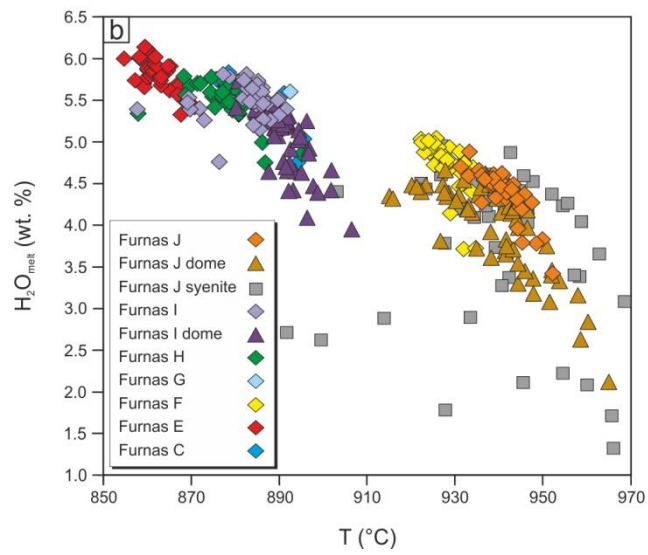
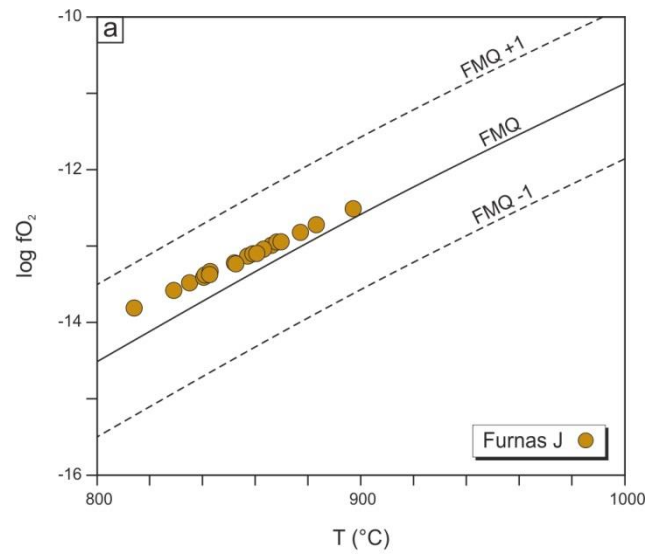
1494

1495

1496

1497

1498



**Figure 6**

1499

1500

1501

1502

1503

1504

1505

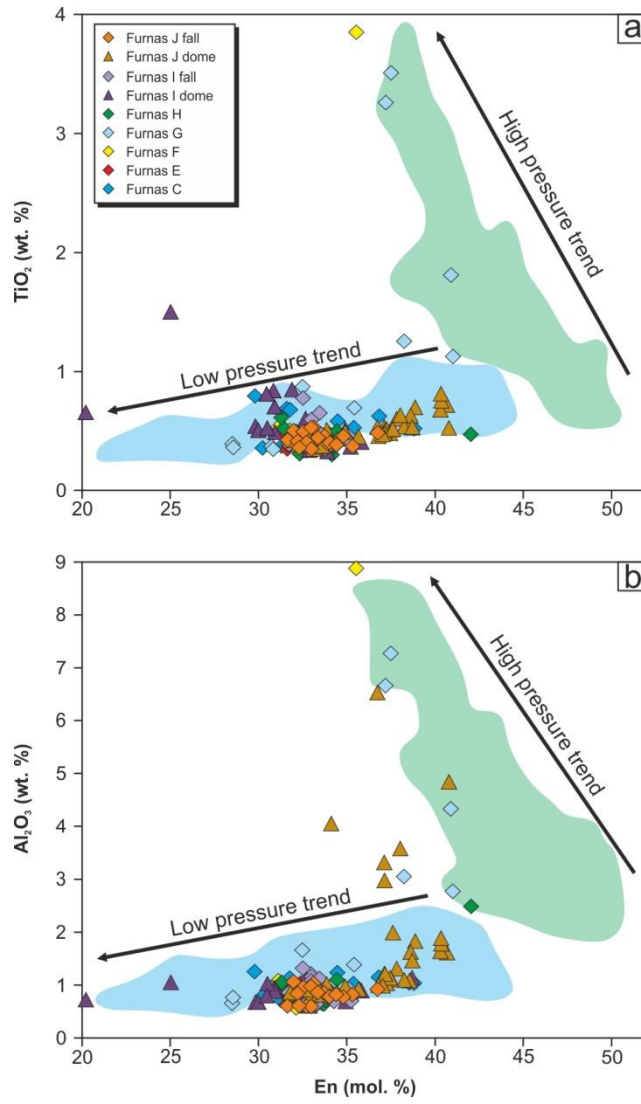
1506

1507

1508

1509

1510



**Figure 7**

1511

1512

1513

1514

1515

1516

1517

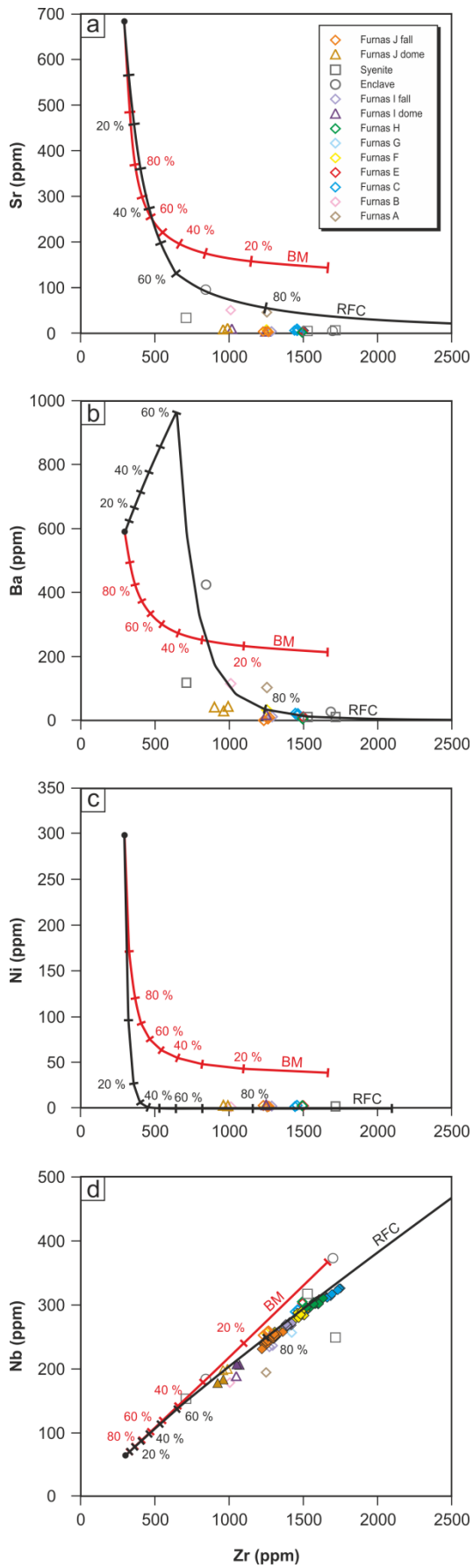
1518

1519

1520

1521

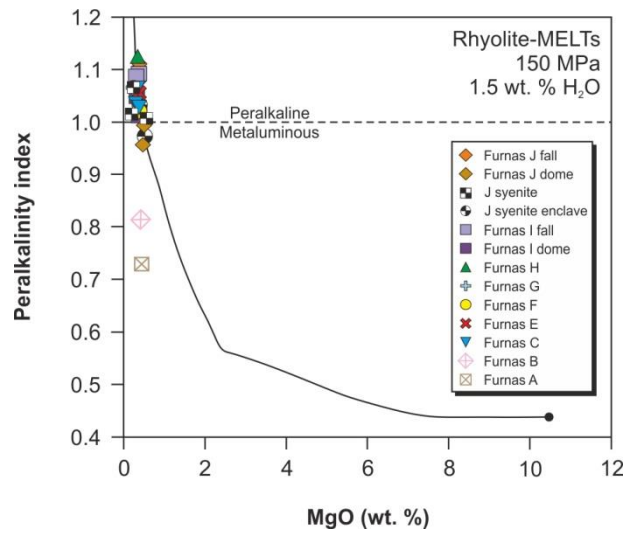
1522



1523

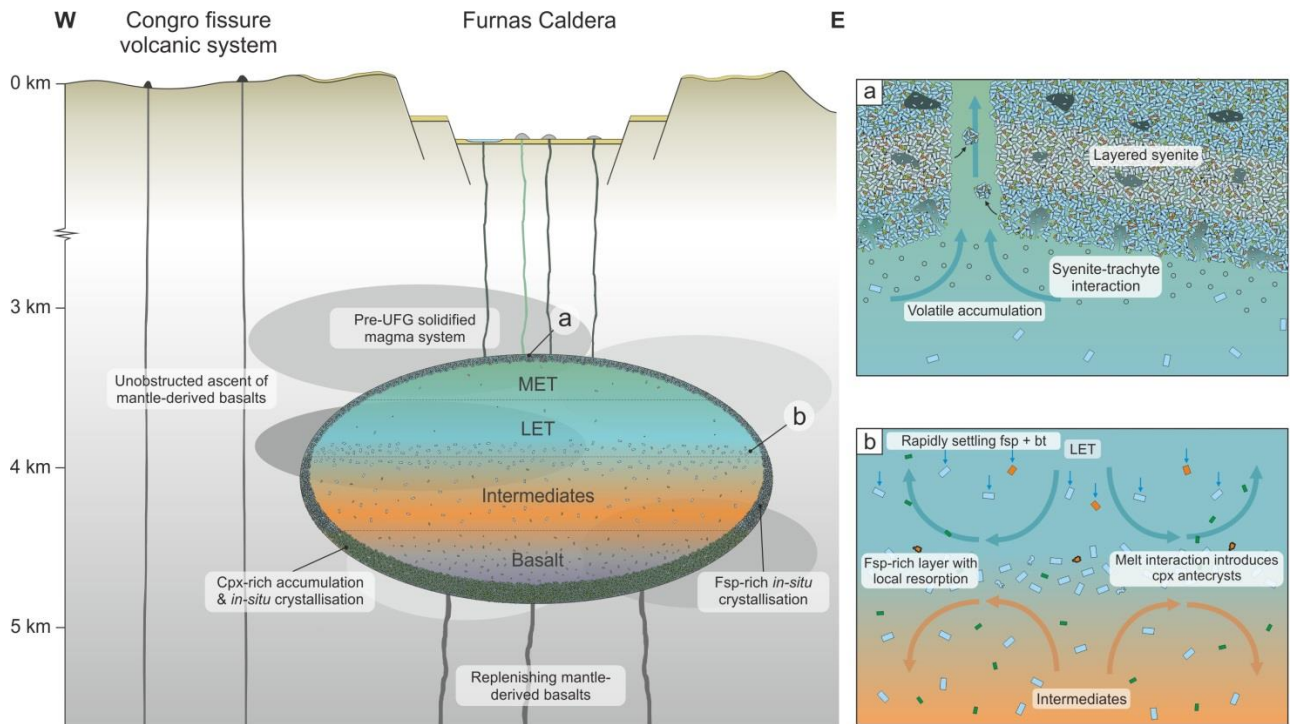
1524

Figure 8



**Figure 9**

- 1525
- 1526
- 1527
- 1528
- 1529
- 1530
- 1531
- 1532
- 1533
- 1534
- 1535
- 1536
- 1537
- 1538
- 1539
- 1540
- 1541
- 1542
- 1543
- 1544



**Figure 10**

1545

1546

1547

1548

1549

1550

1551

1552

1553

1554

1555

1556

1557

1558

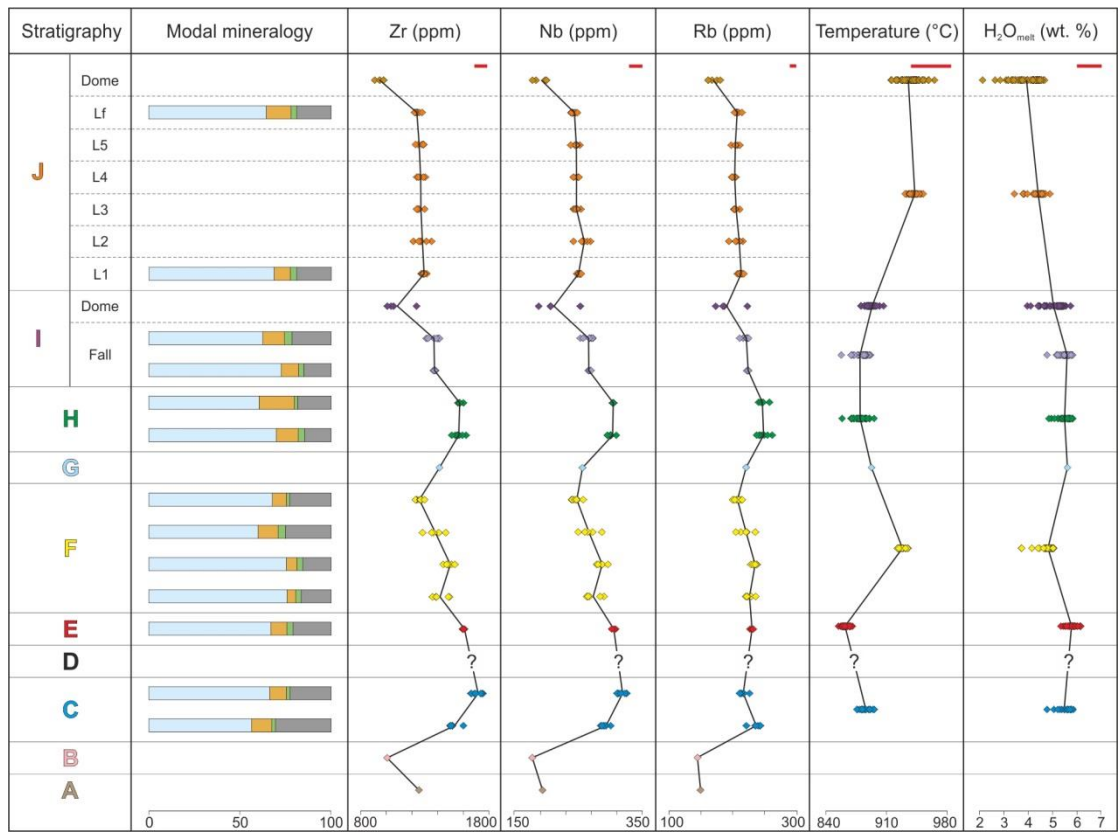
1559

1560

1561

1562





1563

1564

Figure 11

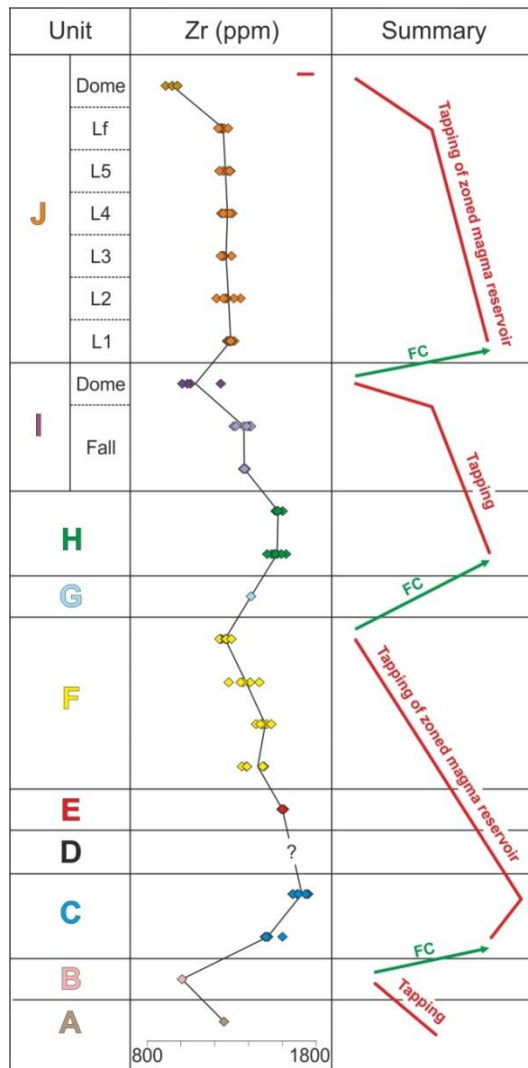


Figure 12

1565

1566

1567

1568

1569

1570

1571

1572

1573

1574

1575

1576

1577

**Table 1** Partition coefficients used for trace element modelling

Step 1 RFC and batch melting								
	Pl	Cpx	OI	Mt	Bt	Ap	Alk fsp	Ilm
Nb	0.010	0.100	0.010	0.900	0.088			
Zr	0.010	0.120	0.060	0.700	0.017			
Sr	10.000	0.700	0.020		0.700	1.200		
Ba	0.300	0.040	0.030		3.680	0.050		
Ni	0.270	2.500	34.000	48.000	23.900			
Step 2 RFC								
Nb				0.180	0.740		0.004	11.600
Zr				0.250	0.100		0.030	0.074
Sr				0.000	0.490	2.000	2.300	0.740
Ba				0.100	10.000	8.000	5.900	
Ni					89.625	0.450	0.905	6.200

All coefficients sourced from the GERM database (Nielsen, 2006), see Online Resource 4 for details. pl = plagioclase, cpx = clinopyroxene, ol = olivine, mt= Ti-magnetite, bt = Biotite, ap = apatite, alk fsp – alkali feldspar, ilm = ilmenite

1578  
1579  
1580  
1581

1582

1583

1584

1585

1586

1587

1588

1589

1590

1591

1592

1593

1594

1595

1596

1597

1598

1599

1600

1601

1602

1603

**Table 2** Mineral compositions used for major element mass balance modelling

Mineral	Pl	Alk fsp	Ol	Cpx	Mt-1	Mt-2	Ilm	Bt	Ap
SiO <sub>2</sub>	54.71	65.15	40.44	52.26				36.70	
TiO <sub>2</sub>		0.09		0.84	14.65	20.47	48.97	6.51	
Al <sub>2</sub> O <sub>3</sub>	29.28	19.00		2.21	0.99	4.39	0.03	13.30	
Fe <sub>2</sub> O <sub>3</sub>				3.37	40.01	26.61		6.85	
FeO	0.82	0.37	11.75	0.38	39.25	42.36	42.26	9.01	
MnO			0.22	0.12	2.63	0.52	3.93	0.41	
MgO		0.01	47.32	16.71	1.77	4.99	3.31	13.91	
CaO	10.30	0.54	0.28	22.37					55.70
Na <sub>2</sub> O	4.83	6.87		0.28				0.97	
K <sub>2</sub> O	0.43	6.69						8.71	
P <sub>2</sub> O <sub>5</sub>									41.82
H <sub>2</sub> O									0.59

Mineral compositions for alkali feldspar (alk fsp), Ti-magnetite (Mt-1), and ilmenite (Ilm) were taken from the dataset of this study. Compositions for plagioclase (Pl), olivine (Ol), clinopyroxene (Cpx), Ti-magnetite (Mt-2), and biotite (Bt) were taken from Beier et al. (2006)

1604

1605

1606

1607

1608

1609

1610

1611

1612

1613

1614

1615

1616

1617

1618

1619

1620

1621

1622

1623

1624

1625

1626

1627

1628

1629

1630

1631

**Table 3** Results of major element mass balance modelling

Step 1: Alkali basalt to LET						
Sample	Liquid Composition			Fractionating Phases		
	Basalt		LET			
	98SM33		S022			
	Obs	Calc	Obs			
SiO <sub>2</sub>	46.22	46.17	63.20	Cpx	45.1	
TiO <sub>2</sub>	3.41	3.29	0.57	Plg	23.9	
Al <sub>2</sub> O <sub>3</sub>	10.89	10.93	17.83	Mt	12.3	
FeO <sup>T</sup>	11.44	11.48	3.34	Bt	10.2	
MnO	0.17	0.19	0.22	Ol	7.1	
MgO	11.53	11.53	0.38	Ap	1.6	
CaO	12.01	12.04	0.88			
Na <sub>2</sub> O	2.12	2.13	7.37			
K <sub>2</sub> O	1.59	1.65	6.03			
P <sub>2</sub> O <sub>5</sub>	0.63	0.6	0.09			
$\Sigma r^2$	0.023					
F	0.125					
Step 2: LET to MET						
Sample	Liquid Composition			Fractionating Phases		
	LET		MET			
	S022		SM 1-2			
	Obs	Calc	Obs			
SiO <sub>2</sub>	63.20	63.28	63.16	Alk fsp	94.1	
TiO <sub>2</sub>	0.57	0.57	0.42	Bt	3.2	
Al <sub>2</sub> O <sub>3</sub>	17.83	17.86	17.40	Mt	1.4	
FeO <sup>T</sup>	3.34	3.44	4.21	Ap	0.7	
MnO	0.22	0.22	0.30	Ilm	0.6	
MgO	0.38	0.4	0.32			
CaO	0.88	0.83	0.77			
Na <sub>2</sub> O	7.37	7.31	7.75			
K <sub>2</sub> O	6.03	5.94	5.49			
P <sub>2</sub> O <sub>5</sub>	0.09	0.16	0.06			
$\Sigma r^2$	0.018					
F	0.618					

Abbreviations: Ap = apatite, plg = plagioclase, alk fsp = alkali feldspar, cpx = clinopyroxene, bt = biotite, ol = olivine, mt – magnetite, ilm = ilmenite

1632  
1633  
1634

1635

1636

1637

1638

1639

1640

1641

1642

1643

1644

1645

1646

1647

1648

1649

1650

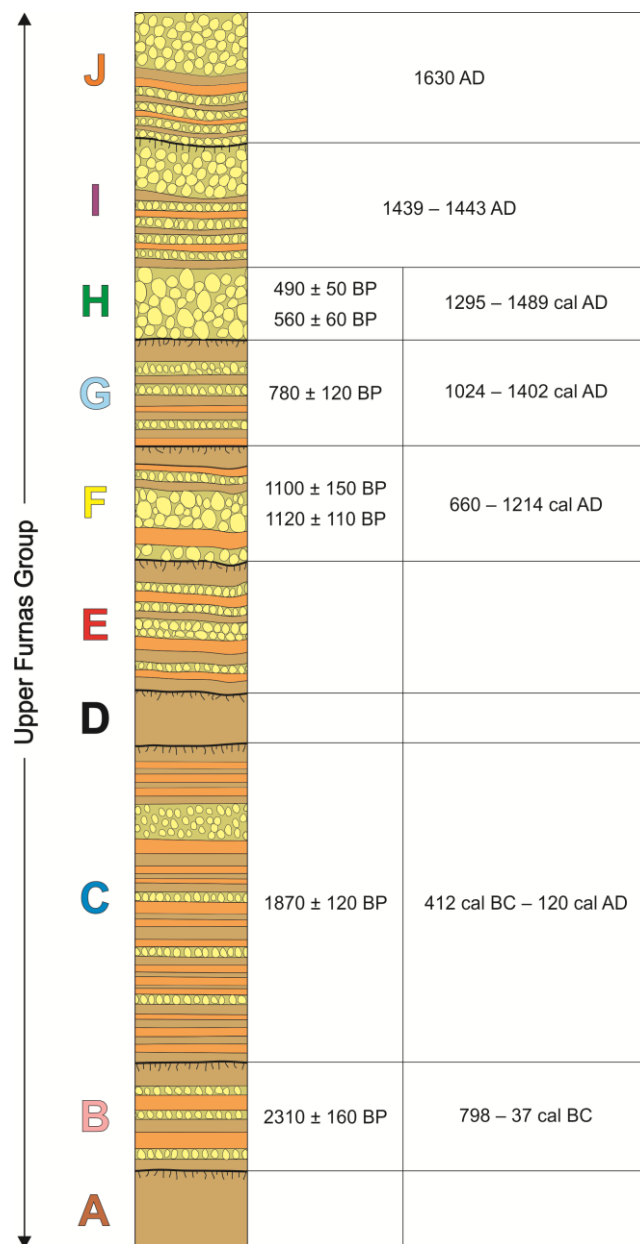
1651

1652

1653

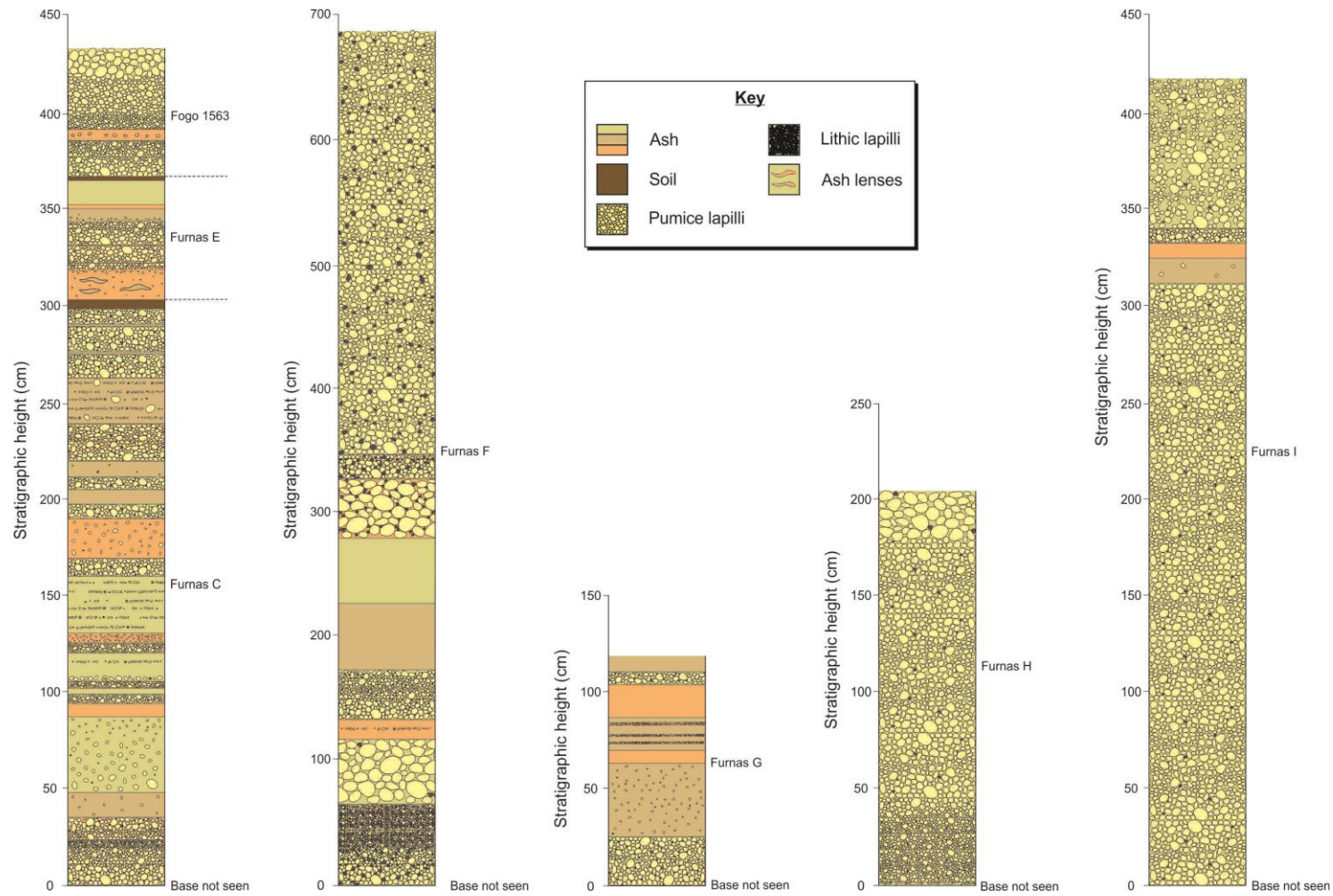
1654

1655



1656 **Online Resource 1:** Schematic stratigraphy of the Upper Furnas Group (UFG), after  
 1657 Guest et al. (1999). Uncalibrated radiocarbon ages were taken from Moore and  
 1658 Rubin (1991) and Guest et al. (1999), and converted into calibrated 2 sigma age  
 1659 ranges using Calib 7.1 (Stuiver et al. 2010) and the IntCal13 calibration dataset  
 1660 (Reimer et al. 2013)

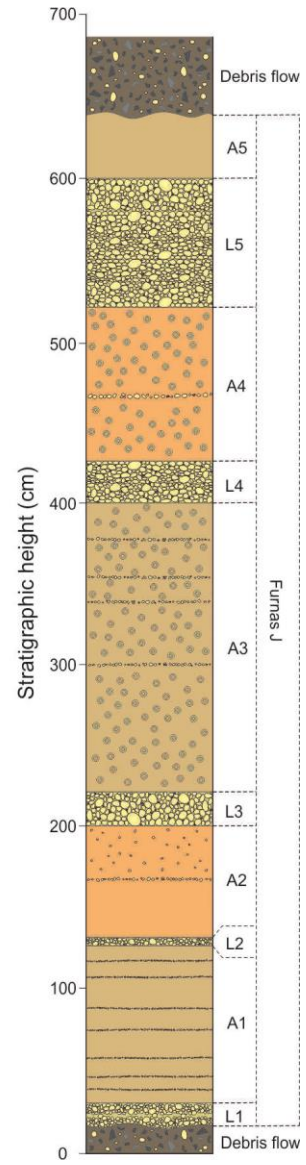
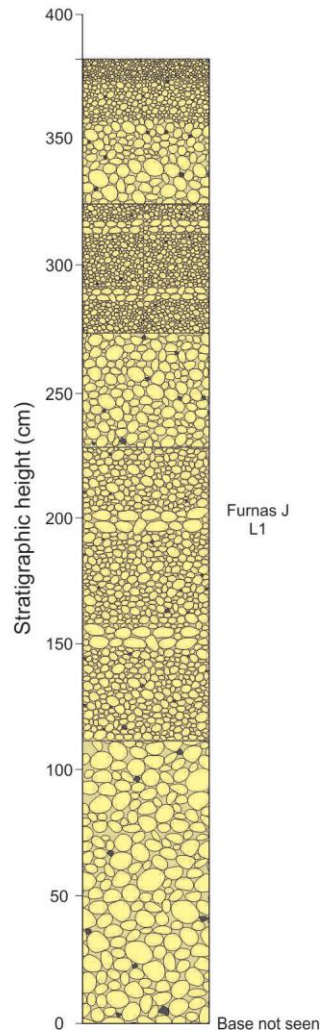
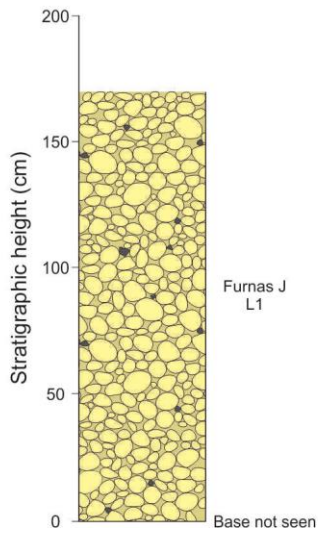
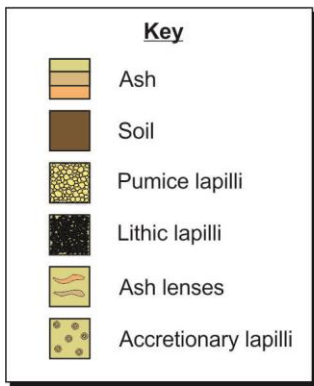
1661



1662

1663

Online Resource 2: Logs of the UFG deposits



1664

1665

1666

1667

1668

1669

1670

1671

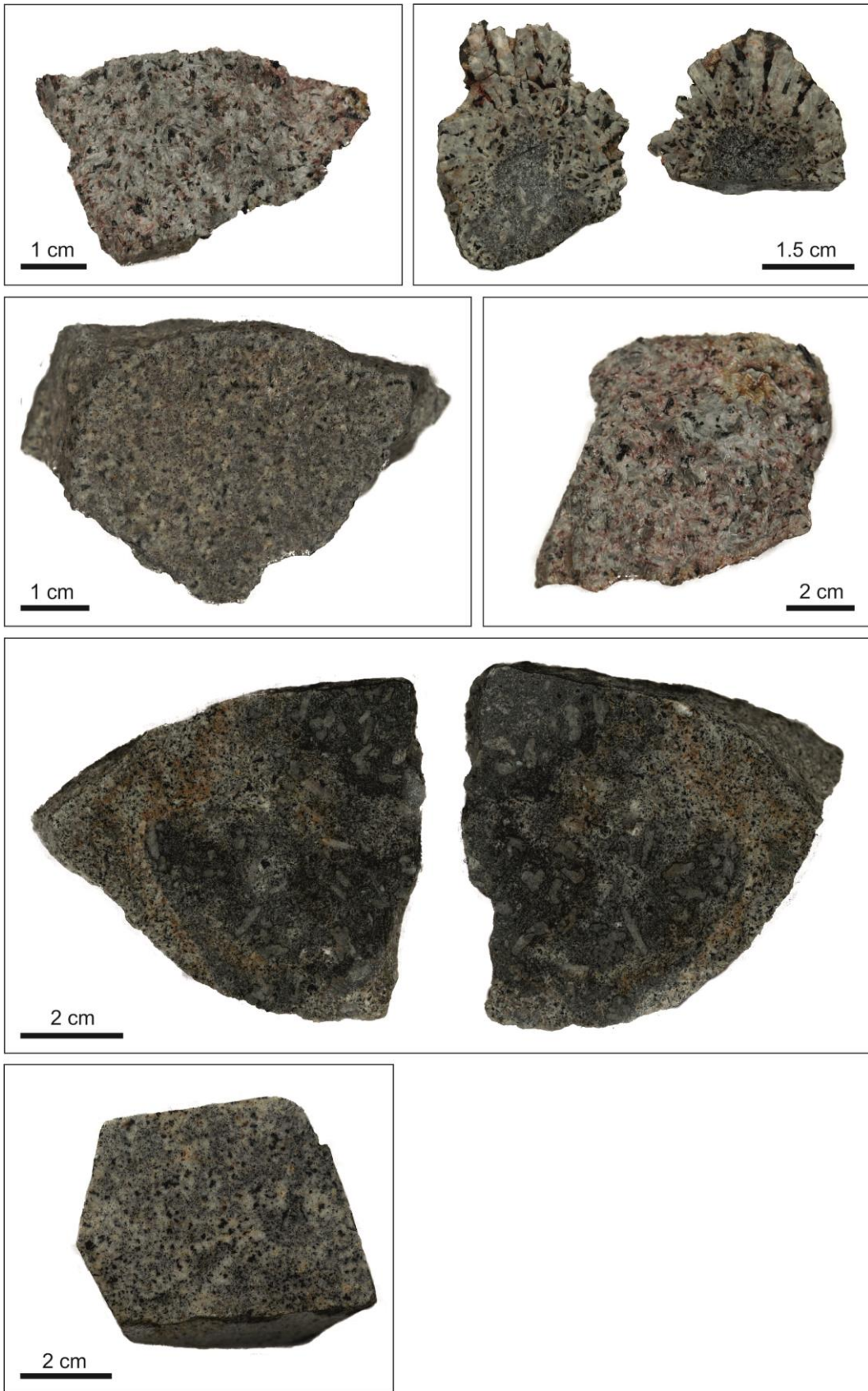
1672

1673

1674

**Online Resource 2 continued**



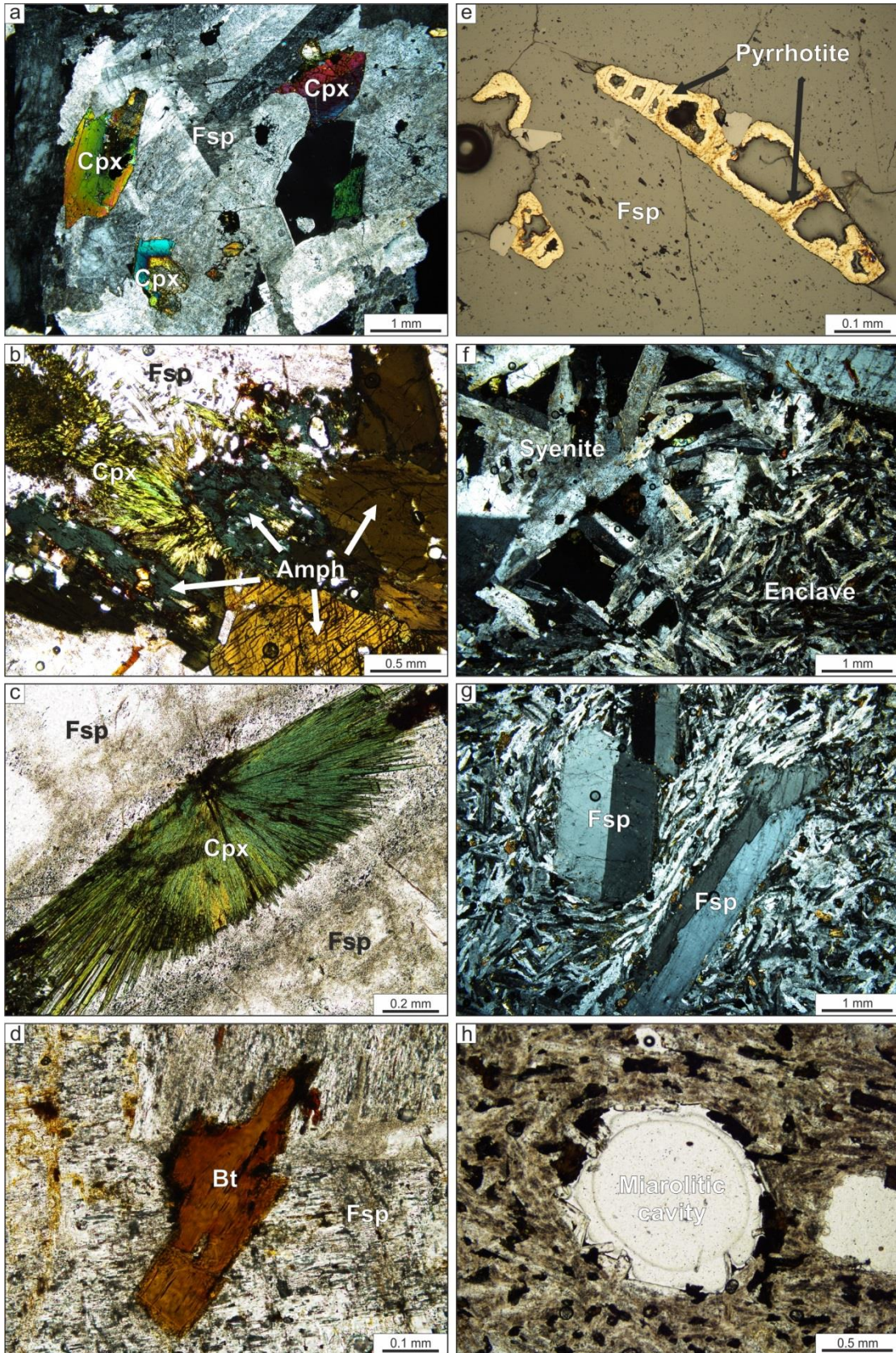


1675

1676

1677

**Online Resource 3: Representative hand specimen photographs of syenite  
ejecta and syenite-hosted enclaves**



1678

1679

1680

Online Resource 3 continued: Representative photomicrographs of syenite  
ejecta and syenite-hosted enclaves

## 1681 **Online Resource 4: Detailed methodology**

### 1682 ***Whole rock analyses***

1683 For WD-XRF analyses, samples were cleaned using distilled water and weathered  
1684 surfaces were removed, before crushing using a jaw-crusher and an agate mill.  
1685 Fused glass discs for major element analysis were prepared using 0.7 g of rock  
1686 powder mixed with 3.5 g of lithium metaborate. Trace element concentrations were  
1687 determined using PVC bound pressed powder pellets. Loss on ignition (LOI) was  
1688 reported as weight difference after ignition at 1,000 °C for two hours. Data quality  
1689 was evaluated by repeated analyses of the following standards: WS-E, OUG94,  
1690 GSP-2, W2a, and AC-E. The mean deviation from the accepted standard values was  
1691 < 10 % and reproducibility was < 3 % for major elements and < 10 % for trace  
1692 elements.

1693 Samples analysed via ICP-MS at the Open University, U.K., were prepared  
1694 using a standard HF-HNO<sub>3</sub> digestion procedure (e.g. Eggins et al. 1997). The  
1695 instrument was calibrated using up to six international standards (BCR1, BIR1, AC,  
1696 RGM1, BHVO1-2, AGV1), using recommended element concentrations. Instrumental  
1697 drift was monitored by and corrected using an internal standard solution containing  
1698 Be, In, Rh, Tm, Re and Bi, which was introduced on-line during analysis. Residual  
1699 drift in individual elements was corrected externally using replicate determinations of  
1700 a representative sample after every 10 unknowns. Samples analysed via ICP-AES  
1701 and ICP-MS at Bureau Veritas Mineral Laboratories, Canada, were cleaned to  
1702 remove altered surfaces, and crushed in an agate mill prior to drying at 60 °C, mixing  
1703 with a LiBO<sub>2</sub>/Li<sub>2</sub>B<sub>4</sub>O<sub>7</sub> flux and dilute nitric digestion. LOI was reported as weight  
1704 difference after ignition for two hours at 1,000 °C. Standard analyses indicate  
1705 accuracy of < 5 % for major elements and < 7 % for trace elements, whilst  
1706 reproducibility was < 1 %

1707 For samples analysed by XRF at Keele University, U.K., a minimum of five  
1708 clasts were analysed for each eruptive unit. Clasts were analysed individually where  
1709 possible. Samples had any visibly weathered surfaces removed prior to soaking in  
1710 distilled water for at least 24 hours and crushing in an agate mill. Elemental  
1711 determinations were performed on PVC bound pressed powder pellets, using 8 g of  
1712 rock powder. Reproducibility was found to be in the order of 0.5 % for Zr, and < 2.5 %  
1713 for Nb and Sr. The data were calibrated with a calibration curve comprising up to 34  
1714 igneous standards, and up to 23 secondary standards comprising various igneous  
1715 rocks from Furnas, and also from nearby Terceira Island, which indicate average  
1716 accuracies of  $\leq 6$  % for Rb, Zr, and Nb, and 11 % for Sr. Repeat analyses indicate  
1717 precision of < 1 % for Zr, Nb, and Sr, and < 2 % for Rb.

1718

#### 1719 ***Electron microprobe analyses***

1720 For mineral phases, peak counting times per element were 10 to 30 seconds using a  
1721 10- $\mu\text{m}$  defocused beam, an acceleration voltage of 20 kV and a beam current of 20–  
1722 27 nA. Major elements in glass (both groundmass glass and melt inclusions) were  
1723 analysed using peak count times between 10 and 30 seconds, and 90 to 120  
1724 seconds, respectively. To minimize Na-loss, the acceleration voltage and beam  
1725 current were reduced to 15 kV and 15 nA, respectively, whilst the beam diameter  
1726 was, where possible, increased to 20  $\mu\text{m}$ . Furthermore, Na was always analysed  
1727 first, with a peak count time of 10 seconds. The following natural minerals and  
1728 synthetic materials (denoted as formulae) were used as primary calibration  
1729 standards: albite, sanidine, anorthite, wollastonite,  $\text{TiO}_2$ , rhodonite, haematite, olivine,  
1730 apatite, NaCl, ZnS,  $\text{Al}_2\text{O}_3$ , celsian,  $\text{SrTiO}_3$ ,  $\text{Cr}_2\text{O}_3$ , NiO, MgO, V, gahnite, and  
1731 celestine. Volcanic glass standards (VG-2, KN-18 and KE-12) were routinely  
1732 analysed as secondary standards (Online Resource 5). Repeat analyses of

1733 secondary standards indicate accuracy of < 4 %, and reproducibility of < 3 % (mean  
1734 standard deviation).

1735

### 1736 ***Thermobarometric models***

#### 1737 **Alkali feldspar-melt models**

1738 To minimize the error introduced by mineral-liquid disequilibrium, the  $K_{\text{Ab-Or}}$   
1739 equilibrium test proposed by Mollo et al. (2015) was applied, allowing a single  
1740 suitable liquid composition to be selected for each case. Input pressure values were  
1741 set at 100 MPa, and it was observed that a variation of 100 MPa led to a change in  
1742 temperature of only 0.1 °C, suggesting that the thermometer is not significantly  
1743 influenced by pressure variations. The standard error of estimate (SEE) associated  
1744 with the thermometer is  $\pm 23$  °C. The average temperature estimate derived from  
1745 alkali feldspar-melt thermometry was used as a primary input for feldspar-melt  
1746 hygrometry, alongside feldspar and potential equilibrium liquid compositions. The  
1747 SEE associated with these results is  $\pm 0.53$  wt. %.

1748

### 1749 ***Geochemical modelling***

#### 1750 **Mass balance and trace element modelling**

1751 Least squares mass balance calculations after Bryan et al. (1969) were performed  
1752 using the IgPet software package. The modelling was performed in two steps: 1)  
1753 fractionation from basalt to the least evolved trachyte (LET), and 2) LET to the most  
1754 evolved trachyte (MET). For a parental basalt composition, a young basaltic lava  
1755 located ~ 6 km south west of Furnas village (sample 98SM33 of Elliott et al. 2007)  
1756 was selected due to its low Zr content and high magnesium number (294 ppm and  
1757 64, respectively). For the LET and MET compositions, the UFG sample with the  
1758 lowest and highest Zr contents were chosen (SM7-1 and SM1-2, respectively). Model  
1759 results were considered acceptable if  $\sum r^2 = < 1$  (e.g. White et al. 2009). To produce

1760 models involving evolution from basalt to trachyte, mineral chemical data for  
1761 plagioclase, clinopyroxene, olivine, biotite and magnetite were taken from analyses  
1762 from Sete Cidades (Beier et al. 2006). Declining  $P_2O_5$  concentration indicates apatite  
1763 fractionation; therefore a stoichiometric  $Ca_5(PO_4)_3(OH)_2$  composition was also  
1764 included. For evolution within the trachytes, mineral compositions derived from the  
1765 UFG were used.

1766 Trace element modelling was undertaken using the Rayleigh fractionation law  
1767 and batch melting equations. Sample 98SM33 of Elliott et al. (2007) was applied to  
1768 represent mantle-derived parental basalt, or partially melted gabbroic crust. Trace  
1769 element Rayleigh fractionation models were performed in two steps, each using the  
1770 predicted fractionating assemblages from mass balance model steps 1 and 2,  
1771 respectively. Based upon the Zr content of the UFG rocks, the change in fractionating  
1772 mineral assemblage from step 1 to 2 was placed at 650 ppm Zr, which corresponds  
1773 to 60 % fractionation from basalt. Batch melting models were calculated assuming  
1774 the following mineral proportions: plagioclase (50 %), clinopyroxene (30 %), olivine  
1775 (15 %), Ti-magnetite (5 %). Bulk distribution coefficients for Rayleigh fractionation  
1776 models were calculated based on the fractionating assemblages predicted from the  
1777 two steps of mass balance modelling. Batch melting models were performed using  
1778 the same partition coefficients as those of the first Rayleigh fractionation step.  
1779 Partition coefficient values were taken from literature values found in the GERM KD  
1780 database (Nielsen 2006). For the batch melting models and step 1 of the Rayleigh  
1781 fractionation models, partition coefficients were selected primarily from alkali basalts,  
1782 whilst those of the second Rayleigh fractionation step were taken primarily from  
1783 peralkaline trachytes and rhyolites. Where a range of partition coefficients were  
1784 available, averages were applied. A partition coefficient for Ta was used for  $KD_{Nb}^{Biotite}$   
1785 and  $KD_{Nb}^{Ti-magnetite}$  due to lack of available literature data.

1786

1787 **Thermodynamic modelling**

1788 Magmatic differentiation processes were modelled thermodynamically using the  
1789 MELTS algorithms (Ghiorso and Sack 1995; Asimow and Ghiorso 1998; Gualda et  
1790 al. 2012), which use an experimentally derived database of solid-liquid phases at  
1791 specific temperature, pressure and  $fO_2$  to calibrate models of thermodynamic phase  
1792 stability. The Rhyolite-MELTS calibration (Gualda et al. 2012) was used to calculate  
1793 100 fractional crystallisation models over five starting water contents (0.5, 1.0, 1.5,  
1794 2.0, and 4 wt. %), four isobaric pressures (50, 150, 300, and 500 MPa) and redox  
1795 conditions ranging from FMQ +2 to FMQ -2. An additional 50 models were generated  
1796 for polybaric conditions, assuming a pressure change from 500 to 150 MPa at  
1797 varying points during magma evolution. Due to the capability of differentiation trends  
1798 in peralkaline magmas to decrease the  $SiO_2$  content of residual liquids (e.g. Scaillet  
1799 and Macdonald 2001), the MgO content and calculated peralkalinity index were used  
1800 as primary differentiation indices. Sample 98SM33 of Elliott et al. (2007) was used as  
1801 a starting melt composition.

1802

1803

1804

1805

1806

1807

1808

1809

1810

1811

1812

1813

1814

1815

1816

**Online Resource 5: Data quality**

1817

**(Excel spreadsheet)**

1818

1819

1820

1821

1822

1823

1824

1825

1826

1827

1828

1829



1830

1831

1832

1833

1834

1835

1836

1837

1838

1839

1840

1841

1842

1843

1844

**Online Resource 6: Whole rock and glass**

1845

**(Excel spreadsheet)**

1846

1847

1848

1849

1850

1851

1852

1853

1854

1855

1856

1857

1858

1859

1860

1861

1862

1863

1864

1865

1866

1867

1868

1869

1870

1871

1872

**Online Resource 7: Mineral chemistry**

1873

**(Excel spreadsheet)**

1874

1875

1876

1877

1878

1879

1880

1881

1882

1883

1884

1885

1886

1887

1888

1889

1890

1891

1892

1893

1894

1895

1896

1897

1898

1899

1900

**Online Resource 8: Thermobarometry data**

1901

**(Excel spreadsheet)**

1902

1903

1904

1905

1906

1907

1908

1909

1910

1911

1912

1913