- 1 Alkaline magmas in zones of continental convergence:
- 2 The Tezhsar volcano-intrusive ring complex, Armenia

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- 4 Krzysztof Sokół^{1,2}, Ralf Halama^{1*}, Khachatur Meliksetian³, Ivan
- 5 P. Savov⁴, Gevorg Navasardyan³, Masafumi Sudo⁵

6

- 7 1 School of Geography, Geology and the Environment, Keele University, Keele, ST5 5BG,
- 8 United Kingdom
- 9 ² (Present address) School of Earth and Environmental Sciences, University of St Andrews,
- 10 St Andrews, KY16 9AL, United Kingdom
- 11 ³ Institute of Geological Sciences, Armenian National Academy of Sciences, 24a Marshal
- 12 Baghramian Avenue, 0019, Yerevan, Republic of Armenia
- ⁴ School of Earth and Environment, University of Leeds, Leeds, LS2 9JT, United Kingdom
- 14 ⁵ Institute of Earth and Environmental Science, University of Potsdam, Karl-Liebknecht-
- 15 Str. 24-25, 14476 Potsdam, Germany

- * Corresponding author contact information:
- 18 Ralf Halama
- 19 School of Geography, Geology and the Environment
- 20 Keele University
- 21 Keele, ST5 5BG, United Kingdom
- 22 E-mail: r.halama@keele.ac.uk
- 23 Phone: +44 (0) 1782 7 34960

Abstract

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Alkaline igneous rocks are relatively rare in settings of tectonic convergence and little is known about their petrogenesis in these settings. This study aims to contribute to a better understanding of the formation of alkaline igneous rocks by an investigation of the Tezhsar volcano-intrusive alkaline ring complex (TAC) in the Armenian Lesser Caucasus, which is located between the converging Eurasian and Arabian plates. We present new petrological, geochemical and Sr-Nd isotope data for the TAC to constrain magma genesis and magma source characteristics. Moreover, we provide a new ⁴⁰Ar/³⁹Ar age of 41.0±0.5 Ma on amphibole from a nepheline syenite that is integrated into the regional context of ongoing regional convergence and widespread magmatism. The TAC is spatially concentric and measures ~10 km in diameter representing the relatively shallow plumbing system of a major stratovolcano juxtaposed by ring faulting with its extrusive products. The plutonic units comprise syenites and nepheline syenites, whereas the extrusive units are dominated by trachytic-phonolitic rocks. The characteristic feature of the TAC is the development of pseudomorphs after leucite in all types of the volcanic, subvolcanic and intrusive alkaline rocks. Whole-rock major element data show a metaluminous (Alkalinity Index = 0-0.1), alkalic and silica-undersaturated (Feldspathoid Silica-Saturation Index <0) character of the TAC. The general trace element enrichment and strong fractionation of REEs (Lan/Ybn up to 70) indicate a relatively enriched magma source and small degrees of partial melting. All TAC rocks show a negative Nb-Ta anomalies typical of subduction zone settings. The initial 87 Sr/ 86 Sr ratios (0.704-0.705) and positive ϵ Nd values (+3 to +5) indicate an isotopically depleted upper mantle and lack of significant crustal influence, which in turn suggests the TAC magma has formed via differentiation from lithospheric mantle melts.

Regionally, the age of \sim 41 Ma places the TAC amid a Lesser Caucasian Eocene period of dominantly calc-alkaline magmatism. The TAC's arc-like geochemical signatures are interpreted to result from prior subduction of the Tethyan slab beneath the Eurasian continental margin. The alkaline character, distinct from regional trends, is attributed to Neotethyan slab rollback causing extension and inducing small degrees of decompression melting of metasomatised lithospheric mantle.

- **Keywords:** Alkaline igneous rocks, ring complex, Armenia, geochemistry, ⁴⁰Ar/³⁹Ar
- dating, pseudoleucite

1. Introduction

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Studies of alkaline magmatism on the global scale have become a point of focus due to the significant role of alkaline magmatic rocks for ore exploration, in particular regarding prospecting for rare earth elements (REEs), niobium (Nb), tantalum (Ta) and zirconium (Zr) (e.g. Chakhmouradian and Zaitsev 2012). Many alkaline igneous rocks are found in rift-related intraplate settings (e.g. Gardar Province/Greenland, Upton et al. 2003; Kola Alkaline Province/Russia, Downes et al. 2005; East African Rift, Woolley 2001), but they also occur, albeit less frequently, in settings of plate convergence (Burke and Khan, 2006; Hou et al. 2006). Plate convergence includes collisional events that cause the welding of terranes into continental land and subsequent post-collisional episodes in which convergence continues (Bonin et al. 1998). The occurrence of magmas with alkaline affinities becomes more common only when the geodynamic context becomes entirely intraplate in a post-orogenic episode (Bonin et al. 1998). In complex collisional and postcollisional settings, the timing of specific types of magmatism depends on the geotectonic geometries and the relative rates of crustal thickening and subsidiary subduction (Harris et al. 1986). Importantly, convergent movement between colliding plates will continue for 30-50 Ma after the initial collision (Harris et al. 1986). On a global scale, deformed alkaline rocks and carbonatites (DARCs) may be used as indicators of where ancient oceans have opened and closed, and the presence of a variety of syenites, carbonatites and other alkaline igneous rocks found in proximity to older DARCs indicate the recycling of material from the underlying lithosphere based on the Wilson Cycle-type model (Burke and Khan 2006). Thus, investigating alkaline magmatism in convergent settings, e.g. in Tibet (Williams et al. 2004; Hou et al. 2006) and the Anatolian-Armenian-Iranian plateau (Jackson et al. 1995; Neill et al. 2015), has become as important as studies of rift-related settings to understand alkaline magma genesis.

The exact mechanisms responsible for magma generation in collisional tectonic
settings remain enigmatic. Models include slab break-off (Keskin 2003; van Hunen and
Allen 2011; Neill et al. 2015), large-scale delamination or thinning of the lithospheric
mantle (Innocenti et al. 1982; Pearce et al. 1990) and small-scale lithospheric detachment
driven by convection cells (Kaislaniemi et al. 2014; Neill et al. 2015). Moreover, the source
of magmas in compressional regimes and their chemical impact on the crust remains
disputed. Processes to generate primary magmas in collision zones may involve melting
of thickened lithosphere due to breakdown of hydrous phases at the continental suture
(Allen et al. 2013) and melting of deeply-subducted continental crust (Zhao et al. 2013).
To explain the alkaline character of the erupted or plutonic igneous rocks, several genetic
models and processes have been proposed:

- Low degrees of partial melting of metasomatized upper mantle (Bodeving et al. 2017; Dawson 1987; Marks et al. 2008).
 - 2. Melting of crustal sources, which could be located in the lower crust and mafic in composition (Smith et al. 1988) or in the middle to upper crust and felsic in composition (Downes 1987; Fitton 1987).
 - 3. Fractional crystallization from alkali basalt parental magmas (Delong et al. 1975; Trumbull et al. 2003), with variable degrees of crustal assimilation (Fitton 1987; Jung et al. 2007; Lan et al. 2011).
 - 4. Fenitisation a high temperature metasomatic alteration driven by alkali-rich fluids incrementally expelled from alkaline or carbonatitic melts (Sindern and Kramm 2000; Suikkanen and Rämö 2017).

Armenia, landlocked between the Black Sea and the Caspian Sea, forms part of the Anatolian-Armenian-Iranian Plateau and is characterised by widespread Cenozoic

volcano-magmatic activity, starting in the Eocene at ~50 Ma and intermittently lasting into the Holocene and historical times (Karakhanian et al. 2002; Moritz et al. 2016; Fig. 1a). Several studies focused on Quaternary volcanic cones on the Anatolian-Armenian-Iranian plateau (Innocenti et al. 1982; Pearce et al. 1990; Keskin et al. 1998), including in the Armenian segments of the Lesser Caucasus mountain range (Karapetian et al. 2001; Karakhanian et al. 2002), and the Miocene/Pliocene magmatic evolution of the region (Dilek et al. 2010; Neill et al. 2013; Kheirkhah et al. 2015). However, investigating the much less studied Paleogene igneous rocks is important to gain a more complete understanding of the long-term magmatic and geodynamic evolution in this setting of continuing convergence and to improve our understanding of collision-driven continental magmatism and mantle dynamics (Dilek et al. 2010; van Hunen and Allen 2011; Moritz et al. 2016).

In this study, we use a range of petrological and geochemical methods to describe and interpret the lithological variations of the Tezhsar volcano-intrusive alkaline ring complex (or Tezhsar Alkaline Complex - TAC) in Armenia. We provide a new ⁴⁰Ar/³⁹Ar age and expand on previous petrological and geochemical studies (Abovyan et al. 1981; Kogarko et al. 1995; Meliksetian 1971, 1989) with the aim to achieve a better understanding of TAC petrogenesis and to integrate that into a model of alkaline magma genesis within a setting of continuing plate convergence. We also highlight and discuss the occurrence of cm-sized pseudoleucites in the TAC.

2. Geological history

2.1 Regional tectonic setting

The TAC, located about 55 km north of Yerevan in the Lesser Caucasus, has formed in the Eocene in a setting of general convergence between the Eurasian and Arabian plates

(Fig. 1a). This region was affected by two distinct collisional events and the emplacement of the alkaline magmas of the TAC is crucial to the understanding of the tectono-magmatic evolution of the region.

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The TAC is located on basement of the South Armenian Block (SAB), which is a microplate of Gondwanaland origin (Knipper and Khain 1980; Rolland 2017; Sosson et al. 2010). Proterozoic metamorphic basement of the SAB is exposed in the Tsakhkunyats massif (Belov 1968; Aghamalyan 1998). Platform sedimentary cover of the SAB is presented by folded Late Devonian to the Late Triassic sedimentary formations (Arakelyan 1964; Aslanyan 1958). Ophiolites representing Jurassic oceanic crust were obducted onto the northern margin of the SAB in the Late Cretaceous (90-84 Ma; Rolland 2017). In the late Cretaceous to early Palaeogene (70-60 Ma), the SAB was welded to the southern margin of Eurasia as a result of the closure of the northern branch of the Neotethys and the termination of subduction (Rolland et al. 2009a, b; Moritz et al. 2016). The collision is marked by the Sevan-Akera suture zone, which is part of the regional northern Neotethys suture (Hässig et al. 2013; Sosson et al. 2010). The closure of the northern Neotethys branch caused a subduction jump towards the south and the accretion of the SAB to the Eurasian margin resulted in formation of a Cretaceous-Eocene flysch basin that overlies the ophiolites (Rolland 2017). At present, the Sevan-Akera suture separates two tectonostratigraphic units, the Southern and Northern Tethyan Provinces, which outline the continental provinces pre-dating the closure of the Tethys Ocean (Fig. 1b; Adamia et al. 2011). The Sevan-Akera suture is located ~6 km northward of the TAC. The second stage of accretion involving collision of the Arabian margin to the SAB and the Tauride-Anatolian block caused the closure of the South Neotethys ocean along the Bitlis-Zagros suture. This closure occurred in late Eocene to early Oligocene times (40-25 Ma) based on geochronological and structural evidence (Agard et al. 2005; Allen and Armstrong 2008; Rolland 2017). The convergence and collision between Arabia and Eurasia induced regional compression and shortening in the overriding (SAB-Eurasia) continental lithosphere (Agard et al. 2011), the formation of the Anatolian-Armenian-Iranian orogenic plateau (Sheth et al. 2015) and lateral ejection of the Anatolian and Iranian blocks, with the Armenian Highland (Lesser Caucasus and Eastern Anatolia) in the centre (Phillip et al. 1989). Protracted Cenozoic magmatism lasted from ~49 Ma to ~21 Ma and marked the final stages of the Neothethyan subduction, the main Arabia-Eurasia collisions and subsequent post-collisional events, including emplacement of the syn-collisional granite-leucogranite plutons of the Lesser Caucasus (Meliksetian 1989; Rezeau et al. 2017). To explain the Palaeogene magmatism of the entire region, Dilek et al. (2010) proposed the opening of an asthenospheric window beneath the arc mantle wedge and the collision zone. The presence of adakites of Early Eocene age in the Pontides interpreted as a result of slab window formation (Eyuboglu et al. 2011) supports this hypothesis. Lordkipanidze et al. (1989) and Sahakyan et al. (2016) consider a subductionmodified upper mantle source for Lower-Middle Eocene volcanism and an increase of

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Caucasus.

Considering the age and location of the TAC (40 Ar/ 39 Ar of 41.0±0.5 Ma, this study; 36.3-37.5 Ma, K-Ar, Baghdasaryan and Ghukasyan 1985; 36-39 Ma, K-Ar, Meliksetian 1989), it formed in a plate convergence setting, in between two major collisional events that occurred in region – first at the northern edge of the SAB in the Late Cretaceous to Early Paleogene, and subsequently to the south of the SAB in the Late Eocene to Early Oligocene. The TAC can thus be described as post-collisional relative to the initial collisional event between the SAB and the Eurasian plate.

crustal input within the Late Eocene-Early Oligocene magmatic series of the Lesser

2.2 Geological setting of the Tezhsar Alkaline Complex

The TAC is located on the Pambak ridge at the northern edge of the SAB within the Sevan-Shirak basin. To the south, the TAC is in contact with the Proterozoic metamorphic basement of the SAB across the Marmarik Fault. Presence of abundant xenoliths from the Tsakhkunyats basement, such as mica schists, confirms the affinity of the TAC to the SAB continental terrane. To the north, the TAC borders the Margahovit intrusion comprising porphyritic granosyenites. Country rocks exposed to the W-NW of the TAC comprise Upper Cretaceous clastic and carbonate strata and Mid-to-Late Eocene extrusive igneous rocks, which also outcrop to the E-SE. The Ulashik Fault cuts the TAC in SW-NE direction with horizontal left-lateral displacement of intrusive and volcanic units reaching 700 m.

The TAC represents a ring complex that can be subdivided into several concentric

The TAC represents a ring complex that can be subdivided into several concentric units of both volcanic and plutonic rocks. Such classical ring complexes are quite rare (Johnson et al., 1999) and are of special interest considering their structural and volcanological evolution as well as petrological aspects. According to Meliksetian (1971), the TAC includes the following major units (Fig. 2):

- 1. Outer cone sheets characterized by inward-dipping contacts
- 2. Ring unit of volcanic alkaline rocks with a thickness up to 600 m characterised by its concentric structure and inward-dipping contacts (Outer Volcanic Unit, OVU)
 - Central intrusive unit comprising syenites and nepheline syenites (Syenitic Unit, SYU)
 - 4. Ring dykes, circular bodies with sub-vertical contacts cutting both the volcanic and central intrusive units
- 5. Resurgent volcanic unit, inside the central intrusive unit, formed by volcanic breccias, dykes and subvolcanic rocks (Central Volcanic Unit, CVU).

For the purpose of the geochemical investigation in this study, we use a simplified subdivision into Outer Volcanic Unit, Syenitic Unit and Central Volcanic Unit (Fig. 2). Based on a structural analysis including bedding attitudes of units and relationships between volcanic ring, cone sheets and central pluton, the presence of circular dykes and remains of a volcanic centre, most researchers, namely Kotlyar (1958), Bagdasaryan (1966) and Meliksetian (1971) concluded that the TAC formed via a caldera collapse and the volcanic ring was emplaced through collapse along concentric faults. The exceptionally large elliptical palaeocaldera structure of the TAC is $\sim 13.6 \times 11.5$ km in size and has an area of ~ 131 km², comparable in dimensions to the Santorini caldera in the Aegean Sea. Such a ring morphology provides a unique insight into the roots of an alkaline volcano-plutonic complex.

Beyond the petrogenetic and structural significance of the TAC, there is also a characteristic widespread development of pseudomorphs after leucite, which have been studied in detail by B. Meliksetian (1970, 1971, 1979, 1989) and Yagi and Gupta (1978). They feature in volcanic, subvolcanic and intrusive alkaline rocks and the largest crystals, reaching up to 8 cm in size (Fig. 3) are found in porphyry tinguaite dykes (Meliksetian 1978; Yagi and Gupta 1978). Their crystallographic habit is either icositetrahedral (in volcanic rocks and dykes) or triakis octahedral (in intrusive syenites). In the Soviet petrological literature according to Zavaricky (1934), pseudomorphism after leucite is divided into two mineralogical and genetic types: "Pseudoleucites" referring to leucite breakdown into nepheline and orthoclase, and "epileucites" describing pseudomorphism after leucite composed of agglomerated orthoclase, muscovite, analcime, chlorite, calcite and zeolites. In the Western petrological literature, usually both types are referred to as pseudoleucites, and both types have been described in the TAC.

3. Field observations

Field campaigns in the TAC were carried out in 2008, 2012 and 2015 in order to achieve two major aims: i) Help the completion of geological map (incl. GIS database) of the complex; ii) sampling the various lithologies of the TAC for petrological and geochemical investigations (Fig. 2). Sampling was focused on the three major units generalised for the purposes of this study: The Outer Volcanic Unit (OVU), the inner Syenitic Unit (SYU) and the Central Volcanic Unit (CVU) (Fig. 2), which have been juxtaposed by ring faulting. In total, 46 samples were collected and analysed, and one of those (sample 6-8-12 from the SYU) was used for 40 Ar/ 39 Ar age determination. Field relations demonstrate that the syenitic magmas of the SYU intruded into the OVU (Fig. 3a). More localized and subordinate lithologies of the complex include syenitic pegmatites (Fig. 3b) and pseudoleucite-bearing phonolites (Fig. 3c-f).

4. Petrography

The pioneering works of Meliksetian (1989) identified >50 different mineral species in rocks of the TAC, including a variety of rare earth element (REE) and high field strength element (HFSE) bearing phases. In our study, we focus on the major rock-forming minerals in the three major rock units of the complex to provide a general overview of the lithologies.

The volcanic rocks of the Outer Volcanic Unit (OVU) are typically porphyritic with an aphanitic groundmass. Major minerals are plagioclase + clinopyroxene + amphibole + biotite + alkali feldspar + Fe-Ti oxides ± nepheline, and apatite and titanite are present as accessory phases. Plagioclase is euhedral to subhedral, weakly zoned and often shows sieve textures (Fig. 4a). Euhedral clinopyroxene phenocrysts are up to 2 mm in size and typically poikilitic. Volcanic breccias are observed occasionally, containing angular

fragments and xenoliths, the latter partly rich in quartz. Volumetrically small occurrences of altered pseudoleucite phonolites are present, where we found pseudomorphed leucite up to several cm in diameter. The deltoidal icositetrahedral crystal habit of the primary leucite is well preserved, but leucite has been completely replaced by secondary minerals. These are dominated by alkali feldspar and cancrinite-group minerals and comprise minor amounts of analcime. Other phases found in the pseudoleucite are clinopyroxene, biotite, apatite and calcite.

The volcanic rocks of the Central Volcanic Unit (CVU) are generally porphyritic with a fine-grained matrix. They contain euhedral plagioclase + alkali feldspar + clinopyroxene + amphibole + biotite + Fe-Ti oxides as major mineral phases. Some samples contain amphibole glomerocrysts and clinopyroxene overgrowing biotite (Fig. 4b). Rare pseudoleucite phonolites occur in this unit as well. The samples of the CVU are often intensely altered.

The Syenitic Unit (SYU) comprises equigranular, phaneritic, medium to coarse-grained syenites and nepheline syenites (Fig. 4c-h). Several samples show a trachytoidal preferential alignment of feldspars. Major mineral phases are alkali feldspar + amphibole + biotite + clinopyroxene + Fe-Ti oxides ± nepheline ± plagioclase. Garnet is rare but very prominent in the coarse grained (pegmatitic) rock varieties, where euhedral to subhedral brown garnet forms clusters with euhedral, black todark green amphibole. Accessory phases observed include zircon, titanite, fluorite, muscovite, apatite, calcite, sodalite and cancrinite. Subhedral alkali feldspar is typically the most abundant phase, frequently exhibiting significant alteration. Primary clinopyroxene commonly shows signs of incipient alteration to green amphibole.

5. Analytical methods

Major and trace elements were analysed by standard X-ray fluorescence (XRF), inductively coupled plasma atomic emission spectrometry (ICP-AES) and inductively coupled plasma mass spectrometry (ICP-MS) methods. Detailed information about the analytical methods used is provided in the supplementary material. Systematic differences between analyses from different laboratories are not observed. If they exist, they are likely to be small relative to the compositional effects of the magmatic processes operating, and considered negligible for the overall interpretation of the dataset.

Strontium (Sr) and neodymium (Nd) isotope analyses were performed at the School of Earth and Environment, University of Leeds. Conventional ion-exchange chromatographic techniques were applied and samples were analyzed on a Thermo Finnigan Triton multicollector mass spectrometer (see Halama et al. 2013 for details of the analytical protocol). Information about reference materials analysed as well as normalization and correction procedures applied is given in the supplementary material.

 40 Ar/ 39 Ar dating of amphibole from syenite sample 6-8-12 was performed using a CO₂ laser stepwise heating technique at the Institute of Earth and Environmental Science, Universität Potsdam. The analytical protocol follows established procedures and a brief summary about procedural aspects, standards used and corrections applied is provided in the supplementary material. Calculation of ages and errors was performed following Uto et al. (1997) using the total 40 K decay constant of 5.543 x $^{10^{-10}}$ a⁻¹.

6. Results

6. 1. Rock classification and major element geochemistry

The Total Alkali versus Silica (TAS) diagram was used to classify the volcanic rocks from the OVU and CVU (Fig. 5a). For the intrusive rocks of the SYU, we used the classification diagram of De La Roche et al. (1980; Fig. 5b). Whole rock geochemical

analyses are presented in Table 1. All volcanic rocks of the TAC are classified as alkaline in the TAS diagram (Fig. 5a). Rocks of the OVU cover a wide compositional range from basaltic trachyandesite to phonotephrite, tephriphonolite and phonolite. The compositional range of the CVU rocks is more restricted, comprising trachyandesites and trachytes. The plutonic rocks of the SYU are classified as nepheline syenites and syenites based on the R1 and R2 parameters (Fig. 5b), which generally agrees with the petrographic observations.

A further geochemical classification was carried out using various geochemical indices (Table 1) that allow an evaluation of petrogenetic relationships (Shand 1947; Frost et al. 2001; Frost and Frost 2008). The majority of the Tezhsar rocks are ferroan, alkalic, metaluminous and silica-undersaturated. The Alkalinity Index (AI; AI = Al-(K+Na) on a molecular basis) typically varies between 0 and 0.1, indicating that peralkaline rocks (AI<0) are largely absent at the TAC. Values for the feldspathoid silica-saturation index (FSSI; normative Q-[Lc+2(Ne+Kp)]/100, where Q = quartz, Lc = leucite, Ne = nepheline and Kp = Kaliophilite) mostly range from -0.6 to 0. The negative FSSI values demonstrate that the rocks are generally silica-undersaturated. Diagrams using the aluminium-saturation index (ASI; molecular Al/(Ca-1.67P+Na+K) and the modified alkali-lime index (MALI; Na₂O+K₂O-CaO) classification demonstrate the predominantly metaluminous and alkalic nature of the TAC rocks (Fig. 5c, d). Peraluminous compositions (ASI>1) are very rare. Compared to the restricted compositions of SYU and CVU, the OVU shows the largest variations in A/NK ratios.

Harker diagrams show a relatively smooth decrease of MgO, total FeO (FeO $_{\rm T}$) and CaO with increasing SiO $_{\rm 2}$ contents (Fig. 6a-c). MgO contents are below 3 wt% for OVU rocks and SYU and CVU rocks have less than 1wt% MgO, demonstrating their highly evolved

character and suggesting substantial fractionation of mafic minerals prior to crystallisation.

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6.2. Trace element geochemistry

Whole-rock trace element concentrations in the TAC are variable and show some significant enrichment in Sr (up to \sim 5000 ppm), Ba (up to \sim 4000 ppm), Zr (up to \sim 1000 ppm) and ΣREE (up to ~1200 ppm), which is typical for alkaline igneous rocks (Chakhmouradian and Zaitsev 2012). Incompatible trace elements such as Th and Zr show pronounced enrichment with increasing silica, in particular evident for SYU and CVU rocks (Fig. 6d, e). In contrast, Sr contents remain relatively constant for intermediate rocks with <58 wt% SiO₂ and diminishing at higher silica contents (Fig. 6f). A chondritenormalised REE diagram (Fig. 7a) shows that both the volcanic and plutonic rocks of the TAC are characterised by a strong fractionation between LREE and HREE with $La_{(N)}/Yb_{(N)}$ ratios predominantly around 10-40 but reaching values as high as 70. Absolute amounts of LREE are generally higher in the SYU (~200-1000 x chondrite) compared to the OVU and CVU (~40-500 x chondrite). Europium anomalies, defined as Eu/Eu*= $\frac{Eu_N}{\sqrt{(Sm_N \times Gd_N)}}$, are moderately negative in the volcanic units OVU (0.80 - 1.08) and CVU (0.68 - 0.91). The majority of the SYU rocks have more pronounced negative Eu anomalies with Eu/Eu* values between 0.44 and 0.97 (Fig. 7a). On primitive mantle-normalised trace element diagrams (Fig. 7b-d), negative anomalies for Nb, Ta and Ti are the most prominent features in all three units. In contrast, a strong relative enrichment of Th and U compared to Rb and Ba is only significant in the SYU and CVU, but not discernible in the OVU.

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6.3 Sr and Nd isotopes

Initial Sr and Nd isotope ratios of volcanic and plutonic rocks from the TAC,

recalculated to an age of 41 Ma, range from 0.7040 to 0.7052 and 0.51274 to 0.51283, respectively (Table 1). 19 of 20 samples fall within the range 0.7040 to 0.7044 for the initial 87 Sr/ 86 Sr ratio. The Nd isotopic compositions correspond to positive ϵ Nd values between +3.0 and +4.8 (Table 1).

6.4 ⁴⁰Ar/³⁹Ar geochronology

One syenite sample (sample number 6-8-12) was dated by 40 Ar/ 39 Ar step heating. The total gas age is 42.1 ± 0.5 Ma (Fig. 8; Table 2). We use the following criteria outlined by Fleck et al. (1977) for defining a plateau age: (1) The plateau includes at least 50% of the total 39 Ar released, (2) the ages of two contiguous steps in the plateau agree within 2s error, excluding the J value error, (3) the plateau consists of three steps or more, and (4) each degassing step contributing to the plateau contains >3% of the total 39 Ar released. For the syenite sample 6-8-12, five plateau steps constituting 98.88% of the total 39 Ar released can thus be used to define a plateau age of 41.0 ± 0.5 Ma (Fig. 8; Table 2). Using the plateau steps only, a normal isochron age of 41.3 ± 2.5 Ma with a 40 Ar/ 36 Ar intercept at 281 ± 58 is obtained. The corresponding inverse isochron yields an age of 41.2 ± 2.1 Ma with $(^{40}$ Ar/ 36 Ar)_i = 289 ± 57. The good agreement between the three ages underlines the reliability of the age determination, with the plateau age of 41.0 ± 0.5 Ma representing the most precise and hence preferred age.

7. Discussion

7.1 Comparison with regional magmatic signatures

The alkaline and highly evolved nature of the TAC rocks makes them distinct from volcanic rocks outcropping in Armenia, which are typically transitional between alkaline and subalkaline. This includes the trachybasaltic to trachyandesitic Pliocene-Quaternary

rocks from northern Armenia (Neill et al. 2013, 2015), as well as rocks from the large polygenetic Aragats volcano (Connor et al., 2011) and from the Gegham, Vardenis and Syunik Volcanic Highlands in South Armenia (Karapetian et al. 2001; Sugden et al. submitted). A comparison with data for regionally related Miocene to Quaternary Armenian igneous rocks from the Yerevan and Shirak regions (Neill et al. 2015) reveals a general enrichment of the TAC rocks in almost all moderately to highly incompatible trace elements (Fig. 7b-d). Key features, such as negative Nb-Ta and Ti anomalies and a relative enrichment of LREE compared to HREE, are similar. Isotopically, the TAC rocks, which plot on the Sr-Nd mantle array, overlap with plutonic rocks from the Meghri-Ordubad pluton and with other Miocene to Quaternary volcanic rocks from Armenia (Fig. 9). Quaternary volcanic rocks from Aragats (Lebedev et al. 2007; Connor et al. 2011) and the Gegham Ridge (Lebedev et al. 2013) also overlap in their Sr-Nd isotopic compositions. This comparison reveals that there is a broad Sr-Nd isotopic homogeneity across a large area of the Armenian highlands from the Eocene to the Quaternary, indicating that similar source regions are involved in magma genesis. Broadly contemporary (47-40 Ma) postcollisional magmatic rocks from the Eastern Pontides (NE Turkey), which are characterized by tholeiitic/calc-alkaline affinities enriched in LILE with pronounced depletions in HFSE, also overlap in their isotopic composition (Aydınçakır & Şen 2013). In contrast, extending the comparison to Eocene magmatic rocks in NW Iran reveals that post-collisional granites and syenites from the Sanandaj-Sirjan Zone and granitoids from the Urumieh-Dokhtar magmatic arc extend to significantly more radiogenic Sr-Nd isotope compositions (Fig. 9).

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7.2. Magma differentiation and magma source geochemistry

Both volcanic and plutonic rocks of the TAC are evolved and only a few samples are

of intermediate composition (Fig. 5a, b). The influence of mixing, fractional crystallization and batch partial melting on the bulk geochemical composition of the rocks can be evaluated using incompatible trace elements with different bulk solid/liquid partition coefficients (Schiano et al. 2010). In the Rb/Nd vs. Rb diagram (Fig. 10a), the nearhorizontal trend for the majority of data points emphasizes the dominant role of fractional crystallization, whereas mixing and differences in batch partial melting would yield positive correlations (Schiano et al. 2010). This interpretation is supported by a curved overall trend in the Rb vs. Rb/V diagram (Fig. 10b), which is consistent with fractional crystallisation or mixing, but not with different degrees of partial melting (Schiano et al. 2010). Moreover, the coherent trend of the Rb/Ba vs. Ba diagram (Fig. 10c) reflects feldspar fractionation and does not indicate any significant effects of hydrothermal alteration. A major role of role of crustal contamination processes can also be excluded based on the unradiogenic initial 87Sr/86Sr isotope ratios that remain relatively constant with increasing silica (Fig. 10d). Crustal contamination typically leads to an coupled increase in (87Sr/86Sr)_i and SiO₂, which is not observed for the TAC. The only sample with an elevated (87Sr/86Sr)_i ratio (2-7-09) has a high Rb/Sr ratio of ~8 and might be affected by a larger uncertainty in recalculation of the initial value and/or post-magmatic Rb or Sr mobilization. There is also no indication of limestone assimilation, which would lead to a significant enrichment in CaO (Fig. 6c). Olivine is conspicuously absent in all TAC rocks, but the low MgO contents and the highly evolved character point to preceding fractionation of mafic minerals (Fig. 6). The decrease in CaO/Al₂O₃ coupled with increasing FeO_t/MgO ratios, Eu anomalies that become more negative with increasing degree of differentiation, and decreasing Sr

contents and Dy/Yb ratios with increasing SiO2 suggest a significant role of

amphibole/clinopyroxene and plagioclase fractionation whereas garnet fractionation

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was insignificant (Fig. 11a-c). Typically, the OVU rocks are more primitive than both CVU and SYU rocks. OVU rocks even retain Eu/Eu* values close to 1, pointing to lack of significant plagioclase fractionation (Fig. 11b).

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The pronounced depletions in HFSE (Nb, Ta, and Ti) in all TAC rocks emphasizes the influence of subduction processes on the mantle source (Fig. 7). Similar negative HFSE anomalies have been observed in alkaline rocks of the Longbaoshan Complex, North China Craton (Lan et al. 2011) and carbonatites from east Tibet in the Himalayan collision zone (Hou et al. 2006) and were attributed to subduction processes influencing the magma source regions prior to continental collision. In addition, various trace element indicators for source enrichment processes support the notion that the OVU and the CVU are geochemically distinct (Fig. 11d-f). The OVU shows elevated Sm/Yb and Ba/La ratios, as well as relatively low La/Sm and Th/Yb ratios compared to the CVU (Fig. 11d-f). Collectively, these geochemical features of the OVU are interpreted as a signature of moderate fluid enrichment via slab dehydration inherited from earlier subduction events. Both CVU and OVU rocks share high Ba/Nb ratios, similar to arc volcanic rocks in general (Fig. 12a). There is little overlap between the two groups as OVU rocks are additionally characterized by, on average, higher La/Nb ratios and Ba/Nb ratios >100, suggesting a temporal evolution towards a decreasing subduction influence from the early OVU to the late stage CVU. The more scattered trend towards lower Ba/Nb ratios in the syenites and nepheline syenites is likely a result of progressive alkali feldspar fractionation and should not be considered as the parental magma signature. The felsic plutonic rocks from the SYU tend to exhibit a larger geochemical variability when compared with the volcanic rocks, which is likely related to the fact that some show cumulate textures and may not represent melt compositions. The CVU, in contrast, has compositions that are more tightly clustered, with a faint indication of source enrichment from subducted sediments. Mechanisms of enrichment of the mantle source can be distinguished using $[Hf/Sm]_N$ and $[Ta/La]_N$ ratios (Fig. 12b), where TAC rocks are characterized by a subduction metasomatism signature, clearly distinct from carbonatitic metasomatism.

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Sr-Nd isotopic compositions are broadly overlapping with Eocene to Pliocene magmas from the Meghri-Ordubad pluton and Pliocene to Quaternary volcanism in central and northern Armenia, pointing to only minor spatial variations in the respective mantle source regions (Fig. 10). The source of the TAC magmas is dominated by a depleted mantle component and crustal contamination is essentially absent, as all of the possible crustal contaminants would greatly enhance the radiogenic isotope signatures of the magmas, which is not the case. Silica-undersaturated alkaline rocks commonly have isotopic compositions that suggest a magma source in the mantle (Dunworth and Bell 2001; Kramm and Kogarko 1994). For instance, nepheline syenites from the Gardar Province (Greenland) show Nd isotopic compositions typical for mantle-derived rocks without any significant crustal assimilation (Halama et al. 2005; Marks et al. 2004). Therefore, evolved silica-undersaturated rocks are interpreted as products of differentiation from more primitive nephelinitic, basanitic or alkali basaltic magmas derived from the upper mantle (Kramm and Kogarko 1994; Trumbull et al. 2003). Basanitic volcanism is common to the south of TAC in the Syunik Volcanic Highland (Sugden et al. submitted) near the Armenia-Azerbaijan-Iran border region.

The trace element evidence for a subduction modifications and the Sr-Nd isotopic evidence for previous melt extraction suggest that the TAC magmas are predominantly derived by low degrees of partial melting from a lithospheric mantle source which has been affected by pre-Eocene subduction i.e., prior to post-collisional melt generation. This magma generation model is also the preferred model for volcanism in East Anatolia (Keskin 2003), and similar geochemical features in volcanic rocks from the Eastern

Pontides (Artvin Province) contemporary (47-40 Ma) to emplacement of the TAC were also interpreted to be derived from a mantle source that had experienced metasomatism by slab-derived fluids (Aydınçakır & Şen 2013). Post-collisional magmatic processes are commonly affected by prior subduction processes and LILE-enriched mantle sources are characteristic for these rocks (Bonin et al. 1998), typically resulting in calc-alkaline magmatic suites (Harris et al. 1986). The TAC represents an unusual case insofar as the post-collisional magmatic rocks are alkaline in character but also derive from a subduction-modified mantle source.

7.3. The age of the Tezhsar Alkaline Complex in a regional context

The mid-Eocene age of 41.0 ± 0.5 Ma falls into a time of widespread magmatism in the Lesser Caucasus region, which lasted from ~ 49 to ~ 38 Ma and comprised the emplacement of alkaline and nepheline-bearing gabbros, monzonites and syenites as well as gabbro-diorite-granodiorite-syenogranite complexes and granites (Ghukasyan et al. 2006; Melkonyan et al. 2008; Moritz et al. 2016). The magmatic activity was accompanied by porphyry-type Cu-Mo mineralization that was dated at 44-40 Ma by Re-Os analyses of molybdenite (Moritz et al. 2016). Slightly younger alkaline magmatism is represented by the Bunduk alkaline complex (38-32 Ma) located ~ 15 km northeast of the TAC (Abovyan et al., 1981; Meliksetian, 1989). This pluton intrudes the Middle-Late Eocene volcanic suite of the Bazum ridge and the Bazum gabbro-granitoid intrusive complex, exhibiting an elongate morphology, parallel to the segment of Pambak-Sevan fault.

Regionally, broadly contemporaneous magmatic activity is also recorded in the Talysh mountain range (Azerbaijan/Iran) at around 41-38 Ma (Vincent et al. 2005), in the Eastern Pontides (Turkey) at ~46-40 Ma (Aydınçakır and Şen 2013) and in western Georgia at ~47-41 Ma (Lebedev et al. 2009). Further to the SE in the Zagros orogen, ~41

Ma old granites and syenites occur in the Piranshahr massif (Mazhari et al. 2009) and ~40 Ma granitoids in the Urumieh-Dokhtar arc (Kazemi et al. 2018). The peak of subduction-related magmatism in Iran is also close to 40 Ma (Allen and Armstrong 2008), and a magmatic flare-up lasting ~18 million years from 55 to 37 Ma has been postulated in the Urumieh-Dokhtar belt and the Alborz Mountains in Iran (Verdel et al. 2011). Throughout the Eocene, the plate convergence between the Arabian and Eurasian plates was proceeding at rates of 2-3 cm/year (McQuarrie et al. 2003). Following the initiation of the Arabia-Eurasia collision, arc magmatism declined in the Late Eocene (Allen and Armstrong 2008). However, convergence was relatively rapid throughout Eocene-Oligocene time, and only slowed since Early Miocene (Rosenbaum et al. 2002).

The age of the TAC falls within this period of extensive magmatism during convergence between the Arabian and Eurasian plates, and its geochemical characteristics demonstrate a subduction-related origin. This subduction signature is inherited from prior northward subduction of the Neotethys ocean underneath the Eurasian margin, leading to a preconditioning of the mantle (Verdel et al. 2011). Typical calc-alkaline, subduction-related Eocene magmatism typical for active arc environments is preserved in the oldest granitoids (49-44 Ma) of the Meghri-Ordubad pluton (Moritz et al. 2016). The Lesser Caucasus experienced extension and crustal thinning at around 40 Ma causing decompression melting of the hydrated, subduction-influenced lithospheric mantle (Verdel et al. 2011), which imparted its geochemical signature onto the TAC magmas. Middle Eocene (ca. 49–40 Ma) extension, accompanied by magmatism, also occurred in Iran (Ballato et al. 2011). The extension-related magmatism in an overall setting of convergence (Rosenbaum et al. 2002) is caused by the rollback of the Neotethys slab (Vincent et al. 2005; Verdel et al. 2011).

The oldest rocks at TAC in the OVU show some geochemical characteristics reminiscent of a dehydration fluid signature in arc magmatic rocks (high Ba/Nb, Ba/La ratios; Figs. 7 and 11e, f). A clear arc signature, most evident in the pronounced negative Nb-Ta anomalies, is present in all of the TAC rocks, similar to the Meghri-Ordubad pluton at the Armenia-Iran border. However, the TAC rock compositions are distinct as they are not calc-alkaline but alkaline (Fig. 5) with a pronounced enrichment in incompatible trace elements (e.g. up to 5000 ppm Sr and typically 100-500 ppm Rb compared to <1000 ppm Sr and 10-200 ppm Rb in rocks from Meghri). This geochemical character is not due to long-lived differences in the mantle source compared to Meghri-Ordubad pluton since the Sr-Nd isotopic characteristics are similar (Fig. 9). Instead, smaller degrees of melting and/or a metasomatic enrichment episode(s) immediately prior to magma generation have to be invoked. The very pronounced subduction signature in the TAC supports the predominant melting of hydrated and HFSE-depleted lithospheric mantle, with subordinate contributions from upwelling astenospheric mantle (Verdel et al. 2011). The occurrence of these alkaline rocks in a general setting of convergence is unusual, but can be attributed to periods of localized extension in the Lesser Caucasus. The overall convergence throughout Eocene and Oligocene is well established based on kinematic data and modelling (Rosenbaum et al. 2002), but if the lithospheric structures allowed ascent of mantle-derived magmas via localized faulting and/or rift tectonics alkaline magmatism can develop even in collision zones (Harris et al. 1986). Development of an extensional regime along this sector of Lesser Caucasus was previously suggested to explain the alkaline character of Paleogene magmatic rocks, particularly those within Armenia (Kogarko et al., 1995).

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7.4. Petrogenesis of pseudoleucite phonolites

Based on optical microscopy and geochemical analyses including XRD, 6 types of "epileucites" and 5 types of pseudoleucites were distinguished by their mineral associations, host rocks and crystallographic habit (Meliksetian 1971, 1978). "Epileucites" are considered to be a result of post-magmatic hydrothermal alterations, whereas pseudoleucites are considered to be a result of disintegration of metastable K-Na leucite into mixture of orthoclase and nepheline under subsolidus conditions (T=600°C) in late magmatic stage (Meliksetian 1978; Gittins et al. 1980). Yagi and Gupta (1978) mention that the K₂O/Na₂O ratio of 4.3 in pseudoleucites of porphyry tinguaite dykes of TAC is the highest among those studied worldwide highlighting the importance of resolving complex's evolutionary story to better understand the conditions of pseudoleucite paragenesis.

The investigated leucite pseudomorphs occur in a phonolite (Fig. 3c-f). Relicts of primary leucite are lacking, and they are generally rarely observed in leucite pseudomorphs. The leucite pseudomorphs mainly consist of alkali feldspar but do not contain nepheline, instead comprising abundant cancrinite (Fig. 13b-c). Different theories about the genesis of leucite pseudomorphs were put forward (see Edgar, 1984, and references therein), including (1) subsolidus breakdown of leucite to orthoclase and nepheline, (2) reaction of leucite with a Na-rich liquid and (3) alkali ion exchange reactions between leucites and Na-rich glass or fluid. We will briefly discuss these theories in relation to the leucite pseudomorphs in the phonolite.

Subsolidus breakdown of common K-rich leucite would produce alkali feldspar and kalsilite, hence a process to cause relative enrichment of Na is required to explain the occurrence of Na-bearing phases in pseudoleucites. Leucite solid solutions with up to 40 wt.% NaAlSi₂O₆ were produced experimentally, and these experienced subsequent breakdown into nepheline and alkali feldspar (Fudali 1963). However, natural leucite

does not contain excess amount of sodium to form this type of intergrowth on decomposition (Viladkar 2010). The mineralogy of the leucite pseudomorph, comprising abundant Na-bearing phases such as cancrinite and analcime (Fig. 13b-c), suggest that they are derived from a Na-rich precursor phase. Hence, subsolidus breakdown of natural K-rich leucite alone cannot explain their occurrence, but formation of a metastable Na-rich leucite before breakdown might be possible (Taylor and MacKenzie 1975).

The pseudoleucite reaction is a reaction of leucite with a Na-rich magma to form alkali feldspar and nepheline in the system NaAlSiO₄ – KAlSiO₄ – SiO₂ (Bowen and Ellestad 1937; Edgar 1984). This reaction terminates the leucite stability field and leucite disappears by reaction with the magma (Bowen and Ellestad 1937). The TAC leucite pseudomorphs, however, are characterized by a well-preserved deltoidal icositetrahedral crystal habit, reflecting the external shape of the precursor phase. It is difficult to envisage this reaction to fully replace primary leucite without modifying the morphology of the leucites (Taylor and MacKenzie 1975), which is so beautifully preserved (Fig. 3). Moreover, various minor mineral phases that contain additional elements occur within the pseudomorphs. Some of these (e.g. clinopyroxene, apatite) may be explained as primary magmatic inclusions, but others (analcime, calcite) texturally appear as secondary phases (Fig. 13b-c). This suggests that explaining the genesis of the leucite pseudomorphs based on the phase relations in this petrogenetic system is an oversimplification (Edgar 1984).

Alkali ion exchange reactions between leucites and Na-rich glass or fluid was proposed as mechanism to produce pseudomorphs after leucite that are similar in composition to natural pseudoleucites (Taylor and MacKenzie 1975). Fluid-induced reactions would facilitate the increase in Na content and formation of Na-dominated phases, such as cancrinite and analcime in TAC. Cancrinite is assumed to replace nepheline due to a reaction between nepheline and volatile-rich melts or fluids, a common

late magmatic- hydrothermal process (Martins et al. 2017). A reaction with fluids was also used to explain pseudoleucite with intergrowth of alkali feldspar, sericite and cancrinite from the Gardar Province, Greenland (Hesselbo 1986) and the replacement of nepheline by analcime, cancrinite, sodalite and muscovite in pseudoleucite from India (Viladkar 2010). Cancrinite is also an important constituent of the pseudoleucite phenocrysts from Spotted Fawn Creek (Yukon Territory, Canada), where also garnet, biotite, calcite, muscovite and plagioclase occur as inclusions within pseudoleucite (Tempelman-Kluit 1969). Removal of K, addition of Na and water was attributed to the entry of a fluid phase to permit the chemical exchange (Tempelman-Kluit 1969). The presence of cancrinite in the TAC leucite pseudomorphs bears evidence for interaction with a H₂O-CO₂-bearing fluid, possibly with minor amounts of S and Cl, as the general formula for cancrinite is $(Na,Ca,K)_{6-8}Al_{6-x}Si_{6+x}O_{24}(CO_3,SO_4,Cl,OH)_{1-2}\cdot nH_2O$ with x << 1 and n = 1-5 (Martins et al. 2017) illustrates. Given the scarcity of analcime in the TAC pseudoleucites, a conversion of primary leucites into analcime via reaction with Na-rich fluids as proposed for pseudoleucites from a phonolite dyke in Bohemia (Pivec et al. 2004) seems unlikely. The texture of the TAC leucite pseudomorphs pseudoleucites has resemblance to a "palisade texture", in which orthoclase laths near the margins of the pseudomorphs are oriented at right angles to the crystal boundaries (Tempelman-Kluit 1969). These textures can be interpreted to form by subsolidus reactions in response to increasing fluid pressure when pervasive fluids come in contact with the leucite (Hesselbo 1986). All these lines of evidence point to a late/post-magmatic hydrothermal alteration for the formation of the leucite pseudomorphs in the investigated phonolite, and they can be referred to as "epileucites". Complementary evidence for fluid-rich conditions during the late to post-magmatic evolution of the TAC are the presence of pegmatites and the widespread alteration in the CVU rocks.

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8. Conclusions

- A combination of small degrees of partial melting and pre-conditioning of the mantle source by slab dehydration and subsequent metasomatic processes can explain the alkaline, subduction-influenced geochemical character of the TAC.
- The Sr-Nd isotopic data demonstrate a mantle source with negligible crustal influence. There is a broad isotopic overlap with Eocene to Quaternary magmatism in other regions of Armenia, suggesting the regional presence of isotopically similar mantle source regions.
- The emplacement of the syenitic units of the TAC was dated by ⁴⁰Ar/³⁹Ar at 41.0 ± 0.5 Ma. The emplacement of the TAC can thus be linked to a previously proposed model of Eocene Neotethyan slab rollback driving decompression melting and extension-related magmatism in Iran and Azerbaijan within a tectonic setting of general convergence between the Arabian and Eurasian plates.
- The formation of leucite pseudomorphs is related to initial leucite crystallization from an evolved, silica-undersaturated magma followed by subsolidus breakdown and interaction with a late to post-magmatic fluid. The magmatic-hydrothermal fluid percolating through the rocks caused alteration of nepheline into cancrinite and amphibolitisation of clinopyroxenes. This fluid overprint may be responsible for the plethora of REE-bearing phases described previously within the TAC and hence be a crucial factor in the (re)distribution of rare elements in alkaline igneous rocks.

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Figure captions

Figure 1 – (a) Geotectonic framework of the Caucasus region showing major tectonostratigraphic provinces, associated terranes and the location of Tezhsar Alkaline Complex (star) about 50 km north of Yerevan (modified after Adamia et al., 2011, and Rezeau et al., 2017). (b) Palaeogeographical reconstruction of the Eurasian-Arabian collision in the Ypresian (52Ma) just before formation of TAC (modified after Mederer et al., 2013). SAB – South Armenian Block, SAS – Sevan-Akera Suture, BZS – Bitlis-Zagros Suture, TAB – Tauride-Anatolian Block, NTP – Northern Tethyan Province, STP – Southern Tethyan Province.

Figure 2 - Geological map of the Tezhsar Alkaline Complex. The inset show a simplified
 subdivision of the TAC which is used for the geochemical diagrams of this study.

Figure 3 – Field relations (a, b) and hand specimen photographs (c, d) of the TAC. (a) Light coloured syenite intruding into dark grey volcanic rocks of the Outer Volcanic Unit. (b) Coarse-grained nepheline syenite pegmatite comprising dark patches of garnet and amphibole. (c) Phonolite handspecimen with idiomorphic leucite pseudomorphs reaching up to 2 cm in diameter. (d) Polished surface of a pseudoleucite phonolite. (e, f) Hand specimen of pseudoleucite megacrysts (up to ~8cm in diameter) as deltoidal icositetrahedra found in TAC phonolites. Samples are from old collections of B. Meliksetian and Z. Chibukhcyan.

Figure 4 – Photomicrographs illustrating characteristic features of rocks from the TAC in 1185 plane polarized (PPL) and cross-polarized (XPL) light. (a) Plagioclase (Pl) phenocrysts in feldspathic matrix of a basaltic trachyandesite, OVU (XPL). (b) Biotite (Bt) surrounded by clinopyroxene (Cpx) with accessory apatite (Ap) in trachyte, CVU (PPL). (c) Clinopyroxene and titanite (Ttn) in nepheline syenite, SYU (PPL). (d) Amphibole (Amp) in syenite, SYU (PPL). (e) Sodalite (Sdl) and nepheline (Nph) in nepheline syenite, SYU (PPL). (f) Amphibolitization of clinopyroxene in syenite, SYU (PPL) (g) Garnet (Grt) in syenite with inclusions of alkali feldspar, SYU (PPL) (h) Garnet-amphibole cluster with alkali feldspar and nepheline in pegmatitic nepheline syenite, SYU (PPL).

Figure 5 – (a) Total Alkali-Silica (TAS) classification diagram of the volcanic units (OVU and CVU) of the TAC. Alkaline-subalkaline division from Irvine & Baragar (1971). (b) R1-R2 classification diagram (from De La Roche et al., 1980) of the intrusive SYU unit of the TAC. (c) A/NK vs A/CNK diagram (after Shand, 1947) based on the molecular proportions of Al (A), Na (N), K (K) and Ca (C), showing that the rocks of the TAC can largely be classified as metaluminous. (d) Modified Alkali-Lime Index (MALI, after Frost and Frost, 2008) plotted as a function of SiO₂ content for the TAC rocks that are generally alkalic in composition. Comparative data for Eocene magmatic rocks from the Talysh mountains, Azerbaijan (Vincent et al., 2005 – pink diamonds) and Pliocene-Quaternary volcanic rocks from central and northern Armenia (Neill et al., 2013, 2015 – orange field).

Figure 6 – Harker diagrams of the TAC samples for selected major (a-c) and trace (d-f) elements. The limestone assimilation trend in (c) was calculated after Costa et al. (2013) using limestone composition WGZ-3 from Zhang et al. (2017). All symbols as in Fig. 5.

Figure 7 – (a) Chondrite-normalised REE diagram highlighting more pronounced LREE fractionation and negative Eu anomalies within the SYU relative to the volcanic units of

the TAC. Normalisation values from Boynton et al. (1984). (b-d) Mantle-normalised trace element diagrams of rocks from the TAC; (b) – OVU, (c) – SYU, (d) – CVU. Normalisation values after McDonough & Sun, 1995. Comparative data from Neill et al. (2015) for Pliocene-Quaternary volcanic rocks from central and northern Armenia.

Figure 8 – ⁴⁰Ar/³⁹Ar age spectrum plot for the amphibole separate from syenite sample 6-8-12.

Figure 9 – Sr-Nd isotope diagram of the TAC data (red squares) in comparison to other Eocene-Quaternary igneous rocks in the Lesser Caucasus and adjacent regions. The mantle array is from Lebedev et al. (2007) after DePaolo & Wasserburg (1979). Data sources: (I) – Moritz et al. (2017); (II) – Aydınçakçır & Şen, (2013); (III) – Kazemi et al. (2018); (IV) – Mazhari et al. (2009); (V) – Connor et al. (2011); (VI) – Kheirkhah et al. (2009); (VII) – Neill et al. (2013; 2015) . Literature data were recalculated using the ⁸⁷Rb decay constant of 1.3972 x 10⁻¹¹ a⁻¹.

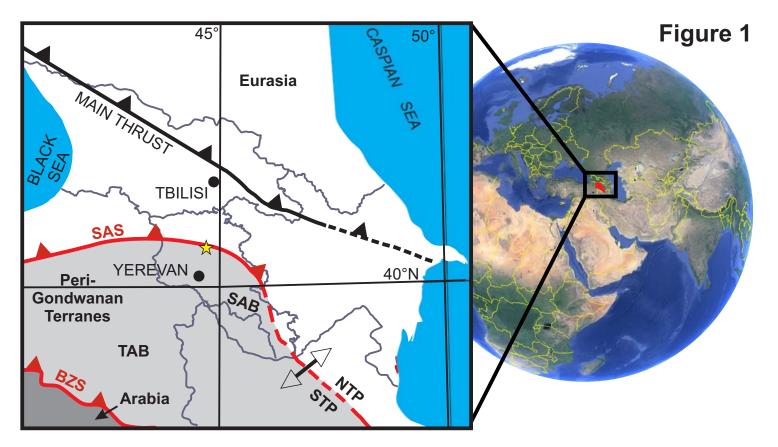
Figure 10 – TAC samples plotted in various diagrams to evaluate effects of distinct magmatic processes. (a) Rb/Nd vs. Rb diagram (after Schiano et al. 2010) where horizontal trends reflect fractional crystallization and positive correlations can be caused by mixing or batch partial melting. (b) Rb vs. Rb/V diagram (after Schiano et al. 2010) where curved trends, as observed for the TAC rocks, reflect fractional crystallisation or mixing. (c) Rb/Ba vs. Ba diagram exhibiting a smooth trend indicative of feldspar fractionation. (d) Initial ⁸⁷Sr/⁸⁶Sr isotope ratios of the TAC samples plotted against SiO₂ content. All samples except one plot in a very narrow range of (⁸⁷Sr/⁸⁶Sr)_i ratios and there is no clear trend with increasing SiO₂ content. The sample with the elevated (⁸⁷Sr/⁸⁶Sr)_i

ratio (2-7-09) has a high Rb/Sr ratio of \sim 8 and might be affected by post-magmatic Rb and/or Sr mobilization and a larger uncertainty in recalculation.

Figure 11 – Major and trace element indicators for fractionation and source enrichment processes. (a) CaO/Al₂O₃ vs FeO_t/MgO diagram showing fractionation trends for plagioclase, olivine and amphibole(am)/clinopyroxene(cpx) after Moritz et al. (2016). (b) Eu/Eu* vs SiO₂ diagram depicting negative Eu anomalies in SYU and CVU samples, indicating plagioclase fractionation. (c) Dy/Yb vs SiO₂ diagram with fractionation trends for garnet and amphibole after Davidson et al. (2007). (d) La/Sm vs Sm/Yb diagram with approximate mineral stability thresholds of in mantle melt residues after Mamani et al. (2010). Note the distinct signatures for the two volcanic units of TAC. (e) Ba vs Nb/Y diagram displaying trends for fluid enrichment due to slab dehydration and mantlederived melt enrichments after Kepehinskas et al. (1997). Slab fluid enrichment is prominent in the OVU rocks. (f) Th/Yb vs Ba/La diagram with trends for enrichment from subducted slab sediments and slab fluids from Woodhead et al. (2001). Elevated Ba/La ratios in OVU rocks suggest source enrichment via slab fluids.

Figure 12 – Trace element ratio diagrams of TAC rocks. (a) Ba/Nb vs La/Nb. Alkali feldspar fractionation trend highlighted as a result of Ba depletion. Field boundaries after Jahn et al. (1999). (b) $(Ta/La)_N$ vs $(Hf/Sm)_N$. Influence of subduction metasomatism is suggested by strongly decreasing $(Ta/La)_N$ ratios. Field boundaries after La Flèche et al. (1998). Comparative data for Eocene magmatic rocks from the Talysh mountains, Azerbaijan (Vincent et al., 2005 – pink diamonds) and Pliocene-Quaternary volcanic rocks from central and northern Armenia (Neill et al., 2013, 2015 – orange field).

Figure 13 – Pseudoleucite from the OVU of the Tezhsar Complex. (a) Scanned thin section image of a single pseudoleucite crystal. (b,c) Back-scattered electron images of (b) the boundary between matrix and pseudoleucite and (c) the interior of the pseudoleucite. Note the presence of cancrinite (Ccn) and analcime (Anl), other mineral abbreviations as in Figure 4.



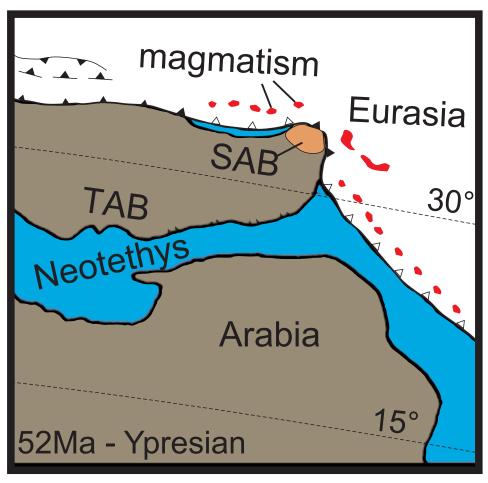


Figure 2

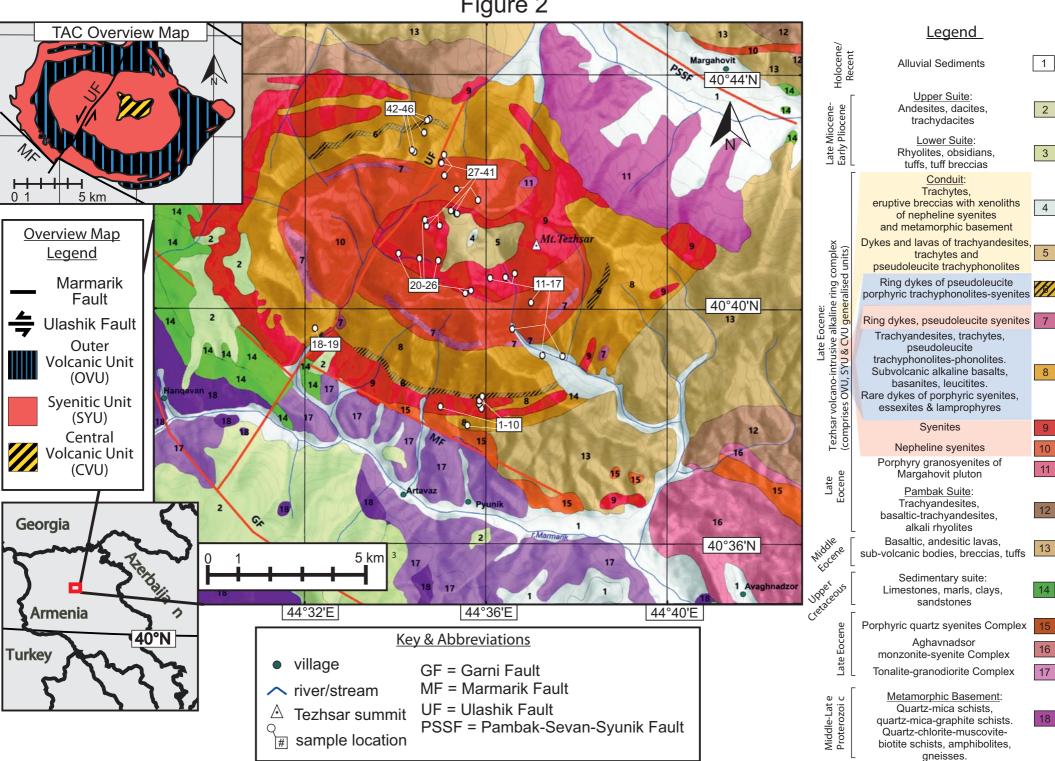


Figure 3

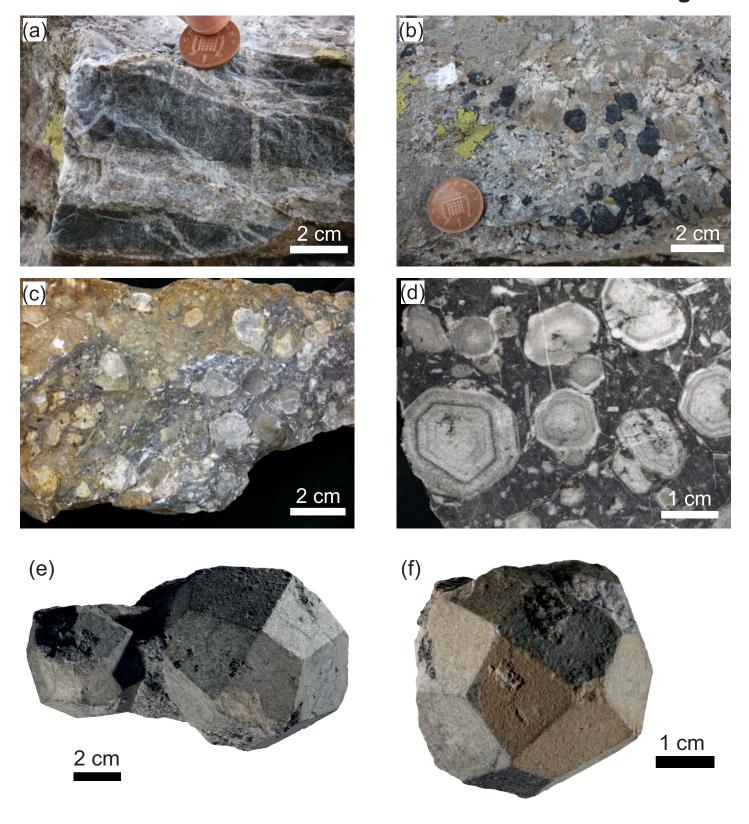
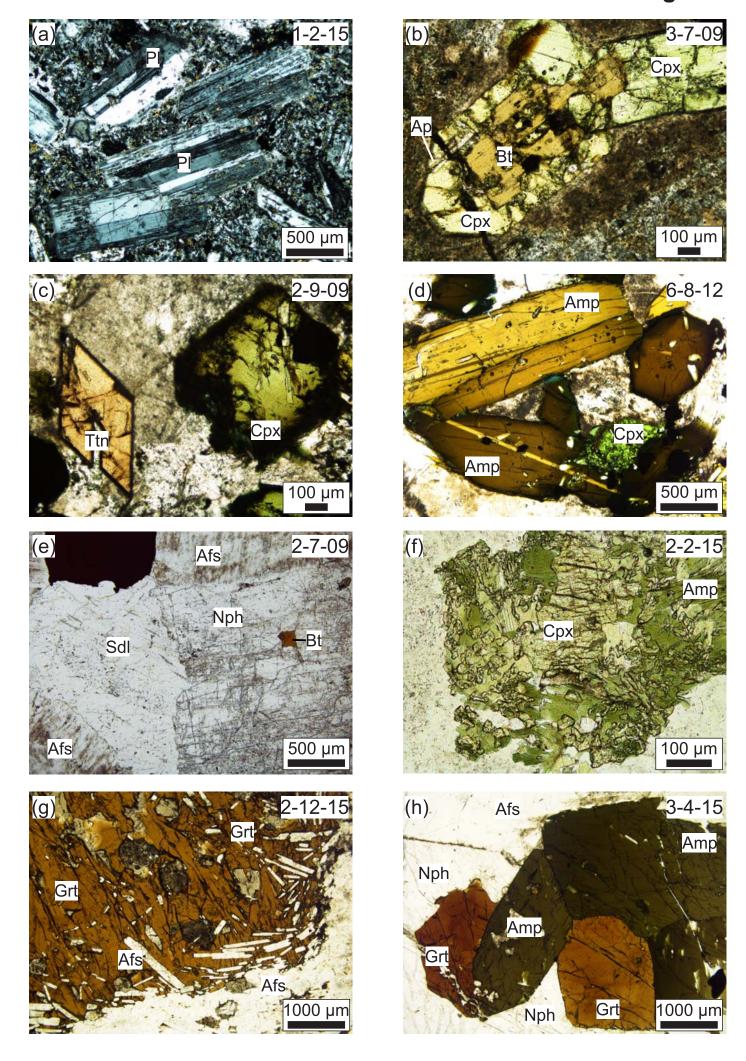
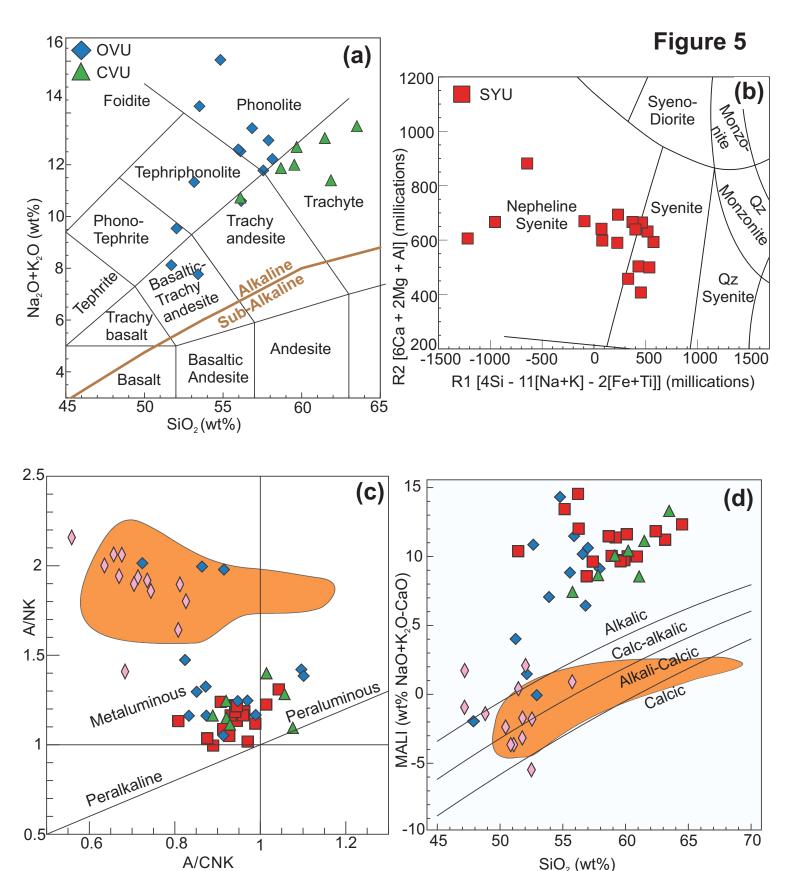


Figure 4





SiO₂ (wt%)

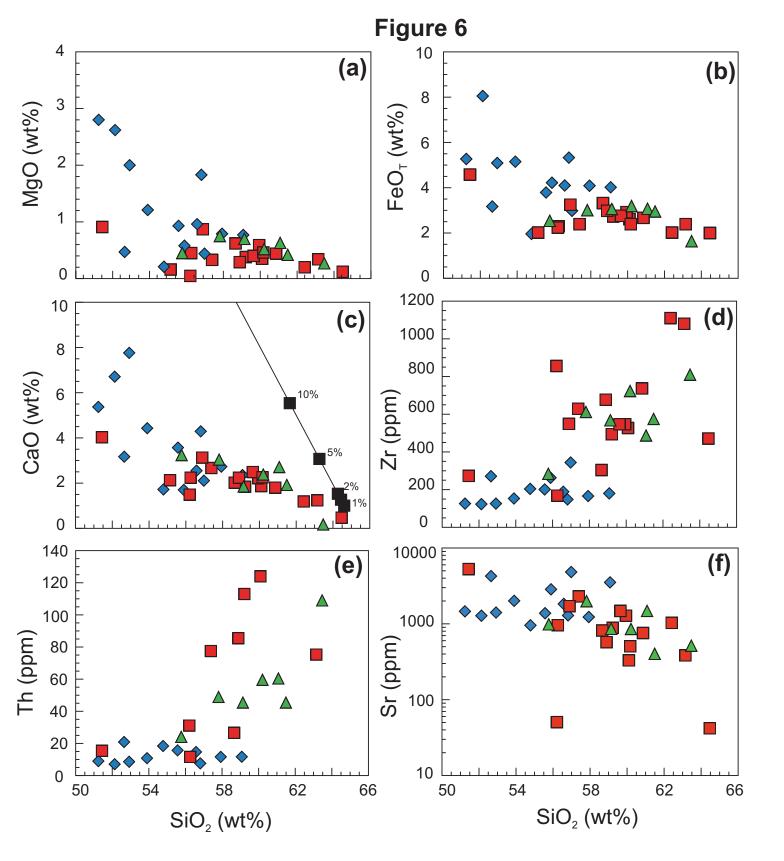
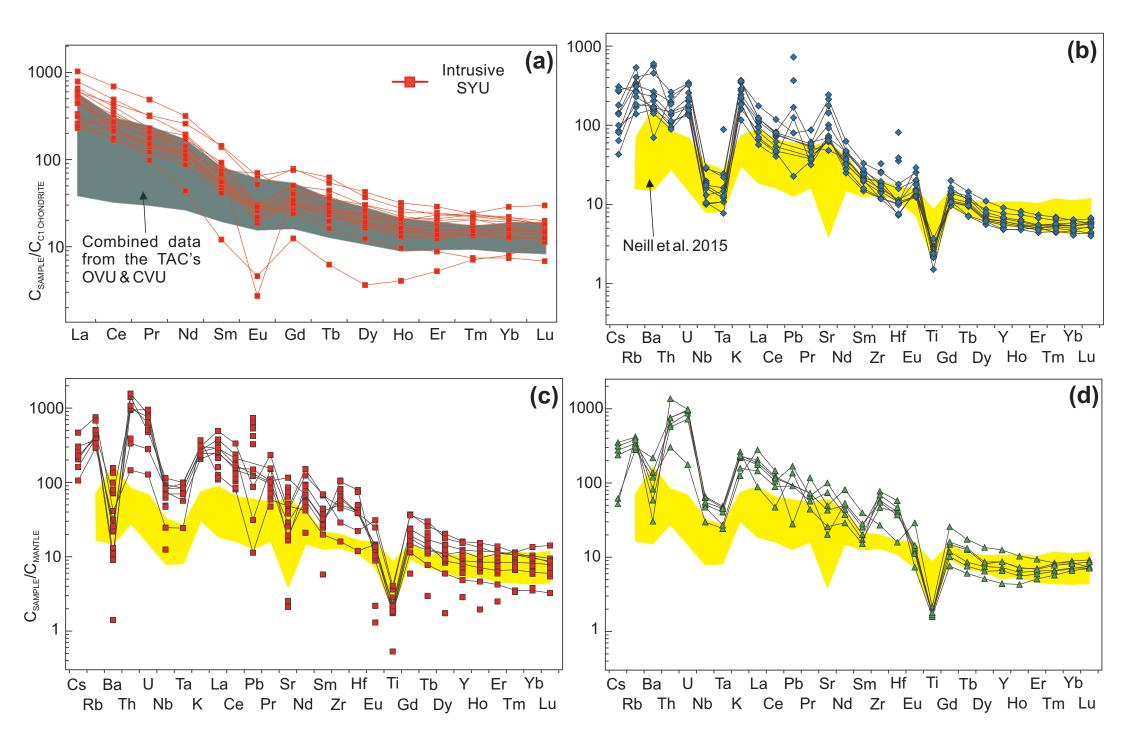


Figure 7



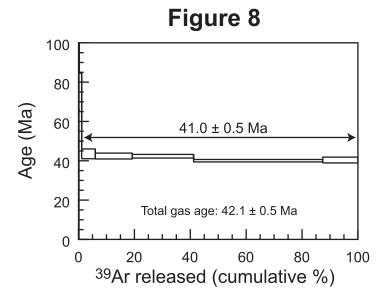


Figure 9

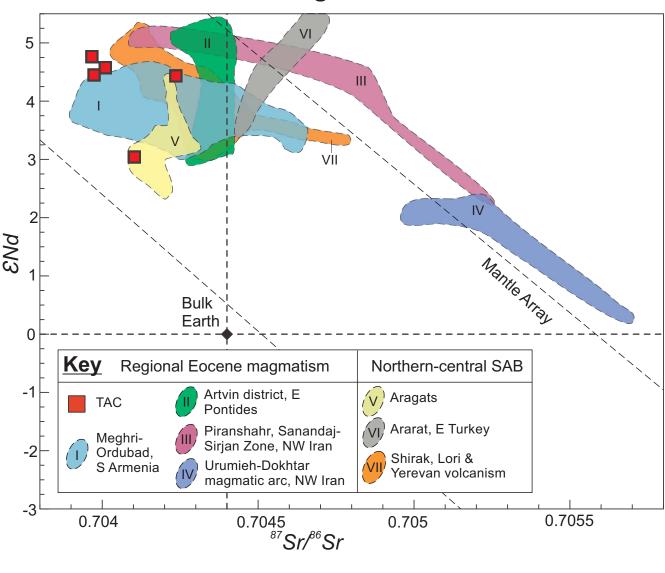
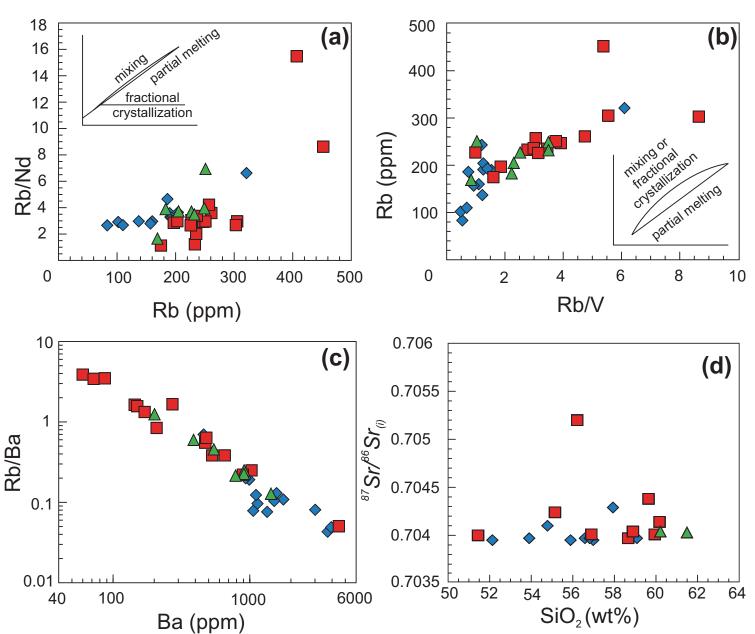


Figure 10



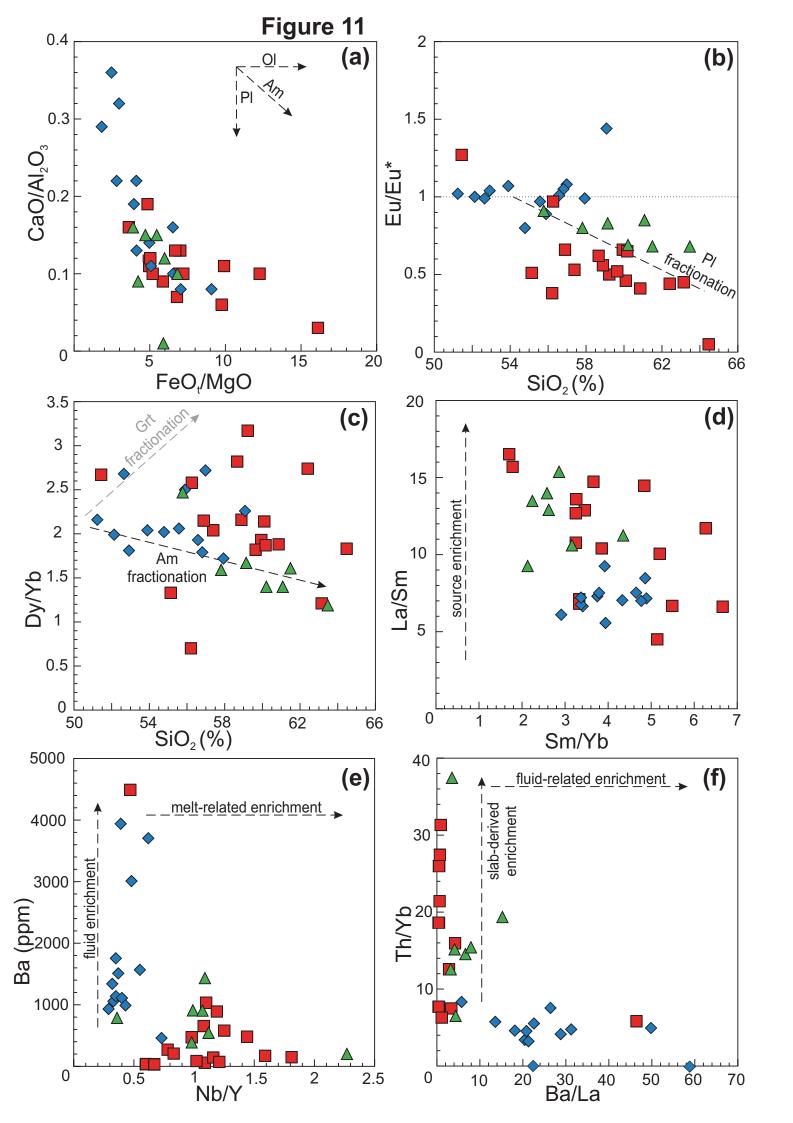
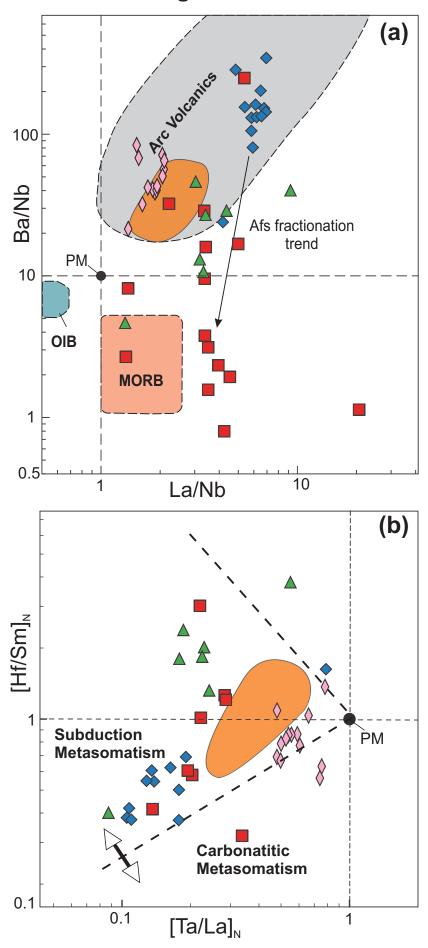
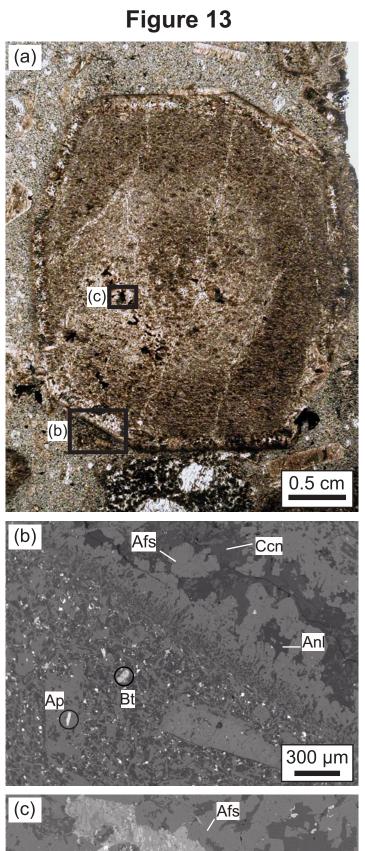


Figure 12





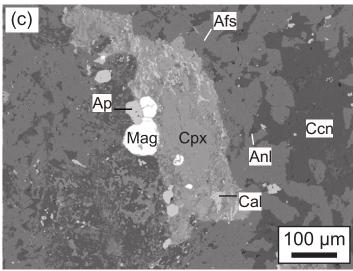


Table 1: Whole-rock ge	eochemical dat	a for sampl	es from the T	ezhsar Alka	line Comple	2X																			
Lithological Unit	OVU	OVU	OVU	OVU	OVU	OVU	OVU	OVU	OVU	OVU	OVU	OVU	OVU	SYU	SYU	SYU	SYU	SYU	SYU	SYU	SYU	SYU	SYU	SYU	SYU
Sample #	1-2-15	3-1-15	3-2-15	3-3-15	6-1-12	6-2-12	10-43-08	10-44-08	2-8-09	2-11-09	2-12-09	2-13-09	3-2-09	1-4-15	2-1-15	2-8b-15	2-11-15	3-5-15	6-3-12	6-4-12	6-5-12	6-8-12	6-9-12	6-10-12	10-45-08
Field Reference Analysed at	27 BV	42 BV	43 BV	44 BV	18 p	19 P	RHL	9 RHL	2 RHL	5 RHL	6 RHL	7 RHL	11 RHL	29 BV	30 BV	37 BV	40 BV	46 BV	20 P	21 P	22 P	24 P	25 P	26 P	10 RHL
Easting	44.58448	44.57898	44.57822	44.57707	44.53700	44.53700	44.58325	44.58325		44.59755	44.59755		44.62817	44.58339	44.58454	44.58279	44.59697	44.57256	44.56770	44.56770	44.57538	44.59230	44.57735	44.58232	44.58325
Northing	40.71049	40.72075	40.72034	40.71685	40.66112		40.63900	40.63900		40.64063	40.64063		40.65332	40.70836	40.70469	40.68375	40.69765	40.71180	40.68247	40.68247		40.67115	40.69195	40.68042	40.63900
					10.00112	10.00112	10.00700	10.00700	10.05505	10.01000	10.01005	10.01100	10.00002			10.000.0			10.00217	10.00217	10.00107	10.07110	10.07170		
SiO ₂	52.92	52.65	51.24	55.57	55.89	56.98	53.90	56.58	56.81	54.78	57.93	59.08	52.13	59.22	60.10	63.15	57.39	56.26	60.86	62.41	59.94	56.89	64.47	59.64	58.66
TiO ₂	0.60	0.45	0.59	0.46	0.54	0.50	0.57	0.43	0.54	0.30	0.43	0.48	0.74	0.73	0.46	0.36	0.38	0.35	0.37	0.40		0.40	0.36	0.57	0.82
Al_2O_3	21.48	20.29	18.24	19.09	20.12	20.66	19.95	20.26	19.91	21.73	19.72	20.61	20.75	19.03	19.35	17.98	19.90	20.72	19.26	18.42	18.56	19.29	17.81	19.24	20.07
Fe ₂ O ₃ *	1.64	1.02	1.70	1.22	1.36	0.96	1.66	1.32	1.72	0.63	1.32	1.30	2.60	0.88	0.84	0.77	0.77	0.74	0.86	0.65	0.94	1.05	0.65	0.89	1.07
FeO*	3.45	2.15	3.57	2.57	2.86	2.02	3.49	2.78	3.61	1.33	2.77	2.72	5.45	1.85	1.77	1.62	1.62	1.56	1.81	1.37	1.98	2.20	1.35	1.87	2.25
Mn0	0.17	0.22	0.17	0.22			0.19	0.20		0.13	0.11		0.23	0.22	0.16		0.15		0.16	0.12			0.07	0.14	0.19
MgO	2.00	0.47	2.80	0.93		0.44	1.21	0.96		0.21	0.79		2.62	0.38	0.35		0.33		0.44	0.20		0.87	0.12	0.40	0.62
Na ₂ O	7.76 4.32	3.17 4.03	5.37 3.93	3.57 5.34		2.11 2.01	4.43 5.26	2.55 4.41		1.72 5.88	2.73 5.36	2.36 1.90	6.71 2.90	1.84 4.42	1.87 5.66		2.67 5.28	2.24 3.59	1.80 5.25	1.19 4.21		3.13 5.15	0.47 6.47	2.49 4.76	2.02 4.68
K ₂ O	3.38	10.00	5.46	7.06	7.88		6.23	8.32		10.16	6.49		5.28	8.80	7.83		7.02		6.55	8.82		6.55	6.33	7.39	8.82
-	0.45	0.09	0.41	0.31	0.21	0.17	0.30	0.17		0.05	0.21	0.14	0.54	0.05	0.05		0.04	0.07	0.09	0.04		0.20	0.03	0.06	0.06
P ₂ O ₅ LOI	0.45	3.88	4.98	2.70	1.73	1.73	4.19	3.11	1.22	2.98	2.78	1.56	0.54	1.63	1.21		3.23		1.50	1.28		3.08	1.12	1.59	0.06
SUM	99.09	98.42		99.04			101.39	101.09		99.91	100.65		100.80	99.05	99.65		98.78		98.95	99.10			99.25	99.04	100.25
Ba	1060	3010	1340	1760		3710	1510	991		460	1110		932	60	144		72		655	481		891	87	150	208
Co	30.1 bdl	17.1 bdl	24.9 bdl	21 bdl			9.08 2.81	6.34	9.23 2.28	3.43 1.65	6.23 2.89		18.1 26	13.1 bdl	12.7 bdl		77.2 bdl		na 21	na bdl		na 14	na 12		
Cs	2.1	3	0.9	2.9			6.47	5.63		3.78	1.34		5.88	4.8	3.4		4.7		na	na			na		
Ga	18	15.7					na	na		na	na	na	na	18.7	20.9		19.5		19	20			23		na
Hf	2.9	5.1	2.8	4.5			3.01	11		3.03	3.24		2.14	11	11.7		11		21	bdl			12		
Nb	6.8	18.6					10.5	12.3		19.2	10.5		6.95	75.1	61.7		45.9		41	59			45	56	
Rb Sc	83.2 bdl	243 bdl	102 bdl	191 bdl			157 3.45	190 2.19		321 0.297	137 2.75		186 13.1	233 bdl	235 bdl		247 bdl		251 1.7	305 1.3			303 2.9	236 2.1	
Sn	1	bdl	bdl	bdl			na	na		na	na	na	na	2	2		1	bdl	na	na			na	na	na
Sr	1410	4260	1460	1380			2010	1830		956	1230		1280	885	330		2300		757	1030	1280	1710	42	1480	817
Ta	0.4	0.9	0.4	0.6			0.445	3.28		0.815	0.448		0.286	3.7	3.1		2.6		na	na		na	na	na	
Th	8.6 2.9	20.9	9 2.7	15.7	na na	-	10.8 3.43	14.7 5.21	7.56 4.07	18.4 6.57	11.6 4.53	11.7 3.64	7.08 3.16	113 9.7	124 16.7		77.5 15.5	11.6 2.6	na na	na na			na na	na na	
V	155	202	212	152			169	124		52.6	112	138	250	84	68		63	84	67	55			35	79	
Zr	126	271	126	201	263	344	153	189	148	204	166	180	123	494	527		629	168	737	1110			471	547	304
Y	20.7	39	20.8	33			28.4	28.7	24.9	26.4	26.3	29.6	23.6	68.7	53		38		38	41			44	31	66
La	36.7	114		77.8			72.5	72.9		80.2	61.2	79.1	45.6	319 558	244		161		140	81			204	75 174	186 354
Dr.	67.8	199 22.1	76.1 9.1	138 15.6	127 14		125 14	124 13.6		130 13.4	104 11.7	135 14.8	79.1 9.58	558	396 38.5		264 26.7		204	224 26			328 28	174 18	
Nd	31.3	78.3	35	58.1	56		56.2	53.1		48.5	46.1	59.5	40	190	117		84.4		85	103			113	70	155
Sm	6	13.5	6.5				10.3	9.7		8.67	8.49		8.19	27.2	16.6	10.7	12.5		13	18	10	10	15		
Eu	1.9	3.9	2.1	3.1			3.1	2.74		1.97	2.32		2.26	3.8	2.2		1.9		1.5	2.2			0.2		
Gd Tb	5.2 0.7	10.9 1.4	5.9 0.7	8.8 1.1			7.58 1.17	7.16 1.19		6.48 1.05	6.05 0.988	7.94 1.26	5.82 0.959	20.2	12.8 1.7		9.7 1.3		9.5 1.3	13 2.1			11 1.8	8 1.1	
Dv	3.8	7.4				4.9	4.86	4.93		4.46	4.34	5.35	4.14	13.7	9.7		7.4		7.5	9.6			8.4	6	
Но	0.7	1.2	0.7		0.78	0.77	0.895	0.949	0.795	0.821	0.829	0.932	0.773	2.3	1.8	1.4	1.3	0.7	1.5	1.6	1.1	0.94	1.5	1.2	1.98
Er	2.2	3.2		3			2.37	2.43		2.21	2.35		2.08	6.1	5.1		3.7		3.9	4.2		2.6	4.4		
Tm Yb	0.3 2.1	0.5 2.8		0.4 2.8			0.373 2.38	0.457 2.56		0.355 2.21	0.389 2.52		0.338 2.08	0.8 4.3	0.7 4.5		0.6 3.6		0.66	0.68		0.45 2.6	0.75 4.6	0.53 3.3	0.72 4.22
Lu	0.3	0.4					0.38			0.344	0.409		0.358	0.6	0.6		0.5		0.59	0.45			0.57	0.48	0.59
Мо	0.8	0.2	0.5				1.11		1.69	1.08	1.1		2.21	0.3	1.2		1.9		na	na			na		1.01
Cu	165	2.3	117	6.1	na	na	23.8	11.9	24.3	6.77	91.1	29.6	100	7.4	41.3	8.3	45.8	15	na	na	na	na	na	na	7.98
Li	na	na		na			18.4	19.3		24.9	12.8		10.8	na	na		na		na	na			na		
Pb Zn	18.7	12 86		25.3 97			na 89.2	na 96.2		72.2	na 67.9	na 95.7	na 120	13.1	4.7		22.4 34		97 na	63 na			49 na	80 na	
Ni	1.5	0.2		0.2			1.4	2.35		1.52	1.27	3.48	13.6	0.1	0.4		0.5		na	na na		na	na	na	4.56
87Sr/86Sr (measured)	1.0	5.2	23.0		0.704071		0.704096	0.704139		0.704660	0.704470		0.704195	1			2.0	5.2			0.704341			0.704647	
87Sr/86Sr (initial)					0.70395		0.70397	0.70397		0.70410	0.70429		0.70395								0.70401	0.70401		0.70438	0.70401
143Nd/144Nd (measured))						0.512859			0.512770		0.512851													0.512849
143Nd/144Nd (initial)							0.512829			0.512741		0.512820													0.512820
ENd							4.8			3.0		4.6													4.6
Alkalinity Index (AI)	0.105	0.028	0.058	0.026	0.028	0.056	0.044	0.037	0.046	0.007	0.037	0.056	0.099	0.022	0.015		0.028	0.032	0.035	0.019		0.037	0.003	0.033	0.025
FSSI	-0.05	-0.38	-0.12	-0.23	-0.22	-0.02	-0.24	-0.17	-0.15	-0.51	-0.11	0.37	-0.03	-0.1	-0.16	-0.03	-0.38	-0.25	0	0.79	-0.02	-0.12	1.12	-0.05	-0.17
			- 3±																						
* Fe ₂ O ₃ and FeO conten									ļl												1				
Abbreviations: BV = Bu	ureau Veritas, F	' = Potsdan	ı, KHL = Koya	ii Holloway	London, K =	Kiel; na = no	t analysed; l	oai = below	aetection lim	IT										l					

SYU	SYU	SYU	SYU	SYU	CVU	CVU	CVU	CVU	CVU	CVU	CVU
2-7-09	2-9-09	3-3-09	3-5-09	3-6-09	2-3-15	2-4-15	2-6-15	2-7-15	2-8a-15	3-7-09	3-8-09
1	3	12	14	15	32	33	35	36	37	16	17
RHL	RHL	K	RHL	K	BV	BV	BV	BV	BV	RHL	RHL
44.59320 40.63365	44.59835 40.63848	44.62075 40.65353	44.61660 40.66855	44.61053 40.67683	44.58697 40.69464	44.58697 40.69464	44.58893	44.58929 40.69375	44.58279	44.60143	44.6070
40.63365	40.63848	40.65353	40.66855	40.67683	40.69464	40.69464	40.69384	40.69375	40.69033	40.67552	40.675
84.04	=		#0.00	60.45	F0.44	55.04		60.45	64.00	60.04	
56.21	51.44	55.14	58.89	60.17	59.14	57.81	55.77	63.47	61.08	60.21	61.
0.11	0.47	0.36	0.44	0.37	0.37	0.36	0.43	0.32	0.34	0.34	0.
22.61	20.65	21.91	20.96	19.05	20.13	19.49	21.71	19.61	18.56	20.00	19.
0.72	1.48	0.65	0.96	0.77	0.99	0.97	0.82	0.53	0.99	1.03	0.
1.52	3.10	1.37	2.02	1.62	2.07	2.04	1.72	1.11	2.08	2.16	2.
0.28	0.18	0.22	0.12	0.17	0.18	0.17	0.20	0.05	0.18	0.17	0.
0.04	0.91	0.16	0.29	0.46	0.70	0.75	0.45	0.27	0.63	0.52	0.
1.49 9.51	4.03	2.13	2.24	2.26	1.85	3.04 5.34	3.24 7.04	0.17	2.71	2.38	1.
	4.68	7.63	4.83	5.48				5.91	6.70	6.27	6.3
6.52	9.73	7.94	7.45	6.79	6.97	6.35	3.63	7.57	4.56	6.53	6.1
0.01	0.20	0.03	0.06	0.08	0.15	0.13	0.08	0.04	0.12	0.08	0.
1.77	2.44	1.15	2.66	1.44	1.84	2.06	4.32	0.88	0.79	1.19	100
100.79	99.31	98.69	100.92	98.66	99.34	98.51	99.41	99.93	98.74	100.88	100.
9.31	4490	na	170	530	909	906	787	201	1430	546	3
1.55	7.96	na	5.61	na	20.9	43.1	11.3	12.8	4.6	4.81	3
2.98	2.55	na	4.24	na	bdl	bdl	bdl	bdl	bdl	8.56	13
5.56	2.92	na	6.38	na	5	5.8	1.3	1.1	5	7.36	6.
na	na	na	na	na	20.2	20.5	19	22.6	19	na	
10.9	4.2	na	11.1	na	11.2	11.5	4.5	16.4	11.6	13.2	10
8.2	18	na	54.4	na	31.5	33.6	19.6	43.2	31	41.8	36
407	227	na	226	203	227	205	169	251	183	249	2
0.759	1.04	na	1.04	na	na 1	na 1	na bdl	na 2	na 1	na bdl	ŀ
50.6	na 5270	na na	na 573	na 506	861	1990	984	515	1480	855	4
0.00647	0.752	na	3.12	na	1.4	1.5	0.9	1.8	1 1 1	1.71	1.
31.1	15.4	na	85.5	na	45.4	48.9	24.1	109	60.4	59.6	45
10.4	5.73	na	12.7	na	12.1	17.6	3.6	19.9	19.1	19.7	14
14.1	230	na	71.9	na	90	89	201	242	82	71.3	66
856	273	na	677	na	568	612	284	810	487	723	5
12.3	38.1	na	34.3	na	31.9	31.3	53.9	19	28.4	37.4	37
169	96.7	na	191	138	137	114	181	57.3 79	94.1	133	1
187 12	171 18.6	na na	285 25.5	220 21.4	184 19.1	177 17	246 29.8	11.3	150 14.5	204 18.4	18
26.3	74.5	na	85	68.1	61.3	55.1	102	36.2	47	62.9	66
2.35	14.5	na	13.2	10.8	8.9	8.2	16.1	6.2	7	10.3	11
0.338	5.16	na	2.14	2.09	2.1	1.9	4.5	1.1	1.7	2.05	2.
3.22	10.6	an	10.4	8.85	7	6.5	14	4.1	5.5	8.09	8.
0.295	1.7	na	1.45	1.18	0.9	0.9	1.7	0.6	0.8	1.25	1.
1.17	7.04	na	5.91	6.23	5.2	5	9.1	3.5	4.4	5.49	5.
0.291	1.19	na	1.06	1.17	1	0.9	1.5	0.6	0.8	1.06	1.
1.1 0.229	2.94 0.434	na na	2.86 0.454	3.25 0.488	2.8	2.9 0.5	4.1 0.6	2.2 0.4	2.6 0.4	3.08 0.561	3. 0.5
1.67	2.64	na na	2.73	3.33	3.1	3.2	3.7	2.9	3.1	3.93	3.
0.365	0.398	na	0.403	0.479	0.5	0.5	0.5	0.5	0.5	0.621	0.5
1.11	1.6	na	1.72	na	0.6	3.7	1.2	2.2	0.3	1.44	2.
2.33	20.7	na	70.4	na	32	21.1	19	3.2	17.9	29.5	
66.3	17.7	na	20.9	na	na	na	na	na	na	25	3
na	na	na	na	na	20.5	25.2	4.2	20.4	13.7	na	
181	104	na	77.3	85	60	51	61	55	37	100	8
0.796	2.87	na	3.33	16	2.3	0.6	0.5	0.6	0.6	0.881	0.7040
0.718557	0.704075	0.706855	0.704695	0.704873						0.704519	0.7049
0.70520	0.70400	0.70424	0.70404	0.70414						0.70404	0.704
		0.512837			1						
		0.512813									
0.06-	0.00	4.4								0.0	
-0.003	0.021	0.005	0.047	0.025	0.044	0.038	0.061	0.017	0.026	0.023	0.0
	-0.55	-0.55	-0.07	-0.07	-0.02	-0.11	-0.16	-0.01	-0.03	-0.16	-0.
-0.63	0.55										

Table 2: Argon isotopic data for an amphibole separate of syenite sample 6-8-12.

TZ-6-8-12	Laborator	y ID: C	15038	Irradiati	on ID:	PO-2												
	⁴⁰ Ar/		³⁷ Ar/ ³⁹ Ar			³⁶ Ar/ ³⁹ Ar			K/Ca	⁴⁰ Ar*	$^{39}Ar_{K}$	⁴⁰ Ar	·*/ ³⁹ A	.r _K	Age	±	1 s	
J=0.0009720						(×10 ⁻³)				(%)	fraction (%)				(Ma)			
Laser output																		
1.8%	1397	±	251	7.4	±	137	3703	±	710.85	0.08	21.73	0.13	305	±	98	469	±	132
2.0%	1385	±	938	141	±	566	4558	±	3110	0.00	3.54	0.03	54	±	147	93	±	245
2.4%	352	±	60	6	±	149	919	±	203	0.10	22.92	0.14	81	±	44	137	±	71
2.8%	77	±	3	39	±	15	141	±	33	0.01	49.59	0.82	39	±	10	67	±	17
3.1%	37.0	±	0.4	7	±	4	42	±	5	0.09	67.55	4.85	25.1	±	1.4	44	±	2
3.3%	29.79	±	0.16	6	±	2	20	±	3	0.10	81.79	13.29	24.5	±	0.9	42.4	±	1.5
3.5%	29.80	±	0.14	1.6	±	1.7	18.7	±	1.7	0.37	81.82	21.93	24.4	±	0.5	42.3	±	0.9
3.7%	30.8	±	0.2	2.4	±	0.6	26.9	±	1.0	0.25	74.79	46.26	23.1	±	0.4	40.0	±	0.6
3.9%	31.1	±	0.4	2.4	±	1.9	27	±	3	0.25	74.67	12.56	23.3	±	0.9	40.4	±	1.5
												Total g		42.05	±	0.52		
									Plateau age (step 5-9: 98.8% of total ³⁹ A					40.96	±	0.46		
											No	ormal isochron	age (ste	p 5 to	9)	41.32	±	2.51
											In	verse isochron	age (ste	o 5 to	9)	41.25	±	2.11

Supplementary material

Whole rock major and trace element analyses

1. Royal Holloway University, London, UK

Fourteen samples were analysed by inductively coupled plasma atomic emission spectrometry (ICP-AES) for major elements and some high abundance trace elements (Sr, Zr, Cr, Sc, Zn, Co, Li, V, Be and Ni) and by inductively coupled plasma mass spectrometry (ICP-MS) for low abundance trace elements (Rb, Nb, Y, Mo, Cs, Ba, Hf, Ta, Tl, Pb, Th, U, and all REE) using a Perkin Elmer instrument. The analytical work followed the methodology described by Walsh et al. (1981) and Garbe-Schönberg (1993), respectively. The relative standard deviation (RSD) typically was $\leq 2\%$ for major elements and $\leq 5\%$ for minor and trace elements.

2. Institute of Geosciences, Kiel University, Germany

Three samples were analysed by X-ray fluorescence (XRF) on fused glass discs using a Philips PW1480 XRF spectrometer for major elements and by ICP-MS using an Agilent 7500c instrument for trace elements. For major element oxides, the RSD is ≤ 1.3 % based on multiple analyses of reference material BHVO-1. The RSD for trace elements is generally ≤ 2 % based on multiple analyses of one sample solution. Details about sample preparation and instrument calibration are given in Garbe-Schönberg (1993) and John et al. (2008), and representative data for precision and accuracy during the course of this study are provided by Laeger et al. (2013).

3. GeoForschungsZentrum (GFZ) Potsdam and Potsdam University, Germany

Nine samples were analysed for major and some trace elements (Ba, Cr, Ga, Nb, Ni, Rb, Sr, V, Y, Zn and Zr) by XRF using a Siemens SRS303-AS XRF spectrometer at the GFZ and for REE by ICP-AES using a Varian Vista MPX instrument following the methods described by Zuleger and Erzinger (1998). RSD values are in the range of 1-3% for major oxides and \leq 5% for trace elements and REE (Moazzen and Oberhänsli, 2008; Hadj Zobir et al., 2014).

4. AcmeLabs, Bureau Veritas Minerals, Vancouver, Canada

Fifteen samples were analysed for major elements by XRF using a Panalytical Axios Max instrument and by ELAN 9000 ICP-MS for trace elements and REE. The RSD is <1.2% for major oxides based on the analyses of SY-4(D) diorite gneiss and OREAS72B VMS ore standards, while for trace elements and REEs the RSD was <3.8%.

Strontium (Sr) and neodymium (Nd) isotope analyses

Strontium (Sr) and neodymium (Nd) isotope analyses were performed on a Thermo Finnigan Triton multicollector mass spectrometer at the School of Earth and Environment, University of Leeds. About 30 to 60 mg of powdered whole-rock material (same was used for the major and trace element work) was dissolved in concentrated ultra-clean HF-HNO3-HCl acids and Sr and Nd were extracted from the unspiked solutions by conventional ion-exchange chromatographic techniques (see Halama et al. 2013 for details of the analytical protocol). 87 Sr/ 86 Sr and 143 Nd/ 144 Nd ratios were normalized for mass fractionation to 86 Sr/ 88 Sr = 0.1194 and 146 Nd/ 144 Nd = 0.7219. The average 87 Sr/ 86 Sr obtained from replicate measurements of NIST SRM-987 during this study was 0.710254 and all data were corrected for the offset from the generally accepted value 0.710250 (McArthur et al. 2000). Similarly, Nd isotope data were corrected for the offset from the LaJolla reference material (143 Nd/ 144 Nd = 0.511853; Weis et al. 2005). Initial 87 Sr/ 86 Sr isotope ratios were calculated using the 87 Rb decay constant 1.3972 x 10- 11 a- 1 (Villa et al. 2015). For the calculations of the ϵ Nd values, the following parameters were used: 147 Sm decay constant λ = 6.54 x 10- 12 a- 1 , present-day (143 Nd/ 144 Nd)CHUR = 0.512638, (147 Sm/ 144 Nd)CHUR = 0.1966.

⁴⁰Ar/³⁹Ar dating

About 1 mg of amphibole from syenite sample 6-8-12 was used for ⁴⁰Ar/³⁹Ar analysis by the CO₂ laser stepwise heating technique at the Institute of Earth and Environmental Science,

Universität Potsdam. For details of the analytical protocol see Wilke et al. (2010) and Halama et al. (2014). Mineral grains were obtained by crushing, sieving and selecting the size fraction between 250-500 µm mesh size for magnetic separation and finally by hand-picking under the binocular. Separated amphiboles were cleaned ultrasonically in 10% HNO₃ for 15 minutes and then washed in de-ionized water and dried. Samples, the Fish Canyon Tuff sanidine age standard, prepared by the Geological Survey of Japan (27.5 Ma: Uto et al., 1997; Ishizuka, 1998) and salts of K₂SO₄ and CaF₂ were irradiated at the Oregon State TRIGA Reactor for 4 hours under a neutron flux of 2.5x10¹³ n cm⁻² s⁻¹. Argon isotope ratios of the gas from the samples were analyzed by stepwise heating until total fusion using a New Wave Research DualWave laser ablation system comprising a 50W CO₂ continuous laser with 10.6 μm wavelength. The extracted gas is purified in the ultra-high vacuum line via SAES getter pumps and a cold trap for 10 min. The high sensitivity Micromass 5400 noble gas mass spectrometer used for Ar isotopic analysis is equipped with an electron multiplier pulse counting system for analyzing small amounts of Ar. Raw data were corrected for procedural blank contributions, mass discrimination by analysis of atmospheric Ar, interferences of Ar isotopes derived from Ca and K and decay of radiogenic ³⁷Ar and ³⁹Ar isotopes produced by irradiation. Calculation of ages and errors was performed following Uto et al. (1997) using the total ⁴⁰K decay constant of 5.543 x 10⁻¹⁰ a⁻¹ (Steiger and Jäger, 1977) as well as decay constants of 1.978 x 10⁻² d⁻¹ for ³⁷Ar and 2.58 x 10⁻³ a⁻¹ for ³⁹Ar.

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